#### **ANTIBACTERIAL AGENTS**

#### **CROSS REFERENCE**

This application claims the benefit of U.S. Serial No. 60/405,464, filed on August 23, 2002, under 35 U.S.C. § 119(e)(i).

5

15

20

25

30

#### FIELD OF THE INVENTION

The present invention relates to antibacterial agents that are useful for sterilization, sanitation, antisepsis, and disinfection.

#### BACKGROUND

The inappropriate growth of a variety of bacteria has been a problem for many years. Bacteria have caused degradation of natural product materials, infection in humans and other animals, and spoilage of foods.

Sterilization denotes the use of either physical or chemical agents to eliminate all viable bacteria from a material, while disinfection generally refers to the use of germicidal chemical agents to destroy the potential infectivity of a material.

Sanitizing refers to procedures used to simply lower the bacterial content of utensils used for food. Antisepsis refers to the topical application of chemicals to a body surface to kill or inhibit pathogenic microbes. Disinfectants are widely used for skin antisepsis in preparation for surgery.

Bacteria are the smallest organisms that contain all the machinery required for growth and self-replication. A bacterium includes a rigid cell wall surrounding the cytoplasmic membrane, which itself encloses a single naked chromosome without a nuclear membrane. The cytoplasmic membrane consists primarily of a bi-layer of lipid molecules.

The fundamental criterion of bactericidal action is loss of the ability of the organism to propagate indefinitely, when placed in a suitable environment.

Bactericidal action suggests microbe damage of various types, including the triggering of irreversible damage to the cytoplasmic cell membrane or irreversible impairment of the DNA (or viral RNA replication. Accordingly, sterilization is not identical with destruction of microbes. Additionally, it is understood that damage to nucleic acids

(DNA or RNA) is not always irreversible, as it is known that ultraviolet light-induced damage to viral nucleic acids can be repaired by enzymatic and genetic mechanisms.

#### **SUMMARY OF THE INVENTION**

The invention relates to antibacterial agents that are useful for sterilization, sanitation, antisepsis, and disinfection.

In one aspect, the invention features methods of using antibacterial agents of formula I for sterilizing, sanitizing, antisepsis, or disinfecting. The method includes applying the antibacterial agent to a location in need of sterilization, sanitation, antisepsis, and disinfection. Specifically, a method of sterilization, sanitation, antisepsis, and disinfection, includes applying antimicrobial compounds to a surface in need of sterilization, sanitation, antisepsis, and disinfection. The antimicrobial compounds are applied in a therapeutically acceptable amount, e.g., an amount sufficient to kill or hinder the growth of bacteria on the surface to be sterilized, sanitized, or disinfected.

In general, the antibacterial agents have the formula

$$\begin{array}{c|c} H & R_2 \\ H & R_1 \\ X & Y \\ R_4 & I \end{array}$$

or a pharmaceutically acceptable salt thereof,

wherein

5

10

15

25

X = NH

Y = CO, CS, -C(=N-CN) or

X and Y together form an alkene, or C<sub>3</sub>-C<sub>5</sub> cycloalkyl;

 $R_1$  is –COOH;

R<sub>2</sub> is an electron withdrawing group;

R<sub>4</sub> is an optionally substituted aryl, provided that the aryl is not simultaneously substituted with a sulfonamide and a urea or thiourea, further provided that the aryl is not solely substituted at the ortho-position relative to Y, and still further provided that the aryl is not substituted with a group selected from

$$-O_2S-W_1 \qquad N-R_{10} \qquad -SO_2\text{-NH}(C1\text{-C4 alkyl})\text{-N}(C1\text{-C4alkyl})_2$$

$$-HN \qquad \qquad -O_2S-N \qquad R_{10} \qquad -CH_2\text{-NH}(C1\text{-C4alkyl})$$

$$-CH_2\text{-NH}(C1\text{-C4alkyl})$$

$$-CH_2\text{-NH}(C1\text{-C4alkyl})$$

$$-CH_2\text{-NH}(C1\text{-C4alkyl})$$

W<sub>1</sub> is N or CH;

5

10

15

20

 $R_{10}$  is  $C_1$ - $C_4$  alkyl,  $C_1$ - $C_4$  substituted alkyl, Het, substituted Het, aryl, or substituted aryl; and

 $R_{15}$  is H,  $C_1$ - $C_4$  alkyl,  $C_1$ - $C_4$  substituted alkyl, Het, substituted Het,  $C_4$ - $C_7$  cycloalkyl.

#### DETAILED DESCRIPTION OF THE INVENTION

The term "halo" refers to a halogen atom selected from Cl, Br, I, and F.

The term "alkyl" refers to both straight- and branched-chain moieties. Unless otherwise specifically stated alkyl moieties include between 1 and 9 carbon atoms.

The term "alkenyl" refers to both straight- and branched-chain moieties containing at least one -C=C-. Unless otherwise specifically stated alkenyl moieties include between 1 and 9 carbon atoms.

The term "alkynyl" refers to both straight- and branched-chain moieties containing at least one −C≡C−. Unless otherwise specifically stated alkynyl moieties include between 1 and 9 carbon atoms. between 1 and 6 carbon atoms

The term "alkoxy" refers to -O-alkyl groups.

The term "cycloalkyl" refers to a cyclic alkyl moiety. Unless otherwise specifically stated cycloalkyl moieties will include between 3 and 9 carbon atoms.

The term "cycloalkenyl" refers to a cyclic alkenyl moiety. Unless otherwise specifically stated cycloalkenyl moieties will include between 5 and 9 carbon atoms and at least one -C=C- group within the cyclic ring.

The term "amino" refers to -NH<sub>2</sub>.

The term "sulfonamide" refers to a  $-S(O)_2$ - $N(Q_{10})_2$ 

The term "aryl" refers to phenyl and naphthyl.

The term "het" refers to mono- or bi-cyclic ring systems containing at least one heteroatom selected from O, S, and N. Each mono-cyclic ring may be aromatic, saturated, or partially unsaturated. A bi-cyclic ring system may include a mono-cyclic ring containing at least one heteroatom fused with an cycloalkyl or aryl group. A bi-cyclic ring system may also include a mono-cyclic ring containing at least one heteroatom fused with another het, mono-cyclic ring system.

Examples of "het" include, but are not limited to, pyridine, thiophene, furan, pyrazoline, pyrimidine, 2-pyridyl, 3-pyridyl, 4-pyridyl, 2-pyrimidinyl, 4-pyrimidinyl, 10 5-pyrimidinyl, 3-pyridazinyl, 4-pyridazinyl, 3-pyrazinyl, 4-oxo-2-imidazolyl, 2imidazolyl, 4-imidazolyl, 3-isoxazolyl, 4-isoxazolyl, 5-isoxazolyl, 3-pyrazolyl, 4pyrazolyl, 5-pyrazolyl, 2-oxazolyl, 4-oxo-2-oxazolyl, 5-oxazolyl, 1,2,3oxathiazole, 1,2,3-oxadiazole, 1,2,4-oxadiazole, 1,3,4-oxadiazole, 2-thiazolyl, 4-thiazolyl, 5-thiazolyl, 3-isothiazole, 4-isothiazole, 5-isothiazole, 2-15 furanyl, 3-furanyl, 2-thienyl, 3-thienyl, 2-pyrrolyl, 3-pyrrolyl, 3-isopyrrolyl, 4isopyrrolyl, 5-isopyrrolyl, 1,2,3,-oxathiazole-1-oxide, 1,2,4-oxadiazol-3-yl, 1,2,4oxadiazol-5-yl, 5-oxo-1,2,4-oxadiazol-3-yl, 1,2,4-thiadiazol-3-yl, 1,2,4-thiadiazol-5vl. 3-oxo-1,2,4-thiadiazol-5-yl, 1,3,4-thiadiazol-5-yl, 2-oxo-1,3,4-thiadiazol-5-yl, 1,2,4-triazol-3-yl, 1,2,4-triazol-5-yl, 1,2,3,4-tetrazol-5-yl, 5-oxazolyl, 3-isothiazolyl, 20 4-isothiazolyl, 5-isothiazolyl, 1,3,4,-oxadiazole, 4-oxo-2-thiazolinyl, 5-methyl-1,3,4thiadiazol-2-yl, thiazoledione, 1,2,3,4-thiatriazole, 1,2,4-dithiazolone, phthalimide, quinolinyl, morpholinyl, benzoxazoyl, diazinyl, triazinyl, quinolinyl, quinoxalinyl, naphthyridinyl, azetidinyl, pyrrolidinyl, hydantoinyl, oxathiolanyl, dioxolanyl, imidazolidinyl, and azabicyclo[2.2.1]heptyl. 25

The term "heteroaryl" refers to a mono- or bicylic het in which at least one cyclic ring is aromatic.

The term "substituted alkyl" refers to an alkyl moiety including 1-4 substituents selected from halo, het, cycloalkyl, cycloalkenyl, aryl,  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-SQ_{10}$ ,

$$\begin{split} -S(O)Q_{10}, -OS(O)_2Q_{10}, -C(=NQ_{10})Q_{10}, -C(=N-O-Q_{10})Q_{10}, -S(O)_2-N=S(O)(Q_{10})_2, \\ -S(O)_2-N=S(Q_{10})_2, -NQ_{10}Q_{10}, -C(O)Q_{10}, -C(S)Q_{10}, -C(O)OQ_{10}, -OC(O)Q_{10}, \\ -C(S)NQ_{10}Q_{10}, -N(Q_{10})C(S)NQ_{10}Q_{10}, -C(O)NQ_{10}Q_{10}, -C(O)C(Q_{16})_2OC(O)Q_{10}, -CN, \\ -C(S)NQ_{10}Q_{10}, -C(O)Q_{10}, -$$

10

15

20

=O, =S, -NQ<sub>10</sub>C(O)Q<sub>10</sub>, -NQ<sub>10</sub>C(O)NQ<sub>10</sub>Q<sub>10</sub>, -S(O)<sub>2</sub>NQ<sub>10</sub>Q<sub>10</sub>, -NQ<sub>10</sub>S(O)<sub>2</sub>Q<sub>10</sub>, -NQ<sub>10</sub>S(O)Q<sub>10</sub>, -NQ<sub>10</sub>SQ<sub>10</sub>, -NO<sub>2</sub>, and -SNQ<sub>10</sub>Q<sub>10</sub>. Each of the het, cycloalkyl, cycloalkenyl, and aryl being optionally substituted with 1-4 substituents independently selected from halo and Q<sub>15</sub>.

The term "substituted aryl" refers to an aryl moiety having 1-3 substituents selected from  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-S(O)_2Q_{10}$ ,  $-S(O)Q_{10}$ ,  $-OS(O)_2Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-S(O)_2-N=S(O)(Q_{10})_2$ ,  $-S(O)_2-N=S(Q_{10})_2$ ,  $-NQ_{10}Q_{10}$ ,  $-C(O)Q_{10}$ , -C

The term "substituted het" refers to a het moiety including 1-4 substituents selected from  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-S(O)_2Q_{10}$ ,  $-S(O)Q_{10}$ ,  $-OS(O)_2Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-S(O)_2-N=S(O)(Q_{10})_2$ ,  $-S(O)_2-N=S(Q_{10})_2$ ,  $-NQ_{10}Q_{10}$ ,  $-C(O)Q_{10}$ ,  $-C(O)Q_{1$ 

The term "substituted alkenyl" refers to a alkenyl moiety including 1-3 substituents  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-S(O)_2Q_{10}$ ,  $-S(O)Q_{10}$ ,  $-OS(O)_2Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-S(O)_2-N=S(O)(Q_{10})_2$ ,  $-S(O)_2-N=S(Q_{10})_2$ ,  $-NQ_{10}Q_{10}$ ,  $-C(O)Q_{10}$ ,  $-C(O)Q_{$ 

5

10

15

30

The term "substituted alkoxy" refers to an alkoxy moiety including 1-3 substituents  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-S(O)_2Q_{10}$ ,  $-S(O)Q_{10}$ ,  $-OS(O)_2Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-S(O)_2-N=S(O)(Q_{10})_2$ ,  $-S(O)_2-N=S(Q_{10})_2$ ,  $-NQ_{10}Q_{10}$ ,  $-C(O)Q_{10}$ , -

The term "substituted cycloalkenyl" refers to a cycloalkenyl moiety including 1-3 substituents  $-OQ_{10}$ ,  $-SQ_{10}$ ,  $-S(O)_2Q_{10}$ ,  $-S(O)Q_{10}$ ,  $-OS(O)_2Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-C(=NQ_{10})Q_{10}$ ,  $-S(O)_2-N=S(O)(Q_{10})_2$ ,  $-S(O)_2-N=S(Q_{10})_2$ ,  $-NQ_{10}Q_{10}$ ,  $-C(O)Q_{10}$ , -C(O)

The term "substituted amino" refers to an amino moiety in which one or both of the amino hydrogens are replaced with a group selected from -OQ<sub>10</sub>, -SQ<sub>10</sub>, -S(O)<sub>2</sub>Q<sub>10</sub>, -S(O)<sub>2</sub>Q<sub>10</sub>, -C(=NQ<sub>10</sub>)Q<sub>10</sub>, -C(=NOQ<sub>10</sub>)Q<sub>10</sub>, -S(O)<sub>2</sub>-N=S(O)<sub>2</sub>-N=S(Q)<sub>2</sub>-N=S(Q)<sub>10</sub>, -C(O)Q<sub>10</sub>, -C(O)

Each  $Q_{10}$  is independently selected from -H, alkyl, cycloalkyl, het, cycloalkenyl, and aryl. The het, alkyl, cycloalkyl, cycloalkenyl, and aryl being optionally substituted with 1-3 substitutuents selected from halo and  $Q_{13}$ .

Each  $Q_{11}$  is independently selected from -H, halo, alkyl, aryl, cycloalkyl, and het. The alkyl, aryl, cycloalkyl, and het being optionally substituted with 1-3 substituents independently selected from halo, -NO<sub>2</sub>, -CN, =S, =O, and  $Q_{14}$ .

Each  $Q_{13}$  is independently selected from  $Q_{11}$ ,  $-OQ_{11}$ ,  $-SQ_{11}$ ,  $-S(O)_2Q_{11}$ ,  $-C(O)_2Q_{11}$ ,  $-C(O)_2Q_{$ 

Each Q<sub>14</sub> is –H or a substituent selected from alkyl, cycloalkyl, phenyl, or naphthyl, each optionally substituted with 1-4 substituents independently selected from

15  $-NQ_{16}C(S)Q_{16}$ ,  $-NQ_{16}C(O)NQ_{16}Q_{16}$ ,  $-NQ_{16}C(S)NQ_{16}Q_{16}$ ,  $-S(O)_2NQ_{16}Q_{16}$ , and  $-NQ_{16}S(O)_2Q_{16}$ . The alkyl, cycloalkyl, and cycloalkenyl being furher optionally substituted with =O or =S.

Each Q<sub>15</sub> is alkyl, cycloalkyl, heterocycloalkyl, heteroaryl, phenyl, or naphthyl, each optionally substituted with 1-4 substituents independently selected from -F,

 $\begin{array}{lll} \text{20} & -\text{Cl, -Br, -I, -OQ}_{16}, -\text{SQ}_{16}, -\text{S(O)}_2\text{Q}_{16}, -\text{S(O)}_2\text{Q}_{16}, -\text{OS(O)}_2\text{Q}_{16}, -\text{C(=NQ}_{16})\text{Q}_{16}, \\ & -\text{S(O)}_2\text{-N=S(O)}(\text{Q}_{16})_2, -\text{S(O)}_2\text{-N=S(Q}_{16})_2, -\text{SC(O)}\text{Q}_{16}, -\text{NQ}_{16}\text{Q}_{16}, -\text{C(O)}\text{Q}_{16}, -\text{C(S)}\text{Q}_{16}, \\ & -\text{C(O)}\text{OQ}_{16}, -\text{OC(O)}\text{Q}_{16}, -\text{C(O)}\text{NQ}_{16}\text{Q}_{16}, -\text{C(S)}\text{NQ}_{16}\text{Q}_{16}, -\text{C(O)}\text{C(Q}_{16})_2\text{OC(O)}\text{Q}_{16}, -\text{CN}, \\ & -\text{CN}, \end{array}$ 

 $-NQ_{16}C(O)Q_{16}, -NQ_{16}C(S)Q_{16}, -NQ_{16}C(O)NQ_{16}Q_{16}, -NQ_{16}C(S)NQ_{16}Q_{16}, -NQ_{16}C(S)Q_{16}, -NQ_{16}Q_{16}, -NQ_{16}Q_$ 

S(O)<sub>2</sub>NQ<sub>16</sub>Q<sub>16</sub>, -NQ<sub>16</sub>S(O)<sub>2</sub>Q<sub>16</sub>, -NQ<sub>16</sub>S(O)Q<sub>16</sub>, -NQ<sub>16</sub>SQ<sub>16</sub>, -NO<sub>2</sub>, and -SNQ<sub>16</sub>Q<sub>16</sub>. The alkyl, cycloalkyl, and cycloalkenyl being furher optionally substituted with =O or =S.

Each Q<sub>16</sub> is independently selected from -H, alkyl, and cycloalkyl. The alkyl and cycloalkyl optionally including 1-3 halos.

30 Mammal denotes human and animals.

Each Q<sub>17</sub> is independently selected from –H, -OH, and alkyl optionally including 1-3 halos and -OH.

5

10

15

20

25

30

The term "electron withdrawing group" refers to the ability of a substituent to withdraw electrons relative to that of hydrogen if the hydrogen atom occupied the same position on the molecule. The term "electron withdrawing group" is well understood by one skilled in the art and is discussed in Advanced Organic Chemistry by J. March, John Wiley & Sons, New York, New York, (1985) and the discussion therein is incorporated herein by reference. Electron withdrawing groups include, but are not limited to, groups such as halo, nitro, carboxy, cyano, aryl optionally substituted, aromatic het (excluding pyridine) optionally substituted,  $-OC(Z_n)_3$ ,  $-C(Z_n)_2$ ,  $-C(Z_n)_3$ ,  $-(CO)-Q_{17}$ ,  $-SO_2-C(Z_n)_3$ ,  $-SO_2$ -aryl,  $-C(NQ_{17})Q_{17}$ ,  $-CH=C(Q_{17})_2$ ,  $-C\equiv C-Q_{17}$ , in which each Zn and Zm is independently H, halo, -CN,  $-NO_2$  -OH, or  $C_{1-4}$ alkyl optionally substituted with 1-3 halo, -OH,  $NO_2$ , and provided that at least one of Zn is halo, -CN, or  $NO_2$ , and further provided that  $Q_{17}$  is not -OH when the electron withdrawing group is  $-(CO)-Q_{17}$ .

It is to be understood that the present invention encompasses any racemic, optically-active, polymorphic, tautomeric, or stereoisomeric form, or mixture thereof, of a compound of the invention, which possesses the useful properties described herein.

In cases where compounds are sufficiently basic or acidic to form stable nontoxic acid or base salts, use of the compounds as pharmaceutically acceptable salts may be appropriate. Examples of pharmaceutically acceptable salts which are within the scope of the present invention include organic acid addition salts formed with acids which form a physiological acceptable anion and inorganic salts. Examples of pharmaceutically acceptable salts include, but are not limited to, the following acids acetic, aspartic, benzenesulfonic, benzoic, bicarbonic, bisulfuric, bitartaric, butyric, calcium edetate, camsylic, carbonic, chlorobenzoic, citric, edetic, edisylic, estolic, esyl, esylic, formic, fumaric, gluceptic, gluconic, glutamic, glycollylarsanilic, hexamic, hexylresorcinoic, hydrabamic, hydrobromic, hydrochloric, hydroiodic, hydroxynaphthoic, isethionic, lactic, lactobionic, maleic, malic, malonic, mandelic, methanesulfonic, methylnitric, methylsulfuric, mucic, muconic, napsylic, nitric, oxalic, p-nitromethanesulfonic, pamoic, pantothenic, phosphoric, monohydrogen phosphoric, dihydrogen phosphoric, phthalic, polygalactouronic, propionic, salicylic, stearic, succinic, sulfamic, sulfanilic, sulfonic, sulfuric, tannic, tartaric, teoclic toluenesulfonic, primary, secondary, and tertiary amines, substituted amines including

10

15

20

25

30

naturally occurring substituted amines, cyclic amines, such as arginine, betaine, caffeine, choline, N, N-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylamino-ethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, hydrabamine, isopropylamine, lysine, methylglucamine, morpholine, piperazine, piperidine, polyamine resins, procaine, purines, theobromine, triethylamine, trimethylamine, tripropylamine, and the like.

Pharmaceutically acceptable salts may be obtained using standard procedures well known in the art, for example by reacting a sufficiently basic compound such as an amine with a suitable acid affording a physiologically acceptable anion. Alkali metal (for example, sodium, potassium or lithium) or alkaline earth metal (for example calcium) salts of carboxylic acids can also be made.

The antibacterial agents of this invention have useful activity against a variety of organisms. The in vitro activity of compounds of this invention can be assessed by standard testing procedures such as the determination of minimum inhibitory concentration (MIC) by agar dilution as described in "Approved Standard. Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria That Grow Aerobically", 3rd. ed., published 1993 by the National Committee for Clinical Laboratory Standards, Villanova, Pennsylvania, USA.

The antibacterial agents described herein are useful for sterilization, sanitation, antisepsis, and disinfection. The antibacterial agents can be applied to a location in need of sterilization, sanitation, antisepsis, or disinfection, by methods known to those skilled in the art. For instance, the antibacterial agents may be incorporated into a cleaning solution that is applied, such as by spraying or pouring, to an item in need of sterilization, sanitation, antisepsis, or disinfection. The antibacterial agents may be used alone or in combination, e.g., agents disclosed herein with one another or agent(s) disclosed herein with other antibacterial agents. The antibacterial agents may be applied in varying concentrations depending upon the bacterial susceptibility to antibacterial agent(s) being applied and the desired level of sterilization, sanitation, antisepsis, or disinfection.

The antibacterial compounds of this invention may be synthesized by various methods known to those skilled in the art. Non-limiting examples of synthetic schemes for producing the antibacterial agents are described below.

#### **EXAMPLES**

Without further elaboration, it is believed that one skilled in the art can, using the preceding description, practice the present invention to its fullest extent. The following detailed examples describe how to prepare the various compounds and/or perform the various processes of the invention and are to be construed as merely illustrative, and not limitations of the preceding disclosure in any way whatsoever. Those skilled in the art will promptly recognize appropriate variations from the procedures both as to reactants and as to reaction conditions and techniques.

10

20

25

#### **Example 1: Sulfonyl Derivatives**

### Scheme 1.1

OH

$$O \cap OH$$
 $O \cap OH$ 
 $O \cap OH$ 

a) oxalyl chloride; b) Methyl-2-amino-5-bromobenzoate; c) HN(Q<sub>17</sub>)<sub>2</sub>; d) KOH

# 15 Methyl 5-bromo-2-{[4-(chlorosulfonyl)benzoyl]amino}benzoate

Methyl 5-bromo-2-{[4-(chlorosulfonyl)benzoyl]amino} benzoate (1) was prepared as a common intermediate for the formation of sulfonamides by the procedure below: 4-(chlorosulfonyl)benzoic acid (18.37 g, 8.33 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (140 mL) and 4 drops of DMF. The solution was cooled to 0° C and oxalyl chloride (1.8 mL, 20.6 mmol) was added and stirred for 1 hour, removed from ice bath, and stirred overnight. The clear solution was concentrated *in vacuo*, redissolved in CH<sub>2</sub>Cl<sub>2</sub>, and concentrated *in vacuo*. The resulting product was dissolved in toluene (140 mL) and refluxed for 30 minutes to remove any HCl gas. After cooling to room temperature, methyl-2-amino-5-bromobenzoate (15.96 g, 69.4 mmol) was added, and the

suspension was refluxed overnight. The suspension was cooled to  $0^0$  C and filtered, washing with toluene and quickly with ethyl acetate. The solid was dried in a vacuum oven overnight to obtain sulfonyl chloride **1** (19.8 g, 66%). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  12.19, 8.82, 8.27-8.19, 7.73, 4.00; IR 1700, 1683, 1604, 1585, 1524 (s), cm<sup>-1</sup>; MS (ESI-) for  $C_{15}H_{11}BrClNO_5S$  m/z 429.8 (M-H)<sup>-</sup>.

# General Method A (sulfonamide preparation with anilines, primary, and secondary amines)

### 10 Methyl 5-bromo-2-({4-[(diethylamino)sulfonyl]benzoyl}amino)benzoate.

To a solution of the sulfonyl chloride 1, (694.1 mg, 1.61 mmol, 1.0 eq) in toluene (4.0 mL) was added diethyl amine (500  $\mu$ L, 4.83 mmol, 3.0 eq). The suspension was shaken at 50° C for overnight. The product was extracted with EtOAc, washed with 1 N HCl and water, and concentrated *in vacuo*. The compound was dried in a vacuum oven at 50° C overnight to obtain 624.4 mg (83%). <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  11.48, 8.31, 8.11, 8.05, 7.99, 7.87, 3.86, 3.20, 1.04; IR 1700, 1676 (s), 1600, 1519 (s), 1338, 1330, 1306 (s), cm<sup>-1</sup>. Anal. Calcd for C<sub>19</sub>H<sub>21</sub>BrN<sub>2</sub>O<sub>5</sub>S: C, 48.62; H, 4.51; N, 5.97; Br, 17.02; S, 6.83. Found: C, 48.76; H, 4.53; N, 5.89; Br, 16.98; S, 6.73.

# 20 General Method B (hydrolysis of the methyl ester)

5

15

25

30

## 5-bromo-2-({4-[(diethylamino)sulfonyl]benzoyl}amino)benzoate, 8.

Methyl 5-bromo-2-({4-[(diethylamino)sulfonyl]benzoyl}amino)benzoate (329.6 mg, 0.704 mmol) was dissolved in 2 mL of dioxane and 0.2 mL of water. KOH (1 pellet, ~80 mg) was added to the mixture as it was heated at  $50^{\circ}$  C for 3 hours. The reaction was cooled, extracted with EtOAc, washed with 1 N HCl and brine, dried (Na<sub>2</sub>SO<sub>4</sub>), concentrated *in vacuo*, and dried in a vacuum oven at  $50^{\circ}$  C overnight to yield 313.8 mg (98%). <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  12.05, 8.55, 8.11, 8.09, 8.00, 7.86, 3.19, 1.04; IR 1703, 1661, 1202, 1185, cm<sup>-1</sup>. MS (FAB) m/z (rel. intensity) 455 (MH<sup>+</sup>, 45), 457 (37), 455 (45), 240 (99). HRMS (FAB) calcd for

C<sub>18</sub>H<sub>19</sub>BRN<sub>2</sub>O<sub>5</sub>S +H<sub>1</sub> 455.0276, found 455.0260. Anal. Calcd for C<sub>18</sub>H<sub>19</sub>BrN<sub>2</sub>O<sub>5</sub>S: C, 47.48; H, 4.21; N, 6.15; Br, 17.55; S, 7.04. Found: C, 47.31; H, 4.25; N, 6.12.

5-bromo-2-({4-[(dimethylamino)sulfonyl]benzoyl}amino)benzoic acid 6, was prepared by method B from its methyl ester, i.e., Methyl 5-bromo-2-({4-

- 5 [(dimethylamino)sulfonyl]benzoyl}amino)benzoate, in a 47% yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.89, 8.31, 8.18, 7.96, 7.78, 2.78; IR 3135, 1700, 1350 (s), 1191 (s), cm<sup>-1</sup>. MS (ESI-) for C<sub>16</sub>H<sub>15</sub>BrNO<sub>5</sub>S *m/z* 426.9 (M-H, Br isotope)<sup>-</sup>. Anal. Calcd for C<sub>16</sub>H<sub>15</sub>BrN<sub>2</sub>O<sub>5</sub>S: C, 44.98; H, 3.54; N, 6.56; Br, 18.70; S, 7.50. Found: C, 44.82; H, 3.55; N, 6.46; Br, 18.43; S, 7.36.
- 5-bromo-2-({4-[(1H-indol-5-ylamino)sulfonyl]benzoyl}amino)benzoate 7, was prepared by general method B from PNU-263551 in a 52% yield. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 12.05 (s, 1 H), 11.05 (s, 1 H), 10.00 (s, 1 H), 8.52 (d, *J* = 9 Hz, 1 H), 8.10 (d, *J* = 2 Hz, 1 H), 8.02 (d, *J* = 8 Hz, 2 H), 7.85 (m, 3 H), 7.30 (t, *J* = 1 Hz, 1 H), 7.25440 (s, 1 H), 7.24 (d, *J* = 9 Hz, 1 H), 6.82 (dd, *J* = 9, 1 Hz, 1 H), 6.34 (s, 1 H); 18 1687, 1664, 1607, 1524, 1338, 1314, 1300, 1189, 1170 (s), 825, 801, 756, 743, 681, 616 (s), cm<sup>-1</sup>. MS (FAB) *m/z* (rel. intensity) 514 (MH<sup>+</sup>, 55), 516 (59), 515 (67), 514 (55), 132 (99), 131 (97). HRMS (FAB) calcd for C<sub>22</sub>H<sub>16</sub>BRN<sub>3</sub>O<sub>5</sub>S +H<sub>1</sub> 514.0073, found 514.0066. HPLC [1] shows one main peak at 16.3 min (95%). Anal. Calcd for C<sub>22</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>5</sub>S: C, 51.37; H, 3.13; N, 8.17; Br, 15.53; S, 6.23. Found: C, 51.16; H, 3.23; N, 8.01.
- **5-bromo-2-[(4-{[(3-furylmethyl)amino]sulfonyl}benzoyl)amino]benzoate 9,** was prepared by method B from PNU-276173 in a 48% yield. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.60 (d, J = 9 Hz, 1 H), 8.41 (t, J = 6 Hz, 1 H), 8.14 (d, J = 2 Hz, 1 H), 8.07 (d, J = 8 Hz, 2 H), 7.93 (d, J = 8 Hz, 2 H), 7.87 (dd, J = 9, 2 Hz, 1 H), 7.46 (s, 1 H), 6.28 (s, 1 H), 6.18 (s, 1 H), 4.08 (d, J = 6 Hz, 2 H); IR 3252, 1702, 1172 (s), 1165 (s), cm<sup>-1</sup>. MS (FAB) m/z (rel. intensity) 479 (MH<sup>+</sup>, 13), 481 (14), 479 (13), 135 (99), 73 (64). HRMS (FAB) calcd for  $C_{19}H_{15}BRN_2O_6S + H_1$  478.9913, found 478.9922. Anal. Calcd for  $C_{19}H_{15}BrN_2O_6S$ : C, 47.61; H, 3.15; N, 5.84; Br, 16.67; S, 6.69. Found: C, 47.55; H, 3.22; N, 5.69; Br, 16.26; S, 6.60.
- 5-bromo-2-[(4-{[4-(ethoxycarbonyl)-1-piperazinyl]sulfonyl}benzoyl)amino]
  benzoic acid 10 was prepared by method A followed by B with a 26% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)

 $\delta$  8.60 (d, J = 9 Hz, 1 H), 8.18 (d, J = 8 Hz, 2 H), 8.13 (d, J = 2 Hz, 1 H), 7.94 (d, J = 8 Hz, 2 H), 7.79 (dd, J = 9, 2 Hz, 1 H), 3.97 (q, J = 7 Hz, 2 H), 3.45 (br. s, 4 H), 2.95 (br. s, 4 H), 1.12 (t, J = 7 Hz, 3 H); IR 1692 (s), 1675 (s), 1584, 1518 (s), 1287, 1276, 1250, cm<sup>-1</sup>. MS (FAB) m/z (rel. intensity) 540 (MH<sup>+</sup>, 46), 542 (44), 540 (46), 159 (95), 157 (99). HRMS (FAB) calcd for C<sub>21</sub>H<sub>22</sub>BRN<sub>3</sub>O<sub>7</sub>S +H<sub>1</sub> 540.0440, found 540.0428. HPLC [1] shows one major peak at 16.2 min (97%). Anal. Calcd for C<sub>21</sub>H<sub>22</sub>BrN<sub>3</sub>O<sub>7</sub>S: C, 46.67; H, 4.10; N, 7.78; Br, 14.79; S, 5.93. Found: C, 46.34; H, 4.19; N, 7.63; Br, 14.18; S, 5.79.

5

30

# $5-bromo-2-\{[4-(\{methyl[2-(2-pyridinyl)ethyl]amino\}sulfonyl)benzoyl]amino\}$

benzoic acid 11 was prepared by method A followed by B with a 57% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 12.19 (s, 1 H), 8.58 (d, *J* = 9 Hz, 1 H), 8.52 (d, *J* = 4 Hz, 1 H), 8.13 (d, *J* = 3 Hz, 1 H), 8.12 (d, *J* = 6 Hz, 2 H), 7.96 (d, *J* = 8 Hz, 2 H), 7.87 (dd, *J* = 9, 2 Hz, 1 H), 7.78 (td, *J* = 8, 2 Hz, 1 H), 7.35 (d, *J* = 8 Hz, 1 H), 7.30 (td, *J* = 6, 2 Hz, 1 H), 3.42 (t, *J* = 7 Hz, 2 H), 2.99 (t, *J* = 8 Hz, 2 H), 2.77 (s, 3 H); IR 1692 (s), 1518 (s), 1340 (s), 1297 (s), 1162 (s), 763 (s), 755 (s), 747 (s) cm<sup>-1</sup>. MS (ES-) for C<sub>22</sub>H<sub>20</sub>BrN<sub>3</sub>O<sub>5</sub>S *m/z* 518.0 (M-H<sup>+</sup>, Br isotope); HRMS (FAB) calcd for C<sub>22</sub>H<sub>20</sub>BRN<sub>3</sub>O<sub>5</sub>S +H<sub>1</sub> 518.0386, found 518.0388. HPLC [1] shows one major peak (13.58 min, 99%).

# 2-({4-[(benzylamino)sulfonyl]benzoyl}amino)-5-bromobenzoic acid 12 was

prepared by method A followed by B with a 17% yield over both steps. The methyl ester was not fully characterized.  $^{1}$ H NMR (300 MHz, DMSO- $d_{6}$ )  $\delta$  12.09 (s, 1 H), 8.60 (d, J = 9 Hz, 1 H), 8.39 (t, J = 6 Hz, 1 H), 8.14 (d, J = 2 Hz, 1 H), 8.08 (d, J = 8 Hz, 2 H), 7.97 (d, J = 8 Hz, 2 H), 7.88 (dd, J = 9, 2 Hz, 1 H), 7.30-7.20 (m, 5 H), 4.05 (d, J = 6 Hz, 2 H); HRMS (FAB) calcd for  $C_{21}H_{17}BRN_{2}O_{5}S$  + $H_{1}$  489.0120, found 489.0129; HPLC [1] shows one major peak (20.60 min, 99%).

# 5-bromo-2-[(4-{[(2-hydroxy-1-methylethyl)amino]sulfonyl}benzoyl)amino]

**benzoic acid 14** was prepared by method A followed by B with a 35% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.56 (d, J = 9 Hz, 1 H), 8.11 (d, J = 2 Hz, 1 H), 8.09 (d, J = 8 Hz, 2 H), 8.00 (d, J = 8 Hz, 2 H), 7.86 (dd, J = 9, 2 Hz, 1 H), 7.76 (d, J = 7 Hz, 1 H), 3.26 (m, 2 H), 3.12 (m, 1 H), 0.89 (d, J = 6 Hz, 3 H); MS (ES-) for  $C_{17}H_{17}BrN_2O_6S$  m/z 454.9 (M-H<sup>+</sup>); HPLC [1] shows one major peak (14.08 min, 96%).

5-bromo-2-({4-[(4-carboxyanilino)sulfonyl]benzoyl}amino)benzoic acid 15 was prepared from method A followed by method B in a 10% yield. The methyl ester was not fully characterized.  $^{1}$ H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  12.45 (br. s, 1 H), 11.15 (s, 1 H), 8.52 (d, J = 9 Hz, 1 H), 8.08 (d, J = 8 Hz, 3 H), 8.03 (d, J = 9 Hz, 2 H), 7.81 (d, J = 9 Hz, 3 H), 7.26 (d, J = 9 Hz, 2 H); HPLC [1] shows one major peak (15.15 min, 90%).

5

30

5-bromo-2-{[4-(3,4-dihydro-1(2H)-quinolinylsulfonyl)benzoyl]amino}benzoic acid 16 was prepared by method A followed by method B in a 48% yield. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 12.05 (s, 1 H), 8.52 (d, *J* = 9 Hz, 1 H), 8.11 (d, *J* = 3 Hz, 1 H), 8.05 (d, *J* = 9 Hz, 2 H), 7.86 (dd, *J* = 9, 2 Hz, 1 H), 7.82 (d, *J* = 8 Hz, 2 H), 7.61 (d, *J* = 8 Hz, 1 H), 7.25-7.05 (m, 3 H), 3.83 (t, *J* = 6 Hz, 2 H), 2.45 (t, *J* = 7 Hz, 2 H), 1.63 (quintet, *J* = 6 Hz, 2 H); IR 1667, 1601, 1584, cm<sup>-1</sup>. HRMS (FAB) calcd for C<sub>23</sub>H<sub>19</sub>BRN<sub>2</sub>O<sub>5</sub>S +H<sub>1</sub> 515.0276, found 515.0264. Anal. Calcd for C<sub>23</sub>H<sub>19</sub>BrN<sub>2</sub>O<sub>5</sub>S: C, 53.60; H, 3.72; N, 5.43. Found: C, 53.52; H, 3.96; N, 5.57.

5-bromo-2-{[4-({[2-(3,5-dimethoxyphenyl)ethyl]amino}sulfonyl)benzoyl]amino} benzoic acid 17 was prepared by method A followed by B with a 56% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 8.60 (d, *J* = 9 Hz, 1 H), 8.13 (d, *J* = 3 Hz, 1 H), 8.09 (d, *J* = 8 Hz, 2 H), 7.95 (d, *J* = 9 Hz, 2 H), 7.87 (dd, *J* = 9, 2 Hz, 1 H), 6.79 (d, *J* = 8 Hz, 1 H), 6.73 (d, *J* = 2 Hz, 1 H), 6.64 (dd, *J* = 8, 2 Hz, 1 H), 3.70 (s, 3 H), 3.68 (s, 3 H), 3.02 (q, *J* = 6 Hz, 2 H), 2.61 (t, *J* = 7 Hz, 2 H); MS (FAB) *m/z* (rel. intensity) 563 (MH<sup>+</sup>, 86), 565 (86), 564 (82), 563 (86), 562 (56), 348 (77), 199 (46), 165 (56), 164 (32), 152 (49), 151 (99). HRMS (EI) calcd for C<sub>24</sub>H<sub>23</sub>BRN<sub>2</sub>O<sub>7</sub>S 562.0410, found 562.0438. HPLC [1] shows one major peak (16.16 min, 97%).

**5-bromo-2-[(4-{[(3S)-3-hydroxypyrrolidinyl]sulfonyl}benzoyl)amino]benzoic acid 13** was prepared by method A followed by B in a 15% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.62 (d, J = 9 Hz, 1 H), 8.18 (d, J = 8 Hz, 2 H), 8.11 (d, J = 3 Hz, 1 H), 7.92 (d, J = 11 Hz, 2 H), 7.78 (dd, J = 9, 2 Hz, 1 H), 5.16 (m, 1 H), 3.50-3.20 (m, 4 H), 2.10-1.90 (m, 2 H); HPLC [1] shows one major peak (18.94 min, 97%).

5

20

**5-bromo-2-({4-[(ethylanilino)sulfonyl]benzoyl}amino)benzoic acid 19** was prepared by method A followed by B with a 75% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD)  $\delta$  8.75 (d, J = 9 Hz, 1 H), 8.24 (d, J = 2 Hz, 1 H), 8.11 (d, J = 8 Hz, 2 H), 7.76 (dd, J = 9, 2 Hz, 1 H), 7.74 (d, J = 8 Hz, 2 H), 7.34 (m, 3 H), 7.06 (m, 2 H), 3.69 (q, J = 7 Hz, 2 H), 1.07 (t, J = 7 Hz, 3 H); MS (ES-) for C<sub>22</sub>H<sub>19</sub>BrN<sub>2</sub>O<sub>5</sub>S m/z 502.8 (M-H<sup>+</sup>; Br isotope); HRMS (FAB) calcd for C<sub>22</sub>H<sub>19</sub>BRN<sub>2</sub>O<sub>5</sub>S +H<sub>1</sub> 503.0276, found 503.0265. HPLC [1] shows one major peak (18.60 min, 99%).

5-bromo-2-({4-[(3,5-dimethoxyanilino)sulfonyl]benzoyl}amino)benzoic acid 20
was prepared by method A followed by B with a 69% yield over both steps. The
methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 8.73 (d, *J* = 9 Hz, 1 H), 8.24 (d, *J* = 2 Hz, 1 H), 8.09 (d, *J* = 9 Hz, 2 H), 7.96 (d, *J* = 9 Hz, 2 H),
7.74 (dd, *J* = 9, 2 Hz, 1 H), 6.32 (s, 1 H), 6.31 (s, 1 H), 6.20 (s, 1 H), 3.70 (s, 6 H); MS
(ES-) for C<sub>22</sub>H<sub>19</sub>BrN<sub>2</sub>O<sub>7</sub>S *m/z* 532.8 (M-H<sup>+</sup>); HPLC [1] shows one major peak (17.06 min, 96%).

**5-bromo-2-[(4-{[(2-hydroxy-2-phenylethyl)(methyl)amino]sulfonyl} benzoyl)amino] benzoic acid 21** was prepared by method A followed by B with a 15% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD)  $\delta$  12.10 (s, 1 H), 8.57 (d, J = 9 Hz, 1 H), 8.12 (d, J = 2 Hz, 1 H), 8.11 (d, J = 9 Hz, 2 H), 7.95 (d, J = 8 Hz, 2 H), 7.87 (dd, J = 9, 3 Hz, 1 H), 7.35-7.27 (m, 5 H), 4.76 (t, J = 7 Hz, 1 H), 3.22-3.13 (m, 2 H), 2.77 (s, 3 H); MS (FAB) m/z (rel. intensity) 533 (MH<sup>+</sup>, 61), 535 (64), 533 (61), 517 (99), 516 (24), 515 (90), 318 (46), 152 (27), 134 (25), 132 (33), 44 (44). HRMS (FAB) calcd for C<sub>23</sub>H<sub>21</sub>BRN<sub>2</sub>O<sub>6</sub>S +H<sub>1</sub> 533.0382, found 533.0386. HPLC [1] shows one major peak (17.06, 97%).

5-bromo-2-{[4-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid
22 was prepared by method A followed by B in a 55% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 12.00 (s, 1 H), 8.51 (d, *J* = 9 Hz, 1 H), 8.10-8.01 (m, 5 H), 7.84 (dd, *J* = 9, 3 Hz, 1 H), 7.50 (d, *J* = 8 Hz, 1 H), 7.22 (t, *J* = 8 Hz, 1 H), 7.17 (d, *J* = 8 Hz, 1 H), 7.00 (1, *J* = 7 Hz, 1 H), 3.98 (t, *J* = 8 Hz, 2 H), 2.93 (t, *J* = 8 Hz, 2 H); IR 1687, 1667, 1601, 1525 (s), 1365 (s), 1245 (s), 1172 (s), cm<sup>-1</sup>. MS (FAB) *m/z* (rel. intensity) 501 (MH<sup>+</sup>, 36), 503 (41), 502 (43), 501 (36), 500 (31), 286 (35), 118 (99). HRMS (FAB) calcd for

25

30

 $C_{22}H_{17}BRN_2O_5S + H_1$  501.0120, found 501.0118. Anal. Calcd for  $C_{22}H_{17}BrN_2O_5S$ : C, 52.71; H, 3.42; N, 5.59; Br, 15.94; S, 6.39. Found: C, 52.65; H, 3.47; N, 5.58; Br, 15.88; S, 6.24.

### 5-bromo-2-({4-[(5-methoxy-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)

- benzoic acid 23 was prepared by method A followed by B in a 17% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 12.05 (s, 1 H), 8.51 (d, J = 9 Hz, 1 H), 8.10 (d, J = 2 Hz, 1 H), 8.05 (d, J = 8 Hz, 2 H), 7.95 (d, J = 9 Hz, 2 H), 7.86 (dd, J = 9, 2 Hz, 1 H), 7.42 (d, J = 9 Hz, 1 H), 6.78 (d, J = 8 Hz, 1 H), 6.77 (s, 1 H), 3.96 (t, J = 8 Hz, 2 H), 3.68 (s, 3 H), 2.80 (t, J = 8 Hz, 2 H); IR 1702, 1606, 1518, 1489 (s), 1358, 1199 (s), 1168 (s), cm<sup>-1</sup>. MS (FAB) m/z (rel. intensity) 531 (MH<sup>+</sup>, 29), 533 (30), 531 (29), 530 (38), 148 (99). HRMS (EI) calcd for C<sub>23</sub>H<sub>19</sub>BRN<sub>2</sub>O<sub>6</sub>S 530.0148, found 530.0156. Anal. Calcd for C<sub>23</sub>H<sub>19</sub>BrN<sub>2</sub>O<sub>6</sub>S: C, 51.99; H, 3.60; N, 5.27; Br, 15.04; S, 6.03. Found: C, 52.08; H, 3.61; N, 5.29.
- 5-bromo-2-({4-[(5-fluoro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)
  benzoic acid 24 was prepared by method A followed by B with a 41% yield over both steps. The methyl ester was not fully characterized. <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)
  δ 12.05 (s, 1 H), 8.51 (d, J = 9 Hz, 1 H), 8.10 (d, J = 2 Hz, 1 H), 8.07 (d, J = 9 Hz, 2 H), 7.99 (d, J = 9 Hz, 2 H), 7.85 (dd, J = 9, 2 Hz, 1 H), 7.49 (dd, J = 10, 5 Hz, 1 H),
  7.07-7.02 (m, 2 H), 4.01 (t, J = 8 Hz, 2 H), 2.89 (t, J = 8 Hz, 2 H); MS (ES-) for C<sub>22</sub>H<sub>16</sub>BrN<sub>2</sub>O<sub>5</sub>S m/z 518.9 (M-H<sup>+</sup>, Br isotope); HPLC [2] shows one major peak (6.35 min, 96%).
  - **2-{[4-(1H-benzimidazol-1-ylsulfonyl)benzoyl]amino}-5-bromobenzoic acid 26** was prepared from method A followed by hydrolysis of the methyl ester by the hydrolysis procedure in method C below. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  11.98 (s, 1 H), 8.91 (s, 1 H), 8.47 (d, J = 9 Hz, 1 H), 8.41 (d, J = 9 Hz, 2 H), 8.13 (d, J = 9 Hz, 2 H), 8.09 (d, J = 2 Hz, 1 H), 7.93 (d, J = 7 Hz, 1 H), 7.85 (dd, J = 9, 3 Hz, 1 H), 7.78 (d, J = 7 Hz, 1 H), 7.47 (t, J = 6 Hz, 1 H), 7.40 (t, J = 6 Hz, 1 H); IR 1686, 1607, 1522, 1391, 1296, 1262, 1190, cm<sup>-1</sup>. MS (ESI-) for C<sub>21</sub>H<sub>14</sub>BrN<sub>3</sub>O<sub>5</sub>S m/z 497.7 (M-H)<sup>-</sup>. HPLC [2] shows one major peak at 6.01 min (96%). Anal. Calcd for C<sub>21</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>5</sub>S: C, 50.21; H, 3.21; N, 8.36; Br, 15.91; S, 6.38. Found: C, 50.06; H, 2.85; N, 7.93; Br, 15.34; S, 6.22.

### General Method C (sulfonamide preparation with indoles and pyrrole):

Reaction of sulfonyl chloride intermediate 1 with indole derivatives requires modified conditions. Deprotonation of the indole nitrogen with sodium hydride in THF and reaction with the sulfonyl chloride 1 provided the desired intermediate methyl esters. Two equivalents of the indole anion were required because of competitive deprotonation of the amide in 1. Attempted hydrolysis of such methyl esters with aqueous KOH results in hydrolysis of the newly formed sulfonamide. Therefore, dealkylative deesterification conditions were utilized (Scheme 1.2).

#### Scheme 1.2

10

15

20

5

Br 
$$CO_2Me$$
 Br  $CO_2H$   $1)$  a  $26-38$ 

a) R\*, NaH, THF; b) Mel, NaCN

\* R = indoles, pyrrole, indazole, and benzoxazolinone

5-bromo-2-({4-[(5-fluoro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid 26 was prepared by the following procedure: 5-fluoroindole (497.1 mg, 3.68 mmol, 2.2 eq) was dissolved in anhydrous THF (8 mL) and cooled to 0° C. NaH (60% dispersion in mineral oil, 150 mg, 3.75 mmol, 2.2 eq) was added and the cloudy mixture was stirred for 1 hr. at 0-25° C. The suspension was then cooled to 0° C and Methyl 5-bromo-2-{[4-(chlorosulfonyl)benzoyl]amino}benzoate (722.0 mg, 1.68 mmol, 1.0 eq) was added neat and stirred overnight at room temperature. After quenching with water, the product was extracted with EtOAc and washed with 1 N HCl, concentrated *in vacuo*, triturated with MeOH, filtered and washed with MeOH. A mixture of the carboxylic acid and ester (469.0 mg) was obtained. The mixture of products were both committed to the hydrolysis conditions: 4 mL dioxane, 0.4 mL water, and 1 KOH pellet (~90 mg) were added to the mixture of acid and ester and shook at 50° C for 3 hrs. The hydrolysis was monitored by HPLC. The product was

5

10

dissolved in EtOAc and washed with 1 N HCl, concentrated *in vacuo*, triturated with MeOH, filtered, and washed with MeOH to obtain 246.8 mg (28%) of 5-bromo-2-({4-[(5-fluoro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  11.95 (s, 1 H), 8.43 (d, J = 9 Hz, 1 H), 8.19 (d, J = 9 Hz, 2 H), 8.07 (d, J = 3 Hz, 1 H), 8.05 (d, J = 9 Hz, 2 H), 7.96 (dd, J = 9, 4 Hz, 1 H), 7.91 (d, J = 4 Hz, 1 H), 7.82 (dd, J = 9, 2 Hz, 1 H), 7.42 (dd, J = 9, 3 Hz, 1 H), 7.20 (td, J = 9, 3 Hz, 1 H), 6.86 (d, J = 4 Hz, 1 H); IR (drift) 1692, 1670, 1601, 1524 (s), 1462, 1388 (s), 1290, 1242, 1234, 1218 (s), 1181 (s), 1140 (s), 742, 649 (s), 607 (s), cm<sup>-1</sup>. MS (ESI-) for  $C_{22}H_{14}BrFN_2O_5S$  m/z 516.9 (M-H, Br isotope). HPLC [2] shows one major peak at 6.56 min (98%). Anal. Calcd for  $C_{22}H_{14}BrFN_2O_5S$ : C, 51.08; H, 2.73; N, 5.41; Br, 15.44; S, 6.20. Found: C, 51.05; H, 2.64; N, 5.39.

Other compounds were prepared by the above procedure making non-critical variations.

5-bromo-2-{[4-(1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid, 5-bromo-2-({4-[(5-chloro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid, 5-bromo-2-({4-[(6-chloro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid, 5-bromo-2-({4-[(6-chloro-5-fluoro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid, 5-bromo-2-({4-[(6-chloro-5-fluoro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid, 5-bromo-2-{[4-(1H-pyrrol-1-ylsulfonyl]benzoyl]amino}benzoic acid, 5-bromo-2-{[4-(1H-pyrrolo[2,3-b]pyridin-1-ylsulfonyl]benzoyl]amino}benzoic acid

#### Scheme 1.3

$$R^2$$
 $R^2$ 
 $R^2$ 

Preparation of Methyl 3-bromo-5-[(5-bromo-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoate

- A solution of 5-bromoindoline (528 mg, 2.67 mmol, Lancaster) and triethylamine (650 μL, 4.67 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL) was added to a solution of methyl 3-bromo-5- (chlorosulfonyl)benzoate (737 mg, 2.35 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL). The mixture was stirred overnight and then diluted to 100 mL with CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 50% CH<sub>2</sub>Cl<sub>2</sub>/heptane to 75% CH<sub>2</sub>Cl<sub>2</sub>/heptane as eluent. Yield was 945 mg of pale yellow solid.
- Preparation of 3-Bromo-5-[(5-bromo-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoic acid

To a mixture of the corresponding methyl ester (841 mg, 1.77 mmol) in methanol (20 mL) was added 1 M aqueous sodium hydroxide (3.0 mL). The mixture was stirred in a 50 °C oil bath for 10 minutes and then at 60 °C for 15 minutes. The mixture was still a slurry, so 10 mL of dioxane was added. Heat was removed after an additional 40 minutes. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The organics were washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. They were then dried over MgSO<sub>4</sub> and evaporated yielding 807 mg of white solid.

# Methyl 5-bromo-2-({3-bromo-5-[(5-bromo-2,3-dihydro-1H-indol-1-l)sulfonyl]benzoyl}amino)benzoate

5

10

15

20

25

To 3-bromo-5-[(5-bromo-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoic acid (583 mg, 1.26 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) was added DMF (20 μL) and oxalyl chloride (220 μL, 2.52 mmol). The mixture was stirred for 1 hour, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (10 mL), and methyl 2-amino-5-bromobenzoate (267 mg, 1.16 mmol, Avocado) in pyridine (4 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. Some THF was added to help solubility. This mixture was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The organics were evaporated, and the residue was dissolved in hot THF. This solution was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 50% CH<sub>2</sub>Cl<sub>2</sub>/heptane to 100% CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 603 mg of white solid.

General Method D: (hydrolysis of the methyl ester)

# 5-Bromo-2-({3-bromo-5-[(5-bromo-2,3-dihydro-1H-indol-1-yl)sulfonyl}benzoyl}amino)benzoic acid

5

10

15

20

To a mixture of the corresponding methyl ester (374 mg, 0.556 mmol) in dioxane (30 mL) was added 1 M aqueous sodium hydroxide (1.1 mL). The mixture was stirred in a 60 °C oil bath for 90 minutes. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The organics were washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. They were then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol/THF. The solids were washed with ethanol and then dried at 100 °C under vacuum yielding 266 mg of white solid.  $^{1}$ H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.14 (s, 1 H), 8.48 (d, J = 8.7 Hz, 1 H), 8.36 (s, 1 H), 8.31 (s, 1 H), 8.19 (s, 1 H), 8.12 (d, J = 2.0 Hz, 1 H), 7.86 (dd, J = 8.7, 2.5 Hz, 1 H), 7.39-7.49 (m, 3 H), 4.04 (t, J = 8.4 Hz, 2 H), 2.99 (t, J = 8.4 Hz, 2 H).

# Preparation of Methyl 4-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoate

To 4-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoic acid (456 mg, 1.35 mmol) in  $CH_2Cl_2$  (30 mL) was added DMF (15  $\mu$ L) and oxalyl chloride (150  $\mu$ L, 1.72 mmol). The mixture was stirred for 5 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in  $CH_2Cl_2$ 

(10 mL). Methanol (2 mL) and pyridine (2 mL) in CH<sub>2</sub>Cl<sub>2</sub> (6 mL) were added. The mixture was stirred for 30 minutes and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 100 mL of 1 M aqueous HCl, 100 mL of saturated aqueous NaHCO<sub>3</sub>, another 100 mL of HCl, and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was dried over MgSO<sub>4</sub> and evaporated yielding 464 mg of white solid.

# Preparation of {4-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}methanol

10

5

To a solution of methyl 4-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoate (396 mg, 1.13 mmol) in THF (20 mL) was added lithium borohydride (0.40 mL of 2.0 M solution in THF, 0.80 mmol, Aldrich). HPLC analysis after 1.5 hours indicated <10% reaction, so lithium aluminum hydride (0.60 mL of 1.0 M solution in THF, Aldrich) was added at -78 °C. The mixture was stirred at -78 °C for 15 minutes and then warmed to room temperature. The reaction was quenched by the addition of water (25 μL) followed by 6 M aqueous NaOH (25 μL) followed by another portion of water (75 μL). The mixture was filtered, and the filtrate was evaporated in the presence of silica gel. The product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from CH<sub>2</sub>Cl<sub>2</sub> to 10% EtOAC in CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 290 mg of white solid.

#### Preparation of t-butyl 2-nitrobenzoate

A 22 L round bottom flask, equipped with an mechanical stirrer, thermocouple, and a 1 L addition funnel, was charged with 500 g (2.99 moles, 1.0 equiv) of 2-nitrobenzoic acid (Avocado Research Chemicals Ltd, 98%) and 1.44 kg (11.97 moles, 4 equiv) of anhydrous magnesium sulfate (EM Science, 98%). To the solids were charged 12.5 L (25 mL/g) of CH<sub>2</sub>Cl<sub>2</sub> (EM Science, 99.96%) and 1.43 L (2.99 moles, 1.0 equiv) of tbutyl alcohol (Aldrich, 99 + % A.C.S. Reagent). The addition funnel was charged with 1.59 mL (2.99 moles, 1.0 equiv) of concentrated sulfuric acid (Mallinckrodt, 95.7%) and the entire system was sealed via use of a Teflon cap (loose fit; internal pressure does not exceed 11 psi; theory = 10.5 psi). The resulting suspension was cooled to 16 °C using a water bath and 159 mL (2.99 moles, 1.0 equiv) of concentrated sulfuric acid was added at a rate of 2.8 mL/min, maintaining an internal temperature less than 25 °C. The resulting off-white suspension was stirred at room temperature for 14 hours at which time the HPLC assay indicated the reaction was at 92% conversion. The suspension was sparged with nitrogen for 15 min using ½ inch ID Teflon tubing and filtered through a sintered glass funnel (course) with the aid of house vacuum (ca. 16 torr; filtration time of 1.0 h). The cake was rinsed with CH<sub>2</sub>Cl<sub>2</sub> (500 mL, 1 mL/g). The combined filtrates were charged to a 30 L wash tank and diluted with 2 L of water (pH = 1.0). To the resulting biphasic mixture was added 2.5 L of 10% NaOH over a 15 min period (8 °C exotherm; pH = 12.0). The resulting vellow-colored aqueous layers were separated from the clear, colorless organic layer. The organic layer was concentrated in vacuo at 16 torr using a 37 °C water bath to provide a 93% yield (621g, 2.78 moles) as a light yellow oil. To ensure removal of residual CH<sub>2</sub>Cl<sub>2</sub>, the oil was dissolved in 2 L of absolute ethanol (AAPER, 200 proof) and concentrated in vacuo at 16 torr using a 57 °C water bath. The potency of the material was determined to be 99.2% (GC) and 99.0% (HPLC) and was taken on directly to the next step without further purification.

#### Preparation of t-butyl 2-aminobenzoate

30

5

10

15

20

25

Escat 10 catalyst (18.63 g, 3 wt%) was charged to the 10L autoclave followed by t-butyl nitrobenzoate (621g, 2.78 moles) in ethanol (7L). The vessel was sealed and purged three times with nitrogen (60 psig) and three times with hydrogen (60 psig). The vessel was then pressurized to 50 psig with hydrogen and allowed to run holding the exotherm at 40 °C through external cooling. The reaction was run until the hydrogen uptake stopped (45 minutes). The reaction was determined to be complete by both TLC and HPLC after 1 h and 10 min. The reaction was filtered through a 0.4 μ filter to remove the catalyst, and the catalyst cake was rinsed with ethanol (1.5 L). The product solution was then concentrated *in vacuo* at 16 torr using a 45 °C water bath to a volume of 1620 mL (3 mL/g) and taken on directly into the next step. An aliquot of the solution was concentrated and analyzed by both NMR and GC. The GC potency of the final product was 100%, and the NMR spectra were consistent with the structure of the title compound.

#### Preparation of t-butyl 2-amino-5-iodobenzoate

5

10

15

20

25

30

A 12 L round bottom flask, equipped with a thermocouple, nitrogen adapter and a 1 L addition funnel, was charged with a solution of *t*-butyl 2-aminobenzoate (537g, 2.78 moles; lot 36648-tjb-40) in ethanol (1620 ml, 3 ml/g). To this golden solution was added water (615.6 ml) resulting in a biphasic mixture. This mixture was cooled to between 15 and 20 °C with a cold-water bath. A 1.0 M solution of ICl in CH<sub>2</sub>Cl<sub>2</sub> (Aldrich lot #14127JO, 3.11 L, 3.11 moles, 1.12 equiv.) was charged in portions to the addition funnel and was added to the rapidly stirred mixture maintaining the temperature between 15 and 25 °C. The addition time was 2.25 hours and the temperature range observed was 16.5 to 20.4 °C. The resulting red brown mixture was stirred at room temperature for 1 hour at which time the GC assay showed the reaction was complete. The reaction was diluted with 920 mL of water and quenched with 456 mL of 38% aq. sodium bisulfite (Webb Chem lot #10464519) resulting in a slight exotherm to 24.0 °C. This mixture was stirred for 15 minutes before separating the

phases. The methylene chloride layer was combined with water (3.7L) and stirred for 15 minutes before separating the phases. A NaOH solution was prepared by diluting 10% NaOH (460ml) in water (2.3L). To the methylene chloride layer was added this dilute NaOH solution (2.1L). The pH of the basic phase was 6.56. The phases were separated and the methylene chloride layer was concentrated to a low volume *in vacuo* at 16 torr using a bath temp of 45 °C. Pyridine (4L) was added, and the resulting solution was concentrated to ca. 1.0 mL/g *in vacuo* at 16 torr using a 62 °C water bath. The low volume pyridine/product mixture was diluted with pyridine to the target volume of 3.1L (3.5 mL/g). A sample (10mL) was concentrated removing the pyridine on the rotovap and high vacuum to yield 3.12 g of an orange brown solid of 96% potecy by GC. GC assay of pyridine solution indicated that neither EtOH nor methylene chloride were present, so the solution was taken on directly into the next step.

## Preparaion of t-butyl 2-amino-5-cyanobenzoate

A 5 L Morton flask equipped with a mechanical stirrer (sturdy blade), thermocouple, and a reflux condenser was charged with 299g (3.34 moles, 1.2 equiv) of CuCN (Aldrich, 99%). To the slowly stirred CuCN was added a cool (10 °C) solution of *t*-butyl 2-amino-5-iodobenzoate (887g, 2.78 moles, 1.0 equiv) in pyridine (3.5 mL/g including the volume occupied by *t*-butyl 2-amino-5-iodobenzoate). The resulting orange suspension was heated to 115 °C over 45 min to produce a black solution. The solution was maintained at 115 °C for 14 h at which point GC indicated the reaction was complete. The solution was cooled to 90 °C and transferred by ½ inch ID Teflon cannula to a stirred suspension of solka floc (powdered cellulose, 460 g) in 14 L of methyl-*tert*-butyl ether (EM Science, 99.95%) at 2 °C, maintaining an internal temperature less than 13 °C. The resulting yellow-green suspension was filtered through a sintered glass frit (course frit, 16 torr vacuum) and the cake was rinsed with 4 L of MTBE (EM Science, 99.95%). The filtrate was washed (1 x 8 L H<sub>2</sub>O, 3 x 2 L

of 10% NH<sub>4</sub>OH in 23% NH<sub>4</sub>Cl), and the organics were concentrated in vacuo at 16 torr using a 50 °C water bath to a volume of 3 L (3.4 mL/g). The solution was split in half and crystallized in two portions. One half of the solution was charged to a 22 L flask containing heptanes (8L). The flask was set up for atmospheric distillation and heptanes (4L) was added to bring total volume of heptanes to 12 L. The mixture was distilled atmospherically to remove 4 L of distillate (pot temp of 98 °C; head temp of 96 °C). The pot was charged with 4 L of heptanes, and another 4 L of distillate was removed. A second 4 L charge of heptanes was made and 2.4 L of distillate was removed via atmospheric distillation; thus reducing the pot volume to 8.9 L (20mL/g). GC assay of the final distillate indicated the following volume percent ratios of pyridine and MTBE, respectively: 2.08% and 1.51 %. The heating mantle was removed, and the solution was cooled to induce crystallization (crystal formation was first noted at about 56 °C). The slurry was stirred at room temperature for 4 h, and the solids were isolated by vacuum filtration on a 3L frit. The cake was slurry washed with room temperature heptanes (2 x 500 ml) and dried on a nitrogen press to produce 224.2 g of an off-white solid (GC potency of 100%). Crystallization of the second half of the material produced another 241 g; thus a 70% yield from 2-nitrobenzoic acid was achieved.

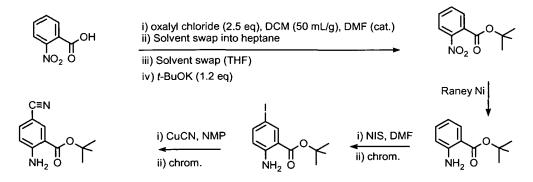
5

10

15

25

An alternative methodology for producing t-butyl 2-amino-5-cyanobenzoate is shown below.



5-Cyano-2-({3-[(3,3-dimethyl-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid

To a solution of 3-(chlorosulfonyl)benzoic acid (456 mg, 2.07 mmol, Aldrich) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was added DMF (15  $\mu$ L) followed by oxalyl chloride (270  $\mu$ L, 3.10 mmol). After stirring for 1.5 hours, the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in toluene (15 mL), and methyl 2-amino-5-cyanobenzoate (370 mg, 2.10 mmol) was added. The mixture was heated in a 105 °C oil bath for 2 hours, and the toluene was then removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (6 mL), and a mixture of 3,3dimethylindoline, describbed by Kucerovy et al. in Synth. Commun. 1992, 22(5), 729-733, (342 mg, 2.32 mmol) and triethylamine (600 µL, 4.31 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (6 mL) was added. This mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from CH<sub>2</sub>Cl<sub>2</sub> to 1% EtOAc in CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 728 mg of white solid as the methyl ester. The methyl ester was hydrolyzed according to method D yielding 292 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ 12.57 (s, 1 H), 8.80 (d, J = 8.7 Hz, 1 H), 8.41-8.44 (m, 2 H), 8.24 (d, J = 7.9 Hz, 1 H), 8.09-8.14 (m, 2 H), 7.83 (t, J = 7.9 Hz, 1 H), 7.55 (d, J = 8.1 Hz, 1 H), 7.24 (t, J = 7.7 Hz, 1 H), 7.18 (d, J = 7.7 Hz, 1 H), 7.02 (t, J = 7.5 Hz, 1 H), 3.73 (s, 2 H), 1.08 (s, 6 H).

5

10

15

20

# $5-Bromo-2-[(4-\{[(4-chlorophenyl)(methyl)amino]sulfonyl\}benzoyl)amino]benzoicacid$

,0055 051

5

10

15

20

25

Dimethyl formamide (15 µL) and oxalyl chloride (1.5 mL, 17 mmol) were added sequentially to a mixture of 4-{[(4-chlorophenyl)(methyl)amino]sulfonyl} benzoic acid (2.82 g, 8.66 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (60 mL). The resulting solution was stirred for 3 hours after which the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (50 mL), and methyl 2-amino-5-bromobenzoate (1.83 g, 7.95 mmol, Avocado) in pyridine (15 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 150 mL of CH<sub>2</sub>Cl<sub>2</sub>. The resulting solution was washed with 2 X 100 mL of 1M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 s siliga cartridge with CH<sub>2</sub>Cl<sub>2</sub> as the eluent. Product was isolated as 3.73 g (87%) of a white solid as the methyl ester. The methyl ester was hydrolyzed according to method B. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.12 (s, 1 H), 8.56 (d, J = 8.7 Hz, 1 H), 8.10-8.14 (m, 3 H), 7.88 (dd, J = 8.7, 2.5 Hz, 1 H), 7.74 (d, J = 8.1 Hz, 2 H), 7.43 (d, J = 8.7 Hz, 2 H), 7.18 (d, J = 8.7 Hz, 2 H), 3.18 (s, 3 H).

### Preparation of 4-Bromo-3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoic acid

A solution of indoline (4.1 g, 34 mmol, Aldrich) and triethylamine (7.0 mL, 50 mmol) in methanol (20 mL) was added by cannula to solid 4-bromo-3- (chlorosulfonyl)benzoic acid (7.30 g, 24.4 mmol) with stirring in an ice bath. The mixture was allowed to warm slowly to room temperature and stirred overnight. It was added to a separatory funnel with 80 mL of aqueous 1 M NaOH, and this solution was washed with 2 X 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The aqueous layer was then acidified with concentrated HCl. The precipitate was washed with water followed by heptane and then recrystallized from toluene/ethanol. The crystals were washed with toluene followed by heptane and then dried at 100 °C under vacuum yielding 2.75 g of white solid. A second crop of 1.39 g of tan solid was also collected.

### Preparation of 4-Cyano-3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoic acid

A mixture of copper (I) cyanide (755 mg, 8.43 mmol) and 4-bromo-3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoic acid (2.05 g, 5.36 mmol) in NMP (15 mL) was heated to 160 °C under nitrogen for 1 hour. The mixture was added to a flask with 150 mL of EtOAc and 100 mL of water and stirred for 30 minutes. It was then filtered through a plug of celite. The phases were separated, and the water was extracted with an additional 2 X 100 mL of EtOAc. The combined EtOAc was washed with 3 X 100 mL of water and dried over MgSO<sub>4</sub>. The solvent was removed, and the brown residue was recrystallized from hot ethanol. The crystals were washed with methanol followed by heptane and then dried at 100 °C under vacuum. Yield was 1.25 g of tan solid.

15

20

5-Bromo-2-{[4-cyano-3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid

To 4-cyano-3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoic acid (1.22 g, 3.72 mmol) in  $CH_2Cl_2$  (30 mL) was added DMF (20  $\mu$ L) and oxalyl chloride (650  $\mu$ L, 7.45 mmol). The mixture was stirred for 2.3 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved as best as possible in  $CH_2Cl_2$  (30 mL), and methyl 2-amino-5-bromobenzoate (762 mg, 3.31 mmol, Avocado) in pyridine (15 mL) was added. The mixture was stirred overnight and then

added to a separatory funnel with 100 mL of  $CH_2Cl_2$ . This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The  $CH_2Cl_2$  was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with  $CH_2Cl_2$  as eluent. Yield was 1.31 g of yellow solid. The methyl ester was hydrolyzed according to Method D to yield 615 mg of yellow solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.24 (s, 1 H), 8.57 (s, 1 H), 8.51 (d, J = 8.7 Hz, 1 H), 8.37 (d, J = 7.6 Hz, 1 H), 8.32 (d, J = 7.6 Hz, 1 H), 8.14 (d, J = 2.5 Hz, 1 H), 7.88 (dd, J = 8.9, 2.3 Hz, 1 H), 7.43 (d, J = 8.1 Hz, 1 H), 7.16-7.24 (m, 2 H), 7.01 (t, J = 7.6 Hz, 1 H), 4.20 (t, J = 8.4 Hz, 2 H), 3.05 (t, J = 8.4 Hz, 2 H).

10

15

20

25

30

5

#### Preparation of Methyl 3-(chlorosulfonyl)-2-methylbenzoate

A flask was charged with methyl 2-methyl-3-nitrobenzoate (Aldrich, 5.0 g, 25.6 mmol) and tin (II) chloride dihydrate (28.9 g, 128 mmol, 5.0 eq). The solids were suspended in EtOAc (80 mL), and upon heating to reflux under N<sub>2</sub> the solids completely dissolved. After two hours the cooled reaction was poured into 350 mL EtOAc and then washed 4x with 1.0M NaOH, 1x with water and 1x with brine (350 mL each). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent evaporated. The resultant crude oil (2.9 g) was suspended in 60 mL of a 2:1 solution of concentrated HCl and glacial acetic acid. The reaction was cooled to -10 °C and a solution of sodium nitrite (1.33g, 19.34 mmol) in 3.0 mL water was added drop wise over stirring at a rate that maintained the internal reaction temperature below -5 °C. The reaction became an orange solution as the SM slowly dissolved. In a separate flask, copper (I) chloride (435 mg, 25 mol%) was suspended in 30 mL of a saturated (30% w/w) solution of sulfur dioxide gas in glacial acetic acid. The mixture was cooled on an ice bath over stirring, and after 2.5 hours the diazonium solution was added portion wise to the copper mixture over 15 minutes. The addition evolved gas and produced a lime green solution, which came to RT and was stirred overnight. The reaction was poured into ice water (200 mL) to afford an oil at the bottom of a pale blue solution. The solution was extracted 2x with CH<sub>2</sub>Cl<sub>2</sub> (150 mL ea) and the organic phase was washed 2x with saturated NaHCO<sub>3</sub> and brine (250 mL ea). The

10

15

20

25

golden organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent evaporated. The crude residue was purified on a Biotage Flash 40M+ (100g) silica cartridge using a gradient of 20% heptane in CH<sub>2</sub>Cl<sub>2</sub> to 100% CH<sub>2</sub>Cl<sub>2</sub>. The combined fractions were evaporated and the product was dried under high vacuum at RT to afford 2.2 g of pale pink solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.96 (dd, J = 7.7, 1.5 Hz, 1 H), 7.59 (dd, J = 7.7, 1.5 Hz, 1 H), 7.23 (t, J = 7.7 Hz, 1 H), 3.82 (s, 3 H), 2.56 (s, 3 H).

### Preparation of 3-{[(4-chlorophenyl)(methyl)amino|sulfonyl}-2-methylbenzoate

Methyl 3-(chlorosulfonyl)-2-methylbenzoate, (494 mg, 1.99 mmol) was taken up in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and treated with 4-chloro-N-methylaniline (1.01 mL, 8.35 mmol, Aldrich) in dry pyridine (15 mL). The bright yellow solution was heated to 75 °C. After one hour HPLC indicated the reaction was complete and the mixture was poured into EtOAc (125 mL). The organic phase was washed 3x with 1.0M HCl, 1x with saturated NaHCO<sub>3</sub> and 1x with brine (100 mL each). After drying over Na<sub>2</sub>SO<sub>4</sub> the solution was filtered and the solvent was evaporated to afford an amber oil, which was purified on a Biotage Flash 40M+ (100g) silica cartridge using a linear gradient of 35% to 5% heptane in CH<sub>2</sub>Cl<sub>2</sub>. The solvent was evaporated from the product fractions and the product was dried under high vacuum at RT to afford 637 mg (90%) of a colorless oil. 508 mg, 1.44 mmol of the oil was dissolved in MeOH (15 mL) and treated with 1.0M LiOH (3.0 mL, 3.0 mmol). After stirring at 40 °C for 1 hour and then overnight at RT, the reaction was complete by HPLC and OAMS showed the correct m/z for product. The reaction was poured into 1.0M HCl (100 mL), and the white precipitate was extracted into EtOAc (150 mL). The organic layer was then 1x with 1.0M HCl and 1x with brine (125 mL each). The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated to dryness. The resultant product was dried under vacuum at 100 °C overnight to afford 461 mg (94%) of off-white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  13.41 (br s, 1 H), 7.94 (d, J = 3.3 Hz, 1 H), 7.92 (d, J = 3.1

5

10

15

20

25

Hz, 1H), 7.49 (t, J = 7.9 Hz, 1 H), 7.39-7.47 (m, 2 H), 7.22-7.31 (m, 2 H), 3.21 (s, 3 H), 2.45 (s, 3 H).

# 5-Bromo-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-2-methylbenzovl)amino]benzoic acid

3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-2-methylbenzoate (404 mg, 1.19 mmol) was suspended in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and DMF (10 µL) under N<sub>2</sub>. The solution was treated with oxalyl chloride (Aldrich, 0.192 mL, 2.2 mmol) and stirred while gas evolved. After one hour the excess solvent and oxalyl chloride were evaporated and the resultant residue was taken up in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL). Methyl-2-amino-5bromobenzoate (Aldrich, 230 mg, 1.0 mmol) was added as a solution in pyridine (3 mL) and the amber solution stirred at RT. After 2 hours HPLC indicated the reaction was complete. The mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> (100 mL) and washed 2x with 1.0M HCl followed by brine (100 mL each). The organic layer was evaporated and purified on a Biotage Flash 25M+ (40 g) silica cartridge using CH<sub>2</sub>Cl<sub>2</sub>. The combined fractions were evaporated and the product was dried under vacuum at 100 °C to afford 535mg (97%) of a glass-like solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  10.88 (s, 1 H), 8.05 (d, J = 8.9 Hz, 1 H), 7.99 (d, J = 2.3 Hz, 1 H), 7.93 (D, J = 7.5 Hz, 1 H), 7.86 (dd, J = 8.8, 2.4 Hz, 1 H), 7.80 (d, J = 7.3 Hz, 1 H), 7.57 (t, J = 7.9 Hz, 1 H), 7.45 (d, J = 7.8 Hz, 1 Hz, 1 Hz), 7.80 (d, J = 7.8 Hz),J = 8.7 Hz, 2 H), 7.29 (d, J = 8.7 Hz, 2 H), 3.83 (s, 3 H), 3.24 (s, 3 H), 2.39 (s, 3 H). 322 mg of the methyl ester solid was dissolved in hot dioxane (10 mL), and after cooling was treated with 1.0M LiOH (1.0 mL, 1.0 mmol). After stirring overnight at RT the reaction was complete by HPLC and OAMS showed correct m/z for the product. The solvent was evaporated and the residue was poured into 1.0M HCl (100 mL) to afford a white precipitate. The product was extracted into EtOAc (125 mL) and washed 3x with 1.0M HCl, and 1x with brine (100 mL each). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated to dryness. The crude product was recrystallized from hot MeOH/EtOH. The resultant product was dried at 100 °C under

vacuum to afford 213 mg (68%) of white crystals. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  11.35 (s, 1 H), 8.39 (d, J = 8.9 Hz, 1 H), 8.07 (d, J = 2.5 Hz, 1 H), 7.92 (dd, J = 8.1, 1.0 Hz, 1 H), 7.81-7.89 (m, 2 H), 7.56 (t, J = 7.8 Hz, 1 H), 7.41-7.48 (m, 2 H), 7.24-7.34 (m, 2 H), 3.23 (s, 3 H), 2.39 (s, 3 H).

5

10

15

20

25

# Preparation of 3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]-2-methylbenzoic acid

Methyl 3-(chlorosulfonyl)-2-methylbenzoate, (673 mg, 2.71 mmol) was taken up in dry CH<sub>2</sub>Cl<sub>2</sub> (5 mL) and dry pyridine (5 mL). The golden solution was cooled to -10 °C and treated with 5-chloroindoline (1.01 mL, 8.35 mmol, Aldrich) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 mL) to afford an intensely red-orange solution. A precipitate formed as the reaction warmed to RT. After one hour HPLC indicated the reaction was complete and the mixture was diluted to 150 mL with CH<sub>2</sub>Cl<sub>2</sub>. The organic phase was washed 1x with 1.0M HCl, 1x with 1.0M NaOH, 1x with 1.0M HCl and 1x with brine (125 mL each). After drying over Na<sub>2</sub>SO<sub>4</sub> the solution was filtered and the solvent was evaporated. The resultant product was dried under high vacuum at RT to afford 900 mg (90%) of a peach colored oil. 780mg (2.13 mmol) of the oil was dissolved in MeOH (15 mL) and treated with 1.0M LiOH (5.0 mL, 5.0 mmol). After stirring at 40 °C for 1 hour and then overnight at RT, the reaction was complete by HPLC and OAMS showed the correct m/z for product. The reaction was poured into 1.0M HCl (125 mL), and the vellowish precipitate was extracted into EtOAc (150 mL). The organic layer was then 2x with 1.0M HCl, 1x with water and 1x with brine (125 mL each). The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated to dryness. The resultant product was dried under vacuum at 100 °C overnight to afford 711 mg (95%) of pinkish-orange solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  13.46 (br s, 1 H), 7.98 (d, J = 8.1 Hz, 1 H), 7.93 (d. J = 7.7 Hz, 1 H), 7.50 (t, J = 7.9 Hz, 1 H), 7.34 (d, J = 1.7 Hz, 1 H), 7.19 (dd, J = 8.5, 2.1 Hz, 1 H), 7.09 (d, J = 8.5 Hz, 1 H), 4.05 (t, J = 8.5 Hz, 2 H), 3.12 (t, J = 8.5 Hz, 3 8.5 Hz, 2 H), 2.66 (s, 3 H).

# 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]-2-methylbenzoyl}amino)-5-cvanobenzoic acid

5

10

15

20

25

3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]-2-methylbenzoic acid (553 mg, 01.57 mmol) was suspended in dry  $CH_2Cl_2$  (15 mL) and DMF (10  $\mu$ L) under  $N_2$ . The solution was treated with oxalyl chloride (0.274 mL, 3.14 mmol, Aldrich) and stirred while gas evolved. The reaction became homogenous and after one hour the excess solvent and oxalyl chloride was evaporated and the resultant residue was taken up in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL). Methyl-2-amino-5-cyanobenzoate (PHA-522499, 264 mg, 1.5 mmol) was added as a solution in pyridine (4 mL) and the amber solution stirred at RT. After 2.5 days HPLC indicated the reaction was nearly complete. After briefly boiling the reaction and cooling, the mixture was diluted to 150 mL with CH<sub>2</sub>Cl<sub>2</sub> and washed 2x with 1.0M HCl followed by brine (125 mL each). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated. The resultant crude product was purified on a Biotage Flash 25M+ (40 g) silica cartridge using a linear gradient of 0-2% EtOAc in CH<sub>2</sub>Cl<sub>2</sub>. The resultant product still contained a small amount of residual cyanoanthranilate. The combined fractions were evaporated and the product was purified a second time on a Biotage Flash 40M+ (100 g) silica cartridge using 100% CH<sub>2</sub>Cl<sub>2</sub>. The combined fractions were evaporated and dried under high vacuum at RT to afford 594mg (77%) of an off-white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  11.21 (s, 1 H), 8.29 (d, J = 1.9 Hz, 1 H), 8.26 (d, J = 8.7 Hz, 1 H), 8.11 (dd, J = 8.6, 2.0 Hz, 1 H), 7.99 (dd, J = 8.1, 1.0 Hz, 1 H), 7.84 (dd, J = 7.7, 1.0 Hz, 1 H), 7.60 (t, J = 7.9Hz, 1 H), 7.36 (d, J = 1.7 Hz, 1 H), 7.21 (dd, J = 8.7, 2.3 Hz, 1 H), 7.16 (d, J = 8.7 Hz, 1 H), 4.09 (t, J = 8.6 Hz, 2 H), 3.84 (s, 3 H), 3.15 (t, J = 8.4 Hz, 2 H), 2.61 (s, 3 H). The methyl ester was hydrolyzed as described above to afford 300 mg (77%) of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  11.73 (s, 1 H), 8.62 (d, J = 8.7 Hz, 1 H), 8.36 (d, J = 2.1 Hz, 1 H), 8.10 (dd, J = 8.7, 2.1 Hz, 1 H), 7.97 (d, J = 8.1 Hz, 1 H), 7.90 (d, J = 8.1 Hz, 1 Hz, 1 Hz), 7.90 (d, J = 8.1 Hz), 7.90 (d, J = 8.1 Hz, 1 Hz), 7.90 (d, J = 8.1 Hz), 7.90 (d, J = 8.1 Hz), 7.90 (d, J = 8.1 Hz), 7.90 (d,

J = 6.8 Hz, 1 H), 7.57 (t, J = 7.9 Hz, 1 H), 7.37 (s, 1 H), 7.20 (dd, J = 8.7, 2.1 Hz, 1 H), 7.16 (d, J = 8.5 Hz, 1 H), 4.07 (t, J = 8.6 Hz, 2 H), 3.15 (t, J = 8.5 Hz, 2 H), 2.62 (s, 3 H).

5 Preparation of 3-{[(4-Chlorophenyl)(methyl)amino]sulfonyl}-2-methoxybenzoic acid\_

10

15

20

25

Methyl 3-amino-2-methoxybenzoate (1.27 g, 6.72 mmol) was dissolved in 30 mL of a 2:1 solution of concentrated HCl and glacial acetic acid. The reaction was cooled to – 10 °C and a solution of sodium nitrite (696 mg, 10.1 mmol) in 3.0 mL water was added drop wise over stirring at a rate that maintained the internal reaction temperature below -5 °C. The reaction became a cloudy yellow-orange suspension. In a separate flask, copper (I) chloride (166 mg, 25 mol%) was suspended in 30 mL of a saturated (30% w/w) solution of sulfur dioxide gas in glacial acetic acid. The mixture was cooled on an ice bath over stirring, and after 30 minutes diazonium solution was added portion wise to the copper mixture over 15 minutes. The addition evolved gas and produced a dark green solution. The reaction was warmed to RT and was stirred for 3 hours with sulfur dioxide bubbling into the solution. The reaction was poured into ice water (200 mL) to afford a fine white precipitate in a pale blue solution. The solution was extracted 3x with EtOAc (150 mL ea) and the organic phase was neutralized by washing 3x with saturated NaHCO<sub>3</sub> (300 mL ea). The organic phase was then washed 2x with water and 1x with brine (250 mL ea). The golden organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent evaporated. The crude residue was dried under high vacuum to afford a dark red oil. The oil was taken up in pyridine (15 mL) and treated with 4-chloro-N-methylaniline (0.280 mL, 2.3 mmol, Aldrich). The amber solution was heated stirred at RT, and after one hour HPLC indicated the reaction was complete. The mixture was diluted to 150 mL with DCM and then washed 2x with 1.0M HCl, 1x with 1.0M NaOH and 1x with brine (125 mL each). The solvent was evaporated to afford an amber oil, which was

15

20

25

purified on a Biotage Flash 40M (90g) silica cartridge using a linear gradient of 0 to 0.75% EtOAc in CH<sub>2</sub>Cl<sub>2</sub>. The solvent was evaporated from the product fractions and the product was dried under high vacuum at RT to afford 614 mg (72%) of a straw colored oil as the methyl ester. The methyl ester was hydrolyzed as described above to afford 544 mg (97%) of peach colored solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  13.50 (s, 1 H), 7.99 (dd, J = 7.7, 1.9 Hz, 1 H), 7.80 (dd, J = 7.9, 1.7 Hz, 1 H), 7.36-7.42 (m, 2 H), 7.30 (t, J = 7.9 Hz, 1 H), 7.19-7.26 (m, 2 H), 3.83 (s, 3 H), 3.32 (s, 3 H).

# 5-Bromo-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-2-methoxybenzoyl)amino]benzoic acid

3-{[(4-Chlorophenyl)(methyl)amino]sulfonyl}-2-methoxybenzoic acid (PHA-733277, 474 mg, 01.33 mmol) was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and DMF (25 μL) under N<sub>2</sub>. The solution was treated with oxalyl chloride (0.232 mL, 2.66 mmol, Aldrich) and stirred while gas evolved. The reaction was stirred at RT and after one hour the excess solvent and oxalyl chloride was evaporated and the resultant residue was taken up in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL). Methyl-2-amino-5-bromobenzoate (288 mg, 1.25 mmol, Avocado) was added as a solution in pyridine (3 mL) and the amber solution stirred at RT. After 90 minutes HPLC indicated the reaction was complete. The mixture was diluted to 150 mL with CH<sub>2</sub>Cl<sub>2</sub> and washed 2x with 1.0M HCl followed by brine (100 mL each). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated. The resultant crude product was purified on a Biotage Flash 40M (90 g) silica cartridge using CH<sub>2</sub>Cl<sub>2</sub>. The combined fractions were evaporated and dried under vacuum at 100 °C to afford 530mg (72%) of an off-white solid as the methyl ester. <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{DMSO-}d_6) \delta 11.52 \text{ (s, 1 H)}, 8.48 \text{ (d, } J = 8.7 \text{ Hz, 1 H)}, 8.10 \text{ (d, } J = 2.5 \text{ Hz,}$ 1 H), 7.98 (dd, J = 7.8, 1.8 Hz, 1 H), 7.91 (dd, J = 8.9, 2.5 Hz, 1 H), 7.81 (dd, J = 7.9, 1.7 Hz, 1 H), 7.32-7.43 (m, 5 H), 3.89 (s, 3 H), 3.82 (s, 3 H), 3.40 (s, 3 H). The corresponding methyl ester was hydrolyzed as described above to afford a white solid. TOO CCOON

10

15

20

25

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ) δ 12.02 (s, 1 H), 8.70 (d, J = 9.1 Hz, 1 H), 8.14 (d, J = 2.5 Hz, 1 H), 8.01 (dd, J = 7.7, 1.2 Hz, 1 H), 7.90 (dd, J = 9.0, 2.4 Hz, 1 H), 7.76 (dd, J = 7.9, 1.5 Hz, 1 H), 7.11-7.44 (m, 5 H), 3.81 (s, 3 H), 3.39 (s, 3 H).

### 5 5-bromo-2-[(4-{[methyl(pyridin-2-yl)amino]sulfonyl}benzoyl)amino]benzoic acid

4-{[methyl(pyridin-2-yl)amino]sulfonyl}benzoic acid (292 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed in vacuo to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography (20% EtOAc in hexane) to provide 317 mg of the desired methyl ester (63%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (281 mg, 91%, 57% overall) was obtained as a tan solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 3.72 (s, 3H), 7.28 (dd, 1H), 7.56 (d, 1H), 7.81-7.91 (m, 4H), 8.07 (d, 2H), 8.12 (d, 1H), 8.32 (dd, 1H), 8.54 (d, 1H), 12.10 (s, 1H). C NMR (100 MHz, DMSO) 36.10, 101.83, 115.38, 120.21, 120.30, 122.15, 122.90, 128.33, 128.42, 133.62, 136.95, 138.77, 139.91, 140.30, 148.52, 153.13, 163.81, 168.81. MS (FAB) m/z (rel. intensity) 490 (MH<sup>+</sup>, 30), 492 (32), 490 (30), 414 (28), 413 (83), 109 (31), 107 (36), 95 (25), 91 (99), 57 (73), 55 (28). HRMS (FAB) calcd for C<sub>20</sub>H<sub>16</sub>BRN<sub>3</sub>O<sub>5</sub>S +H<sub>1</sub> 490.0073, found 490.0067.

3-(1H-indol-1-ylsulfonyl)benzoic acid (301 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed in vacuo to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 287 mg of the desired methyl ester (56%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (53 mg, 11%, 6% overall) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 6.90 (d, 1H), 7.27 (t, 1H), 7.37 (t, 1H), 7.62 (d, 1H), 7.82 (t, 1H), 7.87-7.89 (m, 2H), 8.00 (d, 1H), 8.05 (d, 1H), 8.19-8.25 (m, 3H), 8.47 (s, 1H), 11.35 (s, 1H).

#### 5-bromo-2-{[3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid

20

25

5

10

15

3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoic acid (305 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol)

was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel

5 chromatography (10% EtOAc in hexane) to provide 381 mg of the desired methyl ester (74%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (344 mg, 93%, 68% overall) was obtained as a white solid after recrystalization from MeOH. H NMR

10 (400 MHz, DMSO) 2.94 (t, 2H), 4.00 (t, 2H), 6.99 (t, 1H), 7.15-7.23 (m, 2H), 7.52 (d, 1H), 7.80 (t, 1H), 7.89 (dd, 1H), 8.05-8.07 (m, 2H), 8.20 (d, 1H), 8.28 (d, 1H), 8.35 (s, 1H), 11.40 (s, 1H).

### 5-bromo-2-{[4-(pyrrolidin-1-ylsulfonyl)benzoyl]amino}benzoic acid

15

4-(pyrrolidin-1-ylsulfonyl)benzoic acid (255 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL).
Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 331 mg of the desired methyl ester (71%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (308 mg, 96%, 68% overall) was obtained as a pale yellow solid

after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.67 (m, 4H), 3.19 (m, 4H), 7.88 (dd, 1H), 8.02 (d, 2H), 8.12-8.16 (m, 3H), 8.58 (d, 1H), 12.10 (s, 1H).

#### 5-cyano-2-{[4-(pyrrolidin-1-ylsulfonyl)benzoyl]amino}benzoic acid

5

10

15

20

25

4-(pyrrolidin-1-ylsulfonyl)benzoic acid (255 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-cyanobenzoate (176 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 293 mg of the desired methyl ester (71%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (262 mg, 92%, 65% overall) was obtained as a pale yellow solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.67 (m, 4H), 3.20 (m, 4H), 8.04 (d, 2H), 8.11-8.18 (m, 3H), 8.42 (d, 1H), 8.80 (d, 1H), 12.25 (s, 1H).

#### 5-bromo-2-{[3-(pyrrolidin-1-ylsulfonyl)benzoyl]amino}benzoic acid

3-(pyrrolidin-1-ylsulfonyl)benzoic acid (255 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 0.5 mL, 5.7 mmol). A catalytic amount of

DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 333 mg of the desired methyl ester (71%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (309 mg, 96%, 68% overall) was obtained as a pale yellow solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.67 (m, 4H), 3.20 (m, 4H), 7.85-7.89 (m, 2H), 8.08 (d, 1H), 8.13 (d, 1H), 8.25 (d, 1H), 8.32 (s, 1H), 8.60 (d, 1H), 12.20 (s, 1H).

15

20

25

10

5

#### 5-cyano-2-({3-[(2-methylpyrrolidin-1-yl)sulfonyl]benzoyl}amino)benzoic acid

3-[(2-methylpyrrolidin-1-yl)sulfonyl]benzoic acid (269 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-cyanobenzoate (176 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 350 mg of the desired methyl ester (82%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*.

5

10

15

20

The title compound (308 mg, 91%, 75% overall) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.25 (d, 3H), 1.41-1.47 (m, 2H), 1.59-1.67 (m, 1H), 1.77-1.83 (m, 1H), 3.12-3.18 (m, 1H), 3.36-3.42 (m, 1H), 3.69 (m, 1H), 7.88 (t, 1H), 8.12 (d, 1H), 8.13 (d, 1H), 8.25 (d, 1H), 8.34 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.55 (s, 1H).

#### 5-cyano-2-({3-[(2,5-dimethylpyrrolidin-1-yl)sulfonyl]benzoyl}amino)benzoic acid

3-[(2,5-dimethylpyrrolidin-1-yl)sulfonyl]benzoic acid (283 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed in vacuo to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-cyanobenzoate (176 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 293 mg of the desired methyl ester (66%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (273 mg, 97%, 64% overall) was obtained as a tan solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.29 (d, 6H), 1.49 (m, 4H), 3.67 (m, 2H), 7.88 (t, 1H), 8.12 (d, 1H), 8.13 (d, 1H), 8.25 (d, 1H), 8.34 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.55 (s, 1H).

25

**5-cyano-2-{[3-(pyrrolidin-1-ylsulfonyl)benzoyl]amino}benzoic acid** was produced from methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate. H NMR (300 MHz, DMSO) 1.67 (m, 4H), 3.20 (m, 4H), 7.88 (t, 1H), 8.09-8.14 (m, 2H), 8.26 (d, 1H), 8.33 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.56 (s, 1H)

### 5-chloro-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid

3-(morpholin-4-ylsulfonyl)benzoic acid (271 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-chlorobenzoate (185 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (20% EtOAc in hexane) to provide 382 mg of the desired methyl ester (87%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title

The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (351 mg, 95%, 83% overall) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 2.93 (m, 4H), 3.65 (m, 4H), 7.77 (dd, 1H), 7.91 (t, 1H), 7.99-8.02 (m, 2H), 8.25-8.29 (m, 2H), 8.65 (d, 1H), 12.17 (s, 1H).

20

5

10

15

5-bromo-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino} benzoic acid and 2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}-5-nitrobenzoic acid were produced in a similar fashion utilizing appropriate starting materials.

25 5-fluoro-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid

3-(morpholin-4-ylsulfonyl)benzoic acid (271 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed in vacuo to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-fluorobenzoate (170 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography (20% EtOAc in hexane) to provide 367 mg of the desired methyl ester (87%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (328 mg, 92%, 80% overall) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 2.93 (m, 4H), 3.65 (m, 4H), 7.58 (m, 1H), 7.77 (dd, 1H), 7.90 (t, 1H), 8.00 (d, 1H), 8.26-8.29 (m, 2H), 8.60 (dd, 1H), 12.02 (s, 1H).

### 5-cyano-2-{[3-(piperidin-1-ylsulfonyl)benzoyl]amino}benzoic acid

20

25

5

10

15

3-(piperidin-1-ylsulfonyl)benzoic acid (269 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-cyanobenzoate (176 mg, 1.0 mmol) was added followed by pyridine (1

10

15

20

25

mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (10% EtOAc in hexane) to provide 307 mg of the desired methyl ester (72%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (279 mg, 94%) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 1.37 (m, 2H), 1.56 (m, 4H), 2.95 (m, 4H), 7.90 (t, 1H), 8.02 (d, 1H), 8.13 (dd, 1H), 8.27 (m, 2H), 8.42 (d, 1H), 8.83 (1H), 12.55 (s, 1H).

## 5-cyano-2-{[3-(1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid

Indole (150 mg, 1.25 mmol) was dissolved in 15 ml of THF. NaH (100 mg, 60% disp. in oil, 2.5 mmol) was added and resulting suspension stirrted for 1 h. Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) was then added and the reaction stirred at room temperature of 12 hr. The mixture was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography, providing 252 mg (55%) of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (24 mg, 10%) was obtained as a tan solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 6.89 (d, 1H), 7.28 (t, 1H), 7.37 (t, 1H), 7.61 (d, 1H), 7.81-7.86 (m, 2H), 8.01 (d, 1H), 8.11 (dd, 1H), 8.24 (t, 2H), 8.42 (d, 1H), 8.52 (t, 1H), 8.75 (d, 1H).

#### 5-cyano-2-({3-[(5-methoxy-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid

5-Methoxyindole (190 mg, 1.25 mmol) was dissolved in 15 ml of THF. NaH (100 mg, 60% disp. in oil, 2.5 mmol) was added and resulting suspension stirrted for 1 h. Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) was then added and the reaction stirred at room temperature of 12 hr. The mixture was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography, providing 236 mg (48%) of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (31 mg, 13%) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 3.73 (s, 3H), 6.81 (d, 1H), 6.97 (dd, 1H), 7.11 (d, 1H), 7.79 (d, 1H), 7.82 (t, 1H), 7.89 (d, 1H), 8.11 (dd, 1H), 8.21 (t, 1H), 8.42 (d, 1H), 8.48 (t, 1H), 8.76 (d, 1H).

15

10

5

**5-cyano-2-({3-[(7-methoxy-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid** was produced using 7-Methoxyindole. H NMR (400 MHz, DMSO) 3.84 (s, 3H), 6.79 (d, 1H), 6.83 (d, 1H), 7.29 (t, 1H), 7.61 (d, 1H), 7.75 (d, 1H), 7.80 (m, 2H), 8.17 (d, 1H), 8.28 (d, 1H), 8.32 (d, 1H), 8.56 (t, 1H), 8.73 (d, 1H).

20

**5-cyano-2-({3-[(6-methoxy-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid** was produced using 6-Methoxyindole. H NMR (300 MHz, DMSO) 3.85 (s, 3H), 6.78 (d, 1H), 6.89 (dd, 1H), 7.47-7.49 (m, 2H), 7.71 (d, 1H), 7.79-7.85 (m, 2H), 8.20 (d, 1H), 8.29 (d, 1H), 8.34 (d, 1H), 8.59 (t, 1H), 8.75 (d, 1H).

25

2-({3-[(5-chloro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-cyanobenzoic acid

5-Chloroindole (190 mg, 1.25 mmol) was dissolved in 15 ml of THF. NaH (100 mg, 60% disp. in oil, 2.5 mmol) was added and resulting suspension stirrted for 1 h. Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) was then added and the reaction stirred at room temperature of 12 hr. The mixture was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography, providing 311 mg (63%) of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (27 mg, 9%) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 6.88 (d, 1H), 7.42 (dd, 1H), 7.71 (d, 1H), 7.85 (t, 1H), 7.94 (d, 1H), 8.02 (d, 1H), 8.12 (dd, 1H), 8.25 (m, 2H), 8.43 (d, 1H), 8.52 (t, 1H), 8.76 (d, 1H).

15

10

5

**5-cyano-2-({3-[(5-fluoro-1H-indol-1-yl)sulfonyl]benzoyl}amino)benzoic acid** was produced utilizing 5-Fluoroindole. H NMR (400 MHz, DMSO) 6.89 (d, 1H), 7.23 (dt, 1H), 7.44 (dd, 1H), 7.85 (t, 1H), 7.96 (d, 1H), 8.00 (dd, 1H), 8.13 (dd, 1H), 8.22 (d, 1H), 8.27 (d, 1H), 8.37 (d, 1H), 8.51 (m, 2H).

20

# 5-cyano-2-{[3-({methyl[(1R)-1-phenylethyl]amino}sulfonyl)benzoyl]amino}benzoic acid

10

15

20

Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) was dissolved in 15 mL of CHCl<sub>3</sub>. N-methyl-N-[(1R)-1-phenylethyl]amine (270 mg, 2.0 mmol) and Et<sub>3</sub>N (1 mL) were then added and the reaction stirred at room temperature for 12 hr. The mixture was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography, providing 400 mg (84%) of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (300 mg, 77%) was obtained as a white solid after recrystalization from MeOH. H NMR (300 MHz, DMSO) 1.23 (d, 3H), 2.61 (s, 3H), 5.22 (q, 1H), 7.26-7.35 (m, 5H), 7.87 (t, 1H), 8.12 (d, 1H), 8.13 (d, 1H), 8.25 (d, 1H), 8.36 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.55 (s, 1H).

#### 5-cyano-2-{[3-({methyl[(1S)-1-

phenylethyl]amino}sulfonyl)benzoyl]amino}benzoic acid was produced from N-methyl-N-[(1S)-1-phenylethyl]amine. H NMR (300 MHz, DMSO) 1.23 (d, 3H), 2.61 (s, 3H), 5.22 (q, 1H), 7.26-7.35 (m, 5H), 7.87 (t, 1H), 8.12 (d, 1H), 8.13 (d, 1H), 8.25 (d, 1H), 8.36 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.55 (s, 1H).

#### Scheme 1.4

25

 $2\hbox{-}[(4\hbox{-}\{[(2\hbox{-}aminophenyl)(methyl)amino}]\hbox{sulfonyl}]\hbox{benzoyl})\hbox{amino}]\hbox{-}5\hbox{-}bromobenzoic acid }$ 

10

15

20

(s, 1H).

4-{[{2-[(tert-butoxycarbonyl)amino]phenyl}(methyl)amino]sulfonyl}benzoic (406 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography to provide 346 mg of the desired methyl ester (56%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification. The resulting solid was dried in the air then dissolved in CH<sub>2</sub>Cl<sub>2</sub>/TFA and stirred for 10 additional hours. The solvent was removed *in vacuo* and the remaining solid was recrystalized from MeOH to give the title compound (163 mg, 58%) as a white solid. H NMR (400 MHz, DMSO) 3.12 (s, 3H), 6.36-6.43 (m, 2H), 6.78 (d, 1H), 6.99-7.04

2-[(3-{[(2-aminophenyl)(methyl)amino]sulfonyl}benzoyl)amino]-5-cyanobenzoic acid

(m, 1H), 7.83-7.93 (m, 3H), 8.13-8.16 (m, 2H), 8.28-8.29 (m, 1H), 8.60 (t, 1H), 12.21

3-{[{2-[(tert-butoxycarbonyl)amino]phenyl}(methyl)amino]sulfonyl}benzoic (406 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7

10

15

20

25

30

mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-cyanobenzoate (176 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography to provide 344 mg of the desired methyl ester (61%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification. The resulting solid was dried in the air then dissolved in CH<sub>2</sub>Cl<sub>2</sub>/TFA and stirred for 10 additional hours. The solvent was removed *in vacuo* and the remaining solid was recrystalized from MeOH to give the title compound (34 mg, 12%) as a white solid. H NMR (400 MHz, DMSO) 3.11 (s, 1H), 6.37 (m, 2H), 6.76 (d, 1H), 7.00 (m, 1H), 7.84-7.93 (m, 2H), 8.31 (dd, 1H), 8.31 (m, 2H), 8.42 (d, 1H), 8.83 (d, 1H), 12.55 9s, 1H).

#### Preparation of Methyl 2-amino-5-formylbenzoate

To a solution of methyl anthranilate (7.75 g, 51.3 mmol, Aldrich) in DMF (50 mL) was added NIS (11.5 g, 51.3 mmol, Aldrich). The solution was stirred for 63 hours before being added to a separatory funnel with 200 mL of MTBE and washed with 5 X 200 mL of water. The organics were dried over MgSO<sub>4</sub> and evaporated yielding 13.8 g of tan solid as methyl 2-amino-5-iodobenzoate. A mixture of methyl 2-amino-5-iodobenzoate (3.13 g, 11.3 mmol) and tetrakis(triphenylphosphine)palladium(0) (282 mg, 0.244 mmol, Strem) was placed under 1 atm of CO. THF (20 mL) was added, and the solution was heated to 60 °C. Tri-*n*-butyltin hydride (3.7 mL, 12.7 mmol, Aldrich) was added dropwise with rapid stirring over 4 hours. The dark orange solution was heated a further 45 minutes and then added to a separatory funnel with 150 mL of EtOAc. This solution was washed with 2 X 150 mL of saturated aqueous NaHCO<sub>3</sub> followed by 100 mL of brine. It was dried over MgSO<sub>4</sub> and evaporated leaving a brown oil that was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from CH<sub>2</sub>Cl<sub>2</sub> to 5% EtOAc in CH<sub>2</sub>Cl<sub>2</sub> as eluent. This chromatography failed to remove all of the tin, so the product was re-

chromatographed using a Biotage Flash 40 M silica cartridge with 5% EtOAc in CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 863 mg of white solid.

#### 5-Formyl-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid

5

10

15

20

25

To 3-(morpholin-4-ylsulfonyl)benzoic acid (1.12 g, 4.13 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (60 mL) was added DMF (20 µL) and oxalyl chloride (450 µL, 5.16 mmol). The mixture was stirred for 3.75 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL), and methyl 2amino-5-formylbenzoate (637 mg, 3.56 mmol) in pyridine (8 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 5% EtOAc in CH2Cl2 to 10% EtOAc in CH2Cl2 as eluent. Yield was 636 mg of yellow solid as the methyl ester. To a mixture of the corresponding methyl ester (318 mg, 0.735 mmol) in dioxane (15 mL) was added 1 M aqueous sodium hydroxide (1.5 mL). The mixture was stirred at room temperature for 2 hours. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of EtOAc. The EtOAc was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. It was then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol. The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 64 mg of tan solid. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ 12.65 (s. 1 H), 10.00 (s. 1 H), 8.89 (d, J = 8.7 Hz, 1 H), 8.60 (d, J = 2.1 Hz, 1 H), 8.29-8.33 (m, 2 H), 8.20 (dd, J = 8.7, 2.1 Hz, 1 H), 8.03 (d, J = 8.1 Hz, 1 H), 7.93 (t, J = 7.8 Hz, 1 H), 3.63-3.68 (m, 4 H), 2.92-2.97 (m, 4 H).

#### 5-bromo-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid

3-(morpholin-4-ylsulfonyl)benzoic acid (271 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed in vacuo to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-bromobenzoate (230 mg, 1.0 mmol) was added followed by pyridine (1 mL). The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography (20% EtOAc in hexane) to provide 367 mg of the desired methyl ester (76%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (328 mg, 92%, 70% overall) was obtained as a white solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 2.93 (m, 4H), 3.65 (m, 4H), 7.88 (dd, 1H), 7.90 (d, 1H), 8.00 (d, 1H), 8.13 (d, 1H), 8.25-8.29 (m, 2H), 8.59 (d, 1H), 12.21 (s, 1H).

#### 2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}-5-nitrobenzoic acid

20

25

5

10

15

3-(morpholin-4-ylsulfonyl)benzoic acid (271 mg, 1.0 mmol) was suspended in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and (COCl)<sub>2</sub> added (725 mg, 5.7 mmol). A catalytic amount of DMF was then added and the mixture stirred for 4 hrs. The solvent was then removed *in vacuo* to give the acid chloride as an oil. The oil was dissolved in CHCl<sub>3</sub> (10 mL). Methyl 2-amino-5-nitrobenzoate (196 mg, 1.0 mmol) was added followed by pyridine (1 mL).

10

15

20

The solution was stirred at room temperature for an additional 12 hrs then poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The resulting residue was purified by silica gel chromatography (20% EtOAc in hexane) to provide 108 mg of the desired methyl ester (24%). The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated *in vacuo*. The title compound (70 mg, 67%, 16% overall) was obtained as a yellow solid after recrystalization from MeOH. H NMR (400 MHz, DMSO) 2.94 (m, 4H), 3.65 (m, 4H), 7.94 (t, 1H), 8.04 (d, 1H), 8.29-8.33 (m, 2H), 8.55 (dd, 1H), 8.80 (d, 1H), 8.91 (d, 1H), 12.76 (s, 1H)

Methyl 2-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-formylbenzoate was prepared as described above using 3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoic acid. 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-formylbenzoic acid was prepared by hydrolizing the corresponding methyl ester.  $^{1}$ H NMR (400 MHz, DMSO- $d_6$ ) δ 12.69 (s, 1 H), 10.00 (s, 1 H), 8.87 (d, J = 8.7 Hz, 1 H), 8.61 (d, J = 2.1 Hz, 1 H), 8.40 (s, 1 H), 8.27 (d, J = 7.9 Hz, 1 H), 8.19 (dd, J = 8.7, 2.1 Hz, 1 H), 8.09 (d, J = 8.5 Hz, 1 H), 7.84 (t, J = 7.8 Hz, 1 H), 7.53 (d, J = 8.5 Hz, 1 H), 7.24-7.29 (m, 2 H), 4.02 (t, J = 8.5 Hz, 2 H).

## 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-[(E)-(methoxyimino)methyl]benzoic acid

25

30

A slurry of methyl 2-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-formylbenzoate (475 mg, 0.952 mmol) and O-methylhydroxylamine hydrochloride (526 mg, 6.30 mmol, Aldrich) in 1:1 ethanol/pyridine (25 mL) was stirred for 2 days. The mixture was then added to a separatory funnel with 120 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl followed by 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in

the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from CH<sub>2</sub>Cl<sub>2</sub> to 2% EtOAc in CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 411 mg of white solid as the methyl ester. To a mixture of the corresponding methyl ester (288 mg, 0.545 mmol) in dioxane (20 mL) was added 1 M aqueous sodium hydroxide (1.5 mL). The mixture was stirred at room temperature for 4.5 hours and then in a 50 °C oil bath for 30 minutes. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of EtOAc. The EtOAc was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. It was then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol/THF. The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 127 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.40 (s, 1 H), 8.70 (d, J = 8.7 Hz, 1 H), 8.38 (s, 1 H), 8.31 (d, J = 2.1Hz, 1 H), 8.30 (s, 1 H), 8.25 (d, J = 7.9 Hz, 1 H), 8.07 (d, J = 8.1 Hz, 1 H), 7.91 (dd, J= 8.7, 2.1 Hz, 1 H), 7.83 (t, J = 7.9 Hz, 1 H), 7.52 (d, J = 8.5 Hz, 1 H), 7.24-7.29 (m, 2) H), 4.02 (t, J = 8.5 Hz, 2 H), 3.91 (s, 3 H), 2.95 (t, J = 8.4 Hz, 2 H).

## 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-[(E)-(hydroxyimino)methyl]benzoic acid

20

25

30

5

10

15

A slurry of methyl 2-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-formylbenzoate (627 mg, 1.26 mmol) and hydroxylamine hydrochloride (656 mg, 9.44 mmol, Mallinckrodt) in 1:1 ethanol/pyridine (25 mL) was stirred for 2 days. The mixture was then added to a separatory funnel with 120 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl followed by 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with 5% EtOAc in CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 478 mg of white solid as the methyl ester. To a mixture of the corresponding methyl ester (363 mg, 0.706 mmol) in dioxane (20 mL) was added 1 M aqueous sodium

5

10

15

hydroxide (1.5 mL). The mixture was stirred at room temperature for 4.5 hours. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of EtOAc. The EtOAc was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. It was then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol/THF. The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 280 mg of white solid. Because NMR and CHN analysis were consistent with this material containing residual solvent, 200 mg of the material was heated in 50 mL of methanol. Solvent was removed, and the residue was again dried at 100 °C under vacuum yielding 183 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.37 (s, 1 H), 11.31 (s, 1 H), 8.68 (d, J = 8.7 Hz, 1 H), 8.38 (s, 1 H), 8.29 (d, J = 1.9 Hz, 1 H), 8.25 (d, J = 7.9 Hz, 1 H), 8.20 (s, 1 H), 8.07 (d, J = 8.1 Hz, 1 H), 7.90 (dd, J = 8.8, 2.0 Hz, 1 H), 7.83 (t, J = 7.9 Hz, 1 H), 7.53 (d, J = 8.5 Hz, 1 H), 7.24-7.29 (m, 2 H), 4.01 (t, 8.5, 2 H), 2.95 (t, J = 8.4 Hz, 2 H).

#### Scheme 1.5

#### 20 Scheme 1.6

## Scheme 1.7

$$\begin{array}{c} O_2N \\ O_2N \\ O_2N \\ O_3N \\ O_4N \\ O_5N \\ O_$$

## 5 **Scheme 1.8**

## Scheme 1.9

10

## Scheme 1.10

#### Scheme 1.11

#### **Scheme 1.12**

#### **Scheme 1.13**

5

10

Compounds produced via the above-descirbed synthetic schemes include, but are not limited to, the following:

5-Chloro-2-({4-[(dipropylamino)sulfonyl]benzoyl}amino)benzoic acid

- 5-Chloro-2-({4-[(dipropylamino)sulfonyl]-3-nitrobenzoyl}amino)benzoic acid
- 5-Chloro-2-{[4-[(dipropylamino)sulfonyl]-3-(hydroxyamino)benzoyl]amino} benzoic acid hydrochloride
- 2-({3-Amino-4-[(dipropylamino)sulfonyl]benzoyl}amino)-5-chlorobenzoic acid bydrochloride
  - 2-{[4-(Benzylsulfanyl)-3-nitrobenzoyl]amino}-5-chlorobenzoic acid
  - 5-Chloro-2-({4-[(dipropylamino)sulfanyl]-3-nitrobenzoyl}amino)benzoic acid
  - Methyl 5-chloro-2-({4-[(dipropylamino)sulfinyl]-3-nitrobenzoyl}amino)benzoate
- 5-Chloro-2-{[4-(2,3-dihydro-1H-indol-1-ylsulfonyl)-3-nitrobenzoyl]amino} benzoic acid
  - Cyano 2-{[3-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}-5-ethynylbenzoic acid
  - Methyl 2-({3-amino-4-[(dipropylamino)sulfonyl]benzoyl}amino)-5-chlorobenzoate
- 15 2-({3-Bromo-4-[(dipropylamino)sulfonyl]benzoyl}amino)-5-chlorobenzoic acid

10

## Scheme 1.16

## Preparation of 4-{[4-chloro(methyl)anilino]sulfonyl}benzoic acid

A solution of 4-chloro-N-methylaniline (10.0 g, 0.0706 mol, 1.1 eq) and triethylamine (7.78 g, 0.0770 mol, 1.2 eq) in 140 mL of methanol, cooled in an ice bath at 0-5°C, was treated portionwise over a one minute period with solid 4-chlorosulfonyl benzoic acid (14.2 g, 0.0642 mol, 1.0eq). After the addition was complete, the cooling bath was removed and the reaction mixture was stirred under a nitrogen atmosphere while

warming to room temperature on its own. After 5.5 h, the contents were poured into 270 mL of ice water containing 130 mL of 3 N NaOH, washed the milky solution with methylene chloride (2 X 100 mL), acidified the aqueous layer with 35 mL of concentrated HCl. After cooling the mixture in an ice bath, the white precipitated product was collected and dried in a vacuum oven at  $70^{0}$ C overnight to yield 14.92 g (71%) of **2**. <sup>1</sup>H NMR (DMSO- $d_{6}$ )  $\delta$  13.53 (brs, 1 H), 8.11 (dd, J = 2, 7 Hz, 2 H), 7.63 (dd, J = 2, 7 Hz, 2 H), 7.42 (dd, J = 2, 7 Hz, 2 H), 7.14 (dd, J = 2, 7 Hz, 2 H), 3.15 (s, 3 H) ppm.

#### Scheme 1.17

5

10

15

20

$$NBS$$
 $NBS$ 
 $NBS$ 

To 21 mL of carbon tetrachloride at room temperature was added benzoyl peroxide (0.095 g, 0.393 mmol, 0.10 eq). The solution was slowly heated to reflux at which time N-bromosuccinimide (0.769 g, 4.32 mmol, 1.1 eq) was added at once followed by a slurry of compound 72 (1.64 g, 3.93 mmol, 1.0 eq) in 9 mL of carbon tetrachloride plus 6 mL of carbon tetrachloride as a rinse. Vigorous refluxing was continued for 2 h, the reaction mixture filtered hot and the solids rinsed with additional hot carbon tetrachloride. The filtrate was concentrated at reduced pressure to give more than theoretical amount of crude bromomethyl compound 73. This was dissolved in 35 mL of acetone, treated with NaCN (0.289 g, 5.90 mmol, 1.5 eq) and

10

15

NaI (0.029 g, 0.197 mmol, 0.05 eq) and the mixture refluxed for 24 h. An additional 0.50 eq (0.096 g) of NaCN was added and refluxing continued for 3 h longer. The cooled reaction mixture was filtered, the filtrate concentrated at reduced pressure, the residue dissolved in ethyl acetate and washed successively with 10 mL of water and 10 mL of 50% saturated brine. The combined aqueous washings were back extracted once with ethyl acetate, the combined organic extracts dried with anhydrous sodium sulfate and the filtrate concentrated in vacuo. Chromatography with 100 g of silica gel, packed and eluted with acetone-methylene chloride-heptane (1:4:5), afforded cyanomethyl ester 74 in 20% yield (based on 72) as a white solid. Base hydrolysis of 74 (0.297 g, 0.670 mmol) in 4 mL of methylene chloride, 4 mL of methanol and 1 mL of water using 1N NaOH (3.02 mL, 4.5 eq) at room temperature gave a 55% yield of acid 75 as a white solid. 73: TLC (silica gel GF):  $R_f = 0.36$  acetone-methylene chloride-hexane(1:3:6); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.89 (d, J = 7 Hz, 1 H), 8.41 (t, J = 1 Hz, 1 H), 8.27 (m, 1 H), 8.14 (d, J = 2 Hz, 1 H), 7.97 (m, 1 H), 7.75 (t, J = 6 Hz, 1 H), 7.66 (dd, J = 2, 6 Hz, 1 H), 4.52 (s, 2 H), 3.98 (s, 3 H), 3.78 (t, J = 3 Hz, 4 H), 3.10 (t, J = 4 Hz, 4 H) ppm.

Scheme 1.18 outlines the solid phase synthesis of halogenated anthranilic acid substrates 5.

#### 20 Scheme 1.18

Resin bound iodide 6 was stannylated using the conditions shown in Scheme 10.2. Hunigs base, although not directly involved in the reactions, was used as a proton scavenger. A library based on this template was successfully prepared using Suzuki cross-coupling conditions.

#### Scheme 1.19

5

10

15

Applying the Stille conditions to the template, stannylated product 9 was prepared from iodide 5b. The reaction was monitored via observance of the protodestannylation product after TFA cleavage from resin. Stannylation of the corresponding solid-phase bromide 5a was less successful.

#### Scheme 1.20

Attempts at coupling aryl bromides and iodides with the stannylated resin gave some product, but not in quantities suitable for library production (Scheme 1.21).

Protodestannylation and homocoupling were the major competing reactions, leaving

product purities in the 25 % range. The reactions were monitored by HPLC (at 210 nm), and product identities were confirmed by LC/MS.

#### Scheme 1.21

Suzuki coupling chemistry was conducted under the conditions shown in Scheme 1.22.

### **Scheme 1.22**

5

10

The cross-coupling reaction from the other direction is shown in Scheme 1.23, in which purchased aryl tin compounds were coupled with the resin-bound iodide.

#### **Scheme 1.23**

The best results in the case of tributylphenyl tin were obtained in toluene with 1,1'
bis(diphenylphosphino)-ferrocene as ligand and a reaction time of 2.5 hours at 115 °C.

In the case of 2-(tributylstannyl) thiophene, toluene was the solvent of choice and tricyclohexylphosphine, triphenyl arsine, and 1,1'-bis(diphenylphosphino)-ferrocene worked equally well after 2.5 hours at 115 °C.

#### 10 Table 1.1: Commercially Available Aryl Tin Compounds

## Installation of Ketones via Palladium-Catalyzed Coupling with Acid Chlorides

Acid chlorides were coupled with 9 (see scheme 1.20) using similar, but milder conditions (Scheme 10.7). The ketone product (13) was produced using triphenylphosphine as ligand and THF as solvent in 75 % yield and 70 % purity. A

carbon monoxide atmosphere was used to eliminate small amounts of the corresponding aryl-aryl product formation (12), while Hunigs base was employed as the proton scavenger to help avoid protodestannylation.

## Scheme 1.24

5

- 65 -

## Scheme 1.25

where R3 is a C1-4 alkyl optionally substituted with halo, -OH, CN, and NO<sub>2</sub>.

## 5 **Table 1.2**: Diversity Elements (no. of ≥ 70 % pure products/no. attempted)

sulfonyl chlorides	amines	acid chlorides	
O_OH (39/96)	CI (12/24)	0 (7/16)	O/16)
0=\$=0     CI	NH (12/24)		(a.i.s)
ОН	NH (11/24)	Cí 16/16)	(6/16)
O=S=O CI	NH N-(-)-N-(12/24)	O F F (1/16)	<sup>lodo</sup> (11/16)
	NH N- (6/24)	O (4/16)	H (13/16)
	NH N- (6/24)	CI (6/16)	
	C. (9/24)	O Ci (3/16)	
	NH HCI	(7/16)	
	(4/24)	CI S (8/16)	
	NH <sub>2</sub>		

# Derivatization of Aryl Ketones: Derivatizing the ketones as oximes, alkoxyamines, hydrazones, and amines

Oximes and alkoxyamines (20) were prepared in reasonable purities from their corresponding hydroxylamine hydrochlorides and resin 19 in pyridine (Scheme 1.26). Hydrazone, sulfonylhydrazone, and acyl-hydrazone formations (21) using literature conditions, however, were sluggish and could never be pushed to completion.

#### 10 Scheme 1.26

Amines 22 were prepared on solid-phase using reductive amination. Imine formation, mediated by titanium isopropoxide, typically took four to six hours to go to completion. The sodium triacetoxy borohydride reduction was allowed to proceed overnight to give good quality amine products.

#### Scheme 1.30

15

TOO CCOOL

5

10

15

20

## t-Butyl 2-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-iodobenzoate, a, Compound 11.1

3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoic acid (2.3 g, 6.9 mmol, 1 equivalent) and oxyl chloride (2.6 g, 20.5 mmol, 3 equivalent) were dissolved in methylene chloride (30 ml), followed by the addition of DMF (0.4 ml). Gas evolution was observed. The mixture was stirred at room temperature for 2 h later, then heptane (30 ml) was added. The solution was concentrated to dryness, and the residue was redissolved in DCM (30 ml), followed by the dropwise addition of PHA-561052 (2.2 g, 6.9 mmol, 1 equivalent) in DCM (20 ml) and pyridine (1.2 ml). The resulting solution was stirred overnight, then diluted with MTBE (200 ml) and washed with 0.1N HCl, 1N NaOH, brine, dried (MgSO<sub>4</sub>), filtered, and concentrated *in vacuo*. The residue was recrystallized from hepane to afford 2.4 g (55%) of 1 as a yellow solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.34 (s, 1 H), 8.67 (d, J = 9.0 Hz, 1 H), 8.54 (s, 1 H), 8.33 (m, 1 H), 8.24 (d, J = 8.5 Hz, 1 H), 7.97 (d, J = 8.4 Hz, 1 H), 7.88 (d, J = 8.5 Hz, 1 H), 7.64 (m, 2 H), 7.17 (d, J = 8.5 Hz, 1 H), 7.06 (s, 1 H), 4.08 (t, J = 8.5 Hz, 2 H), 2.95 (t, J = 8.4 Hz, 2 H), 1.67 (s, 9 H).

10

# $2-(\{3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl\}amino)-5-iodobenzoic\ acid \\$

5 General method E: (Hydrolysis of the alkyl ester)

Ester 11.1 (150mg, 0.24mmol) was dissolved in DCM (6 ml), followed by the addition of TFA (1.2 ml). The solution was shaken overnight, then diluted with DCM (5 ml) and heptane (1 ml). The solution was concentrated *in vacuo* to dryness, the residue was pumped for about 1h, then triturated with methanol, filtered to afford 102mg(75%) of a white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.25 (s, 1 H), 8.44 (d, J = 9 Hz, 1 H), 8.33 (s, 2 H), 8.31 (m, 1 H), 8.05 (m, 2 H), 7.81 (t, J = 8.5 Hz, 1 H), 7.71 (d, J = 9 Hz, 1 H), 7.24 (m, 2 H), 4.01 (t, J = 8.1 Hz, 2 H), 2.95 (t, 2 H).

TOO CCOOL

t-Butyl 4-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)[1,1'-biphenyl]-3-carboxylate, 2a

5

10

15

General method F:

Ester 11.1 (150 mg, 0.235 mmol) and tetrakis(triphenylphosphine) palladium(0) (13.6 mg, 0.01175 mmol) were placed in a 50ml one-necked round bottom flask. The system was evacuated and filled with argon several times. Then tributylstannylbenzene (91.75 mg, 0.25 mmol) in toluene (10 ml) was added. The resulting solution was heated at 100°C overnight, cooled to room temperature, then KF (87mg, ) was added. The mixture was stirred at room temperature for 2h, filtered through celite. The filtrate was concentrated *in vacuo* and the residue was purified by silica gel chromatography (EtOAc/heptane 1/25, 1/10) to afford 120 mg (88%) of 11.2a as a yellow solid.

General method G:

10

Ester **11.1** (150 mg, 0.235 mmol) and dichlorobis(triphylphosphine) palladium (II) (8.4 mg, 0.012 mmol) were placed in a 50 ml one-necked round bottom flask. The system was evacuated and filled with argon several times. Then tributylstannylbezene (91.7 mg, 0.25 mmol) in THF (10 ml) was added. The resulting solution was heated at 80°C overnight, cooled to room temperature, KF (87 mg) was added. The mixture was stirred at room temperature for 2h, filtered through celite. The filtrate was concentrated *in vacuo* and the residue was purified by silica gel chromatography (EtOAc/heptane 1/25, 1/10) to afford 101 mg (74%) of **11.2a** as a yellow solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  11.60 (s, 1 H), 8.42 (s, 1 H), 8.35 (D, J = 9 Hz, 1 H), 8.27 (d, J = 8 Hz, 1 H), 8.15 (d, J = 2 Hz, 1 H), 8.06 (d, J = 8 Hz, 1 H), 7.98 (d, J = 9 Hz, 1 H), 7.84 (t, J = 8 Hz, 1 H), 7.70 (d, J = 7 Hz, 2 H), 7.50 (m, 3 H), 7.41 (t, J = 7 Hz, 1 H), 7.25 (d, J = 7 Hz, 2 H), 4.05 (t, J = 8 Hz, 2 H), 2.97 (t, J = 8 Hz, 2 H), 1.53 (s, 9 H).

t-Butyl 4'-chloro-4-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)[1,1'-biphenyl]-3-carboxylate, 11.2c

#### General method H:

Ester 11.1 (160 mg, 0.25 mmol), tetrakis(triphenylphosphine) palladium(0) (14.5 mg, 0.0125 mmol), sodium carbonate (101 mg, 0.95 mmol) and 4-chlorobenzeneboronic acid (43 mg, 0.275 mmol) were placed in a 100ml one-necked round bottom flask. The system was evacuated and filled with argon several times. Then THF (50 ml) and distilled water (5 ml) were added. The solution was heated at reflux temperature for 20h, the solvent was removed in vacuo and residue was purified by silica gel

chromatography (EtOAc/hepatane 1/25, 1/10) to get 92 mg (59%) of **11.2c** as a yellow solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  12.40 (s, 1 H), 8.93 (d, J = 9 Hz, 1 H), 8.59 (s, 1 H), 8.28 (d, J = 8 Hz, 1 H), 8.24 (d, J = 2.3 Hz, 1 H), 7.95 (d, J = 8 Hz, 1 H), 7.80 (dd, J = 2.5, 8.2 Hz, 1 H), 7.64 (m, 6 H), 7.20 (d, J = 8 Hz, 1 H), 7.08 (s, 1 H), 4.12 (t, J = 8 Hz, 2 H), 2.98 (t, J = 8 Hz, 2 H), 1.71 (s, 9 H).

#### 5-cyano-2-({3-[(1-pyrrolidinylsulfonyl)methyl]benzoyl}amino)benzoic acid PHA-

630852,

10

15

3-(chloromethyl)benzoic acid **51**, gave thiomethyl compound **52** in 82% yield<sup>7</sup>. In a manner similar to that described for the preparation of compound **13** above, compound **52** was sequentially treated with gaseous chlorine to obtain the crude sulfonic acid **53** in theoretical yield followed by reaction with thionyl chloride which provided the crude acid chloride **54** as a waxy white solid. This was reacted directly with anthranilate **21** to provide sufficiently pure sulfonyl chloride **55**, which was reacted with pyrrolidine to give a 26% yield of ester **56**. Subsequent hydrolysis with trifluroacetic acid afforded the acid **57** in 83% yield as a white solid. **57**: <sup>1</sup>H NMR

15

20

25

30

(DMSO- $d_6$ )  $\delta$  12.48 (s, 1 H), 8.86 (d, J = 7 Hz, 1 H), 8.42 (d, J = 2 Hz, 1 H), 8.12 (dd, J = 2, 7 Hz, 1 H), 8.05 (s, 1 H), 7.95 (d, J = 6 Hz, 1 H), 7.72 (d, J = 6 Hz, 1 H), 7.64 (t, J = 6 Hz, 1 H), 4.58 (s, 2 H), 3.20 (t, J = 5 Hz, 4 H), 1.82 (m, 4 H) ppm.

5 **2-[(1,3-Benzoxazol-2-ylcarbonyl)amino]-5-cyanobenzoic acid** (36310-jcr-135a, PHA-734774, SPS# 0281864)

To a solution of benzyl 1,3-benzoxazole-2-carboxylate (233 mg, 0.920 mmol) in 1:1 ethanol/THF (20 mL) was added palladium on carbon (56 mg of 5%, Aldrich) and triethylamine (180 µL, 1.29 mmol, Aldrich). The mixture was stirred under 1 ATM of hydrogen for 2 hours and then filtered through a plug of celite. Removal of the solvent left the triethylamine salt as an orange oil (the protonated form of the acid rapidly decarboxylates and should be avoided). This oil was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) and treated with DMF (20 μL) followed by oxalyl chloride (220 μL, 2.52 mmol, Aldrich). Solvent and excess oxalyl chloride were removed by rotary evaporation after 76 hours. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL), and benzyl 2-amino-5cyanobenzoate (250 mg, 0.991 mmol) in pyridine (8 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 of 1.0 M HCl and 100 mL of brine. Product was adsorbed onto silica gel and purified on a Biotage Flash 40 M siliga gel cartridge using CH<sub>2</sub>Cl<sub>2</sub> as eluent. Product was collected as 218 mg of white solid as the benzyl ester. A mixture of benzyl 2-[(1,3-benzoxazol-2-ylcarbonyl)amino]-5-cyanobenzoate (168 mg, 0.423 mmol) and palladium on carbon (33 mg of 5%, Aldrich) in 2:1 THF/ethanol (30 mL) was stirred under 1 ATM of hydrogen for 25 minutes. The mixture was filtered through a plug of celite and then evaporated. The residue was dried at 100 °C under vacuum yielding 116 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO-D6)  $\delta$  ppm 7.56 (t, J=7.67 Hz, 1 H) 7.63 (t, J=7.88 Hz, 1 H) 7.94 (d, J=8.29 Hz, 1 H) 8.00 (d, J=7.67 Hz, 1 H) 8.16 (dd, J=8.81, 1.97 Hz, 1 H) 8.45 (d, J=2.07 Hz, 1 H) 8.87 (d, *J*=8.71 Hz, 1 H) 13.16 (s, 1 H).

The following compounds were produced via the methods described above using appropriate starting materials and making non-critical variations.

```
4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)[1,1'-biphenyl]-
 5
     3-carboxylic acid
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(2-
     furyl)benzoic acid
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(2-
     thienyl)benzoic acid
10
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(2-
     pyrazinyl)benzoic acid,
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(1-methyl-
     1H-pyrrol-2-yl)benzoic acid
     4'-Chloro-4-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)[1,1'-
15
     biphenyl]-3-carboxylic acid
     4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-3'-nitro[1,1'-
     biphenyl]-3-carboxylic acid
     4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-4'-cyano[1,1'-
     biphenyl]-3-carboxylic acid
20
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(5-chloro-2-
     thienyl)benzoic acid
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(4-methyl-2-
     thienyl)benzoic acid
     4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-4'-fluoro[1,1'-
25
     biphenyl]-3-carboxylic acid
     4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-2'-
     (trifluoromethyl)[1,1'-biphenyl]-3-carboxylic acid
     4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-3',5'-
     bis(trifluoromethyl)[1,1'-biphenyl]-3-carboxylic acid
30
     2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(5-methyl-2-
     thienyl)benzoic acid
```

```
4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-2',4'-difluoro[1,1'-biphenyl]-3-carboxylic acid
4'-t-Butyl-4-({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)[1,1'-biphenyl]-3-carboxylic acid
```

4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-3' (trifluoromethyl)[1,1'-biphenyl]-3-carboxylic acid
 4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-4' (trifluoromethyl)[1,1'-biphenyl]-3-carboxylic acid
 4-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-2'-methyl[1,1'-biphenyl]-3-carboxylic acid

```
2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(3,5-dimethyl-4-isoxazolyl)benzoic acid
```

- 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-(2,4-dimethoxy-5-pyrimidinyl)benzoic acid
- 5 2-[(3-{[(4-Chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino]-5-(trifluoromethyl)benzoic acid,
  - 2-[(3-Bromo-5-{[(4-chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino]-5-chlorobenzoic acid
  - $5\text{-}Bromo-2\text{-}[(3\text{-}\{[(4\text{-}chlorophenyl)(methyl)amino}] sulfonyl\}\text{-}5\text{-}$
- 10 nitrobenzoyl)amino]benzoic acid
  - 2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-cyanobenzoic acid
  - 5-Bromo-2-{[3-cyano-5-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}benzoic acid
- 5-Cyano-2-{[3-(2,3-dihydro-1H-indol-1-ylsulfonyl)-5-methylbenzoyl]amino}benzoic acid
  - Methyl 2-{[3-[2-(acetyloxy)ethyl]-5-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl] amino}-5-cyanobenzoate
  - $5-Cyano-2-\{[3-(2,3-dihydro-1H-indol-1-ylsulfonyl)-5-(2-hydroxyethyl)benzoyl]\\$
- 20 amino}benzoic acid
  - $\hbox{$2-\{[3-Bromo-5-(2,3-dihydro-1H-indol-1-y|sulfonyl)$benzoyl]amino}-5-chlorobenzoic acid$
  - 5-Chloro-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino]benzoic acid
- 25 2-[(3-Bromo-5-{[(4-chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino]-5-cyanobenzoic acid
  - 5-cyano-2-[(3-{[(2-hydroxyphenyl)(methyl)amino]sulfonyl}benzoil)amino]benzoic acid
  - $5\text{-}Bromo-2\text{-}[(5\text{-}\{[(4\text{-}chlorophenyl)(methyl)amino}] sulfonyl\}\text{-}2\text{-}methoxybenzoyl)$
- 30 amino]benzoic acid
  - 5-Bromo-2-[(5-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-2-methylbenzoyl)amino]benzoic acid

- 5-Bromo-2-[(2-bromo-5-{[(4-chlorophenyl)(methyl)amino]sulfonyl} benzoyl)amino]benzoic acid
- 5-Bromo-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-4-methoxybenzoyl) amino]benzoic acid
- 5 5-Bromo-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-4-methylbenzoyl)amino]benzoic acid
  - 5-Bromo-2-[(4-bromo-3-{[(4-chlorophenyl)(methyl)amino]sulfonyl} benzoyl)amino]benzoic acid
  - $2\hbox{-}[(3\hbox{-}\{[(4\hbox{-}Chlorophenyl)(methyl)amino}]\hbox{-}s-nitrobenzoic acid$
- 2-[(4-{[(4-Chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino]-5-nitrobenzoic acid
  - 5-Bromo-2-[(3-{[(4-chlorophenyl)(methyl)amino]sulfonyl}-4-morpholin-4-ylbenzoyl)amino]benzoic acid
  - 5-Bromo-2-[(3-bromo-5-{[(4-chlorophenyl)(methyl)amino]sulfonyl}benzoyl)amino] benzoic acid
- 2-{[3-Bromo-5-(2,3-dihydro-1H-indol-1-ylsulfonyl)benzoyl]amino}-5-cyanobenzoic acid
  - 2-{[3-Bromo-5-(morpholin-4-ylsulfonyl)benzoyl]amino}-5-chlorobenzoic acid 5-Chloro-2-{[3-(2,3-dihydro-1H-indol-1-ylsulfonyl)-5-methylbenzoyl]amino}benzoic acid
- 5-Iodo-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid
  2-({4-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-cyanobenzoic acid
  - 2-{[3-(Morpholin-4-ylsulfonyl)benzoyl]amino}-5-thiocyanatobenzoic acid

#### 25 Example 2: Amine, Ether, and Thioether Derivatives

#### Preparation of 3-Bromo-4-fluorobenzoic acid

30

3-Bromo-4-fluoro-benzaldehyde (10.0 g, 49 mmol) in H<sub>2</sub>O(150 mL, followed by the addition of KMnO<sub>4</sub> (15.5 g, 98 mmol) heated at reflux (foams extensively) for 1 h,

then added additional KMnO<sub>4</sub> (15.5 g, 98 mmol) and continued heating for another 3 h. The reaction was cooled to rt, then filtered through Celite. The solution was acidified with HCl, and the resulting white precipitate was filtered off, to afford 6.1 g (56%) of a white solid.

5

10

15

#### Preparation of 3-Anilinobenzoic acid

Methyl 3-bromobenzoate (1000 mg, 4.65 mmol),  $Pd_2(dba)_3$  (53 mg, 0.058 mmol),  $Cs_2CO_3$  (2120 mg, 1.4 mmol) and N-[2'-(dicyclohexylphosphino)-1,1'-biphenyl-2-yl]-N,N-dimethylamine (27mg, 0.07 mmol) were placed in a 100ml one-necked round bottom flask. The system was evacuated and filled with argon several times. Then aniline (519 mg, 5.58 mmol) was added, followed by the addition of toluene (50 ml). The solution was heated at 100°C for 20h, the solvent was removed in vacuo and residue was purified by silica gel chromatography (EtOAc/hepatane 1/3) to get 180 mg (18%) of methyl ester as a yellow solid, which was hydrolyzed by LiOH (50 mg)) in THF (4 ml) and water (1 ml) to afford 140 mg (82%) of **3-Anilinobenzoic acid** as a white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.02 (s, 1 H), 7.65 (s, 1 H), 7.33 (d, J = 7.5 Hz, 1 H), 7.19 (t, J = 8.3 Hz, 2 H), 7.10 (d, J = 7.7 Hz, 1H), 7.03 (d, J = 7.6 Hz, 2 H), 6.96 (m, 1 H), 6.76 (t, J = 7.3 Hz, 1 H);

20

25

#### 2-[(3-Anilinobenzoyl)amino]-5-cyanobenzoic acid

Prepared according to the general methods described above: 3-Anilinobenzoic acid (140 mg, 0.66 mmol) and PHA-561053 (130 mg, 0.59 mmol) afforded 61 mg (25%)

of t-butyl ester as a yellow solid, which was hydrolyzed to 48 mg (91%) of a green solid.

#### Analytical data for PHA-610938

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.81 (d, J = 9.0 Hz, 1 H), 8.46 (s, 1 H), 8.35 (d, J = 2.2 Hz, 1 H), 7.82 (dd, J = 1.9, 8.8 Hz, 1 H), 7.72 (s, 1 H), 7.42 (m, 2 H), 7.27 (m, 3 H), 7.14 (d, J = 7.8 Hz, 2 H), 6.88 (t, J = 7.3 Hz, 1 H).

# Preparation of 3-[(Pyridin-4-ylmethyl)thio]benzoic acid

10

15

5

Water (10 mL) was added to a flask containing 3-mercaptobenzoic acid (2.08 g, 13.5 mmol, Aldrich) and sodium hydroxide (1.16 g, 29.0 mmol). To the resulting solution was added 4-picolyl chloride hydrochloride (2.31 g, 14.1 mmol, Aldrich) and ethanol (20 mL). The mixture was heated in a 75 °C oil bath for 1 hour and then added to a separatory funnel with 100 mL of water and 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This resulted in a suspension in the aqueous layer. This suspension was washed with an additional 100 mL of CH<sub>2</sub>Cl<sub>2</sub> and then filtered. The solid was then dried at 100 °C under vacuum yielding 2.80 g of white solid.

#### 20 Preparation of 3-[(Phenylthio)methyl]benzoic acid

10

15

20

25

30

To a solution of the corresponding methyl ester described by Holoboski, M.A.; Koft, E. in *J. Org. Chem.*, **1992**, *57*, 965-969, (1.23 g, 4.76 mmol) in methanol (15 mL) was added 1.0 M aqueous NaOH (8.0 mL). The resulting mixture was heated in a 50 °C oil bath for 1.5 hours. Most of the methanol was removed by rotary evaporation, and the residue was added to a separatory funnel with 100 mL of 1.0 M aqueous HCl and 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> was washed with another 100 mL of 1.0 M aqueous HCl followed by 100 mL of water and then dried over Na<sub>2</sub>SO<sub>4</sub>. Solvent was removed, and the residue was dried at 100 °C yielding 1.11 g of white solid.

### 5-Bromo-2-({3-[(phenylthio)methyl]benzoyl}amino)benzoic acid

To 3-[(phenylthio)methyl]benzoic acid (400 mg, 1.64 mmol) in  $CH_2Cl_2$  (15 mL) was added DMF (20  $\mu$ L) and oxalyl chloride (200  $\mu$ L, 2.29 mmol). The mixture was stirred for 1.5 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in  $CH_2Cl_2$  (15 mL), and methyl 2-amino-5-bromobenzoate (330 mg, 1.43 mmol, Avocado) in pyridine (8 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of  $CH_2Cl_2$ . This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The  $CH_2Cl_2$  was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 50%  $CH_2Cl_2$ /heptane to 75%  $CH_2Cl_2$ /heptane as eluent. Yield was 544 mg of white solid as the methyl ester.

To a mixture of the corresponding methyl ester (386 mg, 0.845 mmol) in dioxane (20 mL) was added 1 M aqueous sodium hydroxide (2.0 mL). The mixture was stirred for at room temperature for 1.25 hours and then at 50 °C for 1.5 hours. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of brine. It was then dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol (8

mL). The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 279 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.08 (s, 1 H), 8.64 (d, J = 9.2 Hz, 1 H), 8.12 (d, J = 2.5 Hz, 1 H), 7.97 (s, 1 H), 7.86 (dd, J = 9.2, 2.5 Hz, 1 H), 7.80 (d, J = 7.6 Hz, 1 H), 7.61 (d, J = 7.6 Hz, 1 H), 7.51 (t, J = 7.6 Hz, 1 H), 7.35 (d, J = 7.1 Hz, 2 H), 7.29 (t, J = 7.9 Hz, 2 H), 7.18 (t, J = 7.1 Hz, 1 H), 7.35 (s, 2 H).

Other compounds produced via the above-described methodology using appropriate starting materials and maiking non-critical variations include:

- 10 2-{[3-(benzylthio)benzoyl]amino}-5-bromobenzoate
  - 2-{[3-(Benzyloxy)benzoyl]amino}-5-bromobenzoic acid
  - 5-Bromo-2-{[3-(ethylthio)benzoyl]amino}benzoic acid
  - Methyl-5-Bromo-2-({3-[(pyridin-4-ylmethyl)thio]benzoyl}amino)benzoate
  - 5-Bromo-2-({3-[(pyridin-4-ylmethyl)thio]benzoyl}amino)benzoic acid
- 5-bromo-2-({3-[(pyridin-4-ylmethyl)thio]benzoyl}amino)benzoic acid hydrochloride
  - 5-Bromo-2-[(3-phenoxybenzoyl)amino]benzoic acid
  - 5-Bromo-2-{[3-(phenylthio)benzoyl]amino}benzoic acid
  - 5-Cyano-2-[(3-phenoxybenzoyl)amino]benzoic acid
  - 5-Cyano-2-({3-[(pyridin-4-ylmethyl)thio]benzoyl}amino)benzoic acid
- 5-Cyano-2-({3-[(pyridin-4-ylmethyl)thio]benzoyl}amino)benzoic acid hydrochloride
  - 2-{[3-(Benzyloxy)benzoyl]amino}-5-cyanobenzoic acid
  - 2-{[3-(Benzylthio)benzoyl]amino}-5-cyanobenzoic acid
  - 5-cyano-2-({3-[(1-phenylethyl)thio]benzoyl}amino)benzoic acid
  - 5-cyano-2-{[3-(cyclopentylthio)benzoyl]amino}benzoic acid
- 25 5-cyano-2-{[3-(cyclopentylsulfinyl)benzoyl]amino}benzoic acid
  - 5-Chloro-2-[(4-methoxy-3-nitrobenzoyl)amino]benzoic acid
  - 2-{[4-(Benzylsulfanyl)-3-bromobenzoyl]amino}-5-chlorobenzoic acid
  - 5-Cyano-2-{[3-(3-fluorophenoxy)benzoyl]amino}benzoic acid
  - 5-Cyano-2-{[3-(2-methylphenoxy)benzoyl]amino}benzoic acid
- 30 5-Cyano-2-{[3-(4-methoxyphenoxy)benzoyl]amino}benzoic acid
  - 5-Cyano-2-{[3-(3-nitrophenoxy)benzoyl]amino}benzoic acid

#### **Example 3: KETONE DERIVATIVES**

#### 2-[(3-Benzoylbenzoyl)amino]-5-bromobenzoic acid

5

10

15

20

25

To 3-benzoylbenzoic acid (633 mg, 2.80 mmol, Aldrich) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added DMF (20 µL) and oxalyl chloride (450 µL, 5.16 mmol). The mixture was stirred for 1.7 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL), and methyl 2amino-5-bromobenzoate (565 mg, 2.46 mmol, Avocado) in pyridine (6 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 75% CH<sub>2</sub>Cl<sub>2</sub>/heptane to 100% CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 825 mg of white solid as the methyl ester. To a mixture of the corresponding methyl ester (645 mg, 1.47 mmol) in dioxane (20 mL) was added 1 M aqueous sodium hydroxide (3.0 mL). The mixture was stirred in a 50 °C oil bath for 2 hours. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The organics were washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. They were then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol/THF. The solids were washed with ethanol followed by pentane and then dried at 100 °C under vacuum yielding 329 mg of white solid. <sup>1</sup>H NMR (400 MHz. DMSO- $d_6$ )  $\delta$  12.17 (s, 1 H), 8.61 (d, J = 9.2 Hz, 1 H), 8.31 (s, 1 H), 8.23 (d, J =7.6 Hz, 1 H), 8.12 (d, J = 2.0 Hz, 1 H), 7.99 (d, J = 7.6 Hz, 1 H), 7.87 (dd, J = 9.2, 2.5Hz, 1 H), 7.77-7.82 (m, 3 H), 7.73 (t, J = 7.4 Hz, 1 H), 7.61 (t, J = 7.6 Hz, 2 H).

5

10

15

20

# 5-Bromo-2-({3-[hydroxy(phenyl)methyl]benzoyl}amino)benzoic acid

Solid sodium borohydride (82 mg, 2.2 mmol) was added in one portion to a slurry of methyl 2-[(3-benzoylbenzoyl)amino]-5-bromobenzoate (826 mg, 1.88 mmol) in 40 mL of 1:1 methanol/THF. The mixture was stirred for 75 minutes before being quenched by the addition of 1 M aqueous HCl (50 mL). The organics were removed by rotary evaporation, and the product was extracted into 100 mL + 50 mL of CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from CH<sub>2</sub>Cl<sub>2</sub> to 5% EtOAc/CH<sub>2</sub>Cl<sub>2</sub> as eluent. Yield was 433 mg of white solid as the methyl ester. To a mixture of the corresponding methyl ester (348 mg, 0.788 mmol) in dioxane (20 mL) was added 1 M aqueous sodium hydroxide (1.5 mL). The mixture was stirred at room temperature overnight and then heated in a 50 °C oil bath for 30 minutes. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The organics were washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. They were then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol (10 mL). The solids were washed with ethanol followed by pentane and then dried at 100 °C under vacuum yielding 130 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.12 (s, 1 H), 8.66 (d, J = 8.7 Hz, 1 H), 8.13 (d, J = 2.5 Hz, 1 H), 8.05 (s, 1 H), 7.85 (dd, J = 9.2, 2.5 Hz, 1 H), 7.79 (d, J = 7.6 Hz, 1 H)1 H), 7.62 (d, J = 8.1 Hz, 1 H), 7.51 (t, J = 7.6 Hz, 1 H), 7.42 (d, J = 7.1 Hz, 2 H), 7.32 (t, J = 7.6 Hz, 2 H), 7.22 (t, J = 7.1 Hz, 1 H), 6.07 (br s, 1 H), 5.81 (s, 1 H).

# 5-Bromo-2-({3-[(methoxyimino)(phenyl)methyl]benzoyl}amino)benzoic acid (PHA-522146)

10

15

20

Methyl 2-[(3-benzoylbenzoyl)amino]-5-bromobenzoate (763 mg, 1.74 mmol) was dissolved in 60 mL of 1:1 EtOH/pyridine with warming. After this solution was allowed to cool, solid O-methylhydroxylamine hydrochloride (350 mg, 4.19 mmol, Aldrich) was added in one portion. The resulting slurry was stirred at room temperature for 6 days, after which it was a solution. The solvents were removed by rotary evaporation, and the residue was dissolved in 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was dried over MgSO<sub>4</sub> and evaporated leaving 785 mg of white solid that was approximately a 1:1 mixture of oxime isomers by 1H NMR. To a mixture of the corresponding methyl ester (470 mg, 1.01 mmol) in dioxane (15 mL) was added 1 M aqueous sodium hydroxide (2.0 mL). The mixture was stirred at room temperature overnight. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The organics were washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. They were then dried over MgSO<sub>4</sub> and evaporated. The orange residue was recrystallized from hot ethanol (10 mL). The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 255 mg of white solid that was approximately a 1:1 mixture of oxime isomers by 1H NMR. Due to the presence of 2 isomers, the NMR is difficult to assign. At 400 MHz in DMSO- $d_6$ , the amide protons appear as singlets at 12.10 and 12.07 ppm. The aromatic protons appear between 7.32 and 8.63 ppm. The methyl peaks appear as singlets at 3.93 and 3.92 ppm.

25

#### 5-cyano-2-{[3-(cyclopentylcarbonyl)benzoyl]amino}benzoic acid

tert-Butyl 5-cyano-2-[(3-iodobenzoyl)amino]benzoate (1.0 g, 2.23 mmol) was dissolved in 20 ml of  $CH_2Cl_2$ . Hexamethylditin (1.1 g, 3.35 mmol) and allylpalladium chloride dimer (73 mg, 0.2 mmol) were then added and the mixture stirred at room temperature for 5 hr. The reaction was diluted with  $CH_2Cl_2$  then washed with water.

The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The 5 remaining oil was purified via silica gel chromatography to give 670 mg (62%) of the desired tin compound. This product was subsequently dissolved in 15 mL of THF. To this was added DIPEA (1 mL), Pd<sub>2</sub>dba<sub>3</sub> (115 mg, .125 mmol) and cyclopentanecarbonyl chloride (230 mg 1.73 mmol). The reaction was then warmed to 60 °C and stirred for 10 additional hr. After cooling to room temperature the 10 reaction was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The remaining residue was purified via silica gel chromatography, giving 415 mg (72%) of the desired ketone. The ketone was treated with CH<sub>2</sub>Cl<sub>2</sub>/TFA and stirred for 10 additional hours. The solvent was removed in vacuo and the remaining solid was 15 recrystalized from MeOH to give the title compound (329 mg, 91%) as a white solid. 1H NMR (400 MHz, DMSO) 1.62-1.67 (m, 4H), 1.73-1.80 (m, 2H), 1.92-1.98 (m, 2H), 3.90 (quint, 1H), 7.77 (t, 1H), 8.11 (dd, 1H), 8.19 (d, 1H), 8.27 (d, 1H), 8.41 (d, 1H), 8.53 (s, 1H), 8.84 (d, 1H), 12.55 (s, 1H)

20

Other compounds produced via the above-described methodology using appropriate starting materials and maiking non-critical variations include:

- 2-[(3-Benzoylbenzoyl)amino]-5-chlorobenzoic acid
- 2-[(4-Acetylbenzoyl)amino]-5-bromobenzoic acid
- 25 2-[(4-Benzoylbenzoyl)amino]-5-bromobenzoic acid
  - 2-[(3-Acetylbenzoyl)amino]-5-bromobenzoic acid
  - 5-Bromo-2-({3-[(hydroxyimino)(phenyl)methyl]benzoyl}amino)benzoic acid
  - (+)-5-Bromo-2-({3-[hydroxy(phenyl)methyl]benzoyl}amino)benzoic acid
  - (-)-5-bromo-2-({3-[hydroxy(phenyl)methyl]benzoyl}amino)benzoic acid
- 30 2-[(3-Benzoylbenzoyl)amino]-5-nitrobenzoic acid
  - 2-[(3-Benzoylbenzoyl)amino]-5-cyanobenzoic acid
  - 5-Cyano-2-({3-[(hydroxyimino)(phenyl)methyl]benzoyl}amino)benzoic acid
  - 5-Cyano-2-({3-[(methoxyimino)(phenyl)methyl]benzoyl}amino)benzoic acid

#### Solid Phase Synthesis

Additional methodologies for producing compounds of this invention are shown below.

#### **5** Scheme 3.1

10

 $R_3$  is a  $C_{1-4}$ alkyl optionally substituted with 1-3 halo, -OH, NO<sub>2</sub>, or -CN.

Development of a solid phase route to ketones 1 was effected by a similar route and is summarized in Scheme 3.2. Chlorine was selected as the anthranilic acid 5-substituent instead of the 5-bromine of the ketone leads in order to avoid the potential for competing reactions in the ensuing palladium-catalyzed stannylation. Solid-supported aryl halide 8 was prepared by reaction of chloroisatoic anhydride with Wang resin. Coupling with halo (X = Br or I) aroyl chlorides then afforded

benzamides 9, which were stannylated with hexamethyl distannane under the influence of palladium catalyst using the same conditions that were applied in Scheme 3.1. The subsequent carbonylation reactions were found to be optimal using the slightly modified conditions of Ellman. Eliminating the ligand altogether and adding potassium carbonate as another proton scavenger slightly enhanced the rate of the reactions and the product purities in the end. Carbon monoxide was not necessary to eliminate aryl-aryl coupling by-products. One other modification in the synthetic conditions was to decrease the amount of TFA used in the cleavage cocktail in order to avoid trace amounts of a cleavage impurity.

10

5

#### Scheme 3.2

Generation of Oximes and Amines from Solid-Supported Ketones

15

20

Chemistry was developed for amine (12) and oxime (13) derivatization of the ketones on solid-phase (Scheme 3.3). Following TFA cleavage (Scheme 3.4), the amines could be successfully purified by trapping the products on sulfonic acid resin and then washing off with 2 N NH<sub>4</sub>OH/methanol. The remaining compounds were subjected to preparative HPLC purification.

#### Scheme 3.3

1) 
$$R^3NH_2$$
  
Ti (OiPr)<sub>4</sub>  
Toluene, RT

2)  $Na(OAc)_3BH$   
 $AcOH/THF$ , RT,  $N_2$ 

11

13

 $R^3 = H$ , alkyl

#### Scheme 3.4

#### **Ketones**

5

10

15

20

#### Step 1: Preparation of 8

To 4.5 grams of Wang resin (Irori Unisphere, 1.36 mmol/g loading, 6.12 mmol) in a 125 mL serum bottle, 60 mL of DMF were added followed by 6.1 grams (5 eq., 30.6 mmol) of 5-chloroisotoic anhydride and 3.74 grams (5 eq., 30.6 mmol) of 4-dimethylaminopyridine. The serum bottle was purged with nitrogen, capped, and shaken on an orbital mixer at 60 °C. Initially, the reagent cocktail was not homogeneous, but after several hours, a concentrated solution had formed around the swelled resin. After 18 hours, the reaction slurry was cooled and transferred to a 60 mL syringe-barrel reaction vessel. The reagent cocktail was then drained and the resin washed as follows: 3 X (acetonitrile, DMF), then 3 X (acetonitrile, methylene chloride). The resin was treated a second time with 60 mL of DMF, 6.1 grams (5 eq., 30.6 mmol) of 5-chloroisotoic anhydride, and 3.74 grams (5 eq., 30.6 mmol) of 4-dimethylaminopyridine. Following mixing at 60 °C for 6 hours, the reagent cocktail was again drained and the resin washed as above. In a vacuum oven at 25 °C, the resin was dried for 72 hours to give a final weight of 5.36 grams (1.14 mmol/g loading).

#### Step 2: Preparation of 9

To 6.7 mmol of the halo benzoic acids suspended in 20 mL of methylene chloride, 20 μL of DMF and 1.17 mL (1.7 grams, 13.4 mmol, 2 eq.) of oxalyl chloride were added. The flasks were sealed and stirred with occasional release of gas build-up. After 5 stirring overnight, the reaction mixtures had become almost completely homogeneous with no more gas build-up. Solvent and excess oxalyl chloride were then evaporated in vacuo to dryness. The acid chlorides were re-dissolved in 10 mL of methylene chloride and added to 1 gram of resin 8 (1.14 mmol/gram loading, 1.14 mmol) swollen with 10 mL of pyridine in 25 mL vials. Some fuming was observed initially. 10 The mixtures were purged with nitrogen for 10 seconds then the vials capped, and the mixtures shaken at room temperature for 4 hours. By that time, the resins had taken on a light orange color and a tan precipitate had formed in the supernatant. The reagent solutions were then drained in syringe-barrel reaction vessels and the resins rinsed five times with alternating acetonitrile and methylene chloride washes. The resins were kept wet with methylene chloride until used in the next step. Cleavage aliquots (40 % TFA/CH<sub>2</sub>Cl<sub>2</sub>) had purities of > 80% by HPLC and were registered as PHA compounds (Table 1).

#### Step 3: Preparation of 10 20

15

25

30

A stock solution of palladium acetate (0.1 eq., 0.01 mmol, 0.0022 g per 1 mL), triphenylphosphine (0.25 eq., 0.025 mmol, 0.0065 g per 1 mL), and diisopropylethylamine (0.5 eq., 0.05 mmol, 0.0065 g, 0.0087 mL per 1 mL) in 6.5 mL DMF (degassed with N<sub>2</sub>) was prepared. To each of the resins (9) in 8 mL vials, 1 mL of stock catalyst solution was added, followed by 0.042 mL of hexamethyl ditin (2.0 eq., 0.2 mmol, 0.065 g). Each vial was purged with nitrogen and then capped. The reaction mixtures were then heated to 60 °C and mixed in an orbital shaker for 17 h. By that time, the resins had all turned black in color. Following cooling, the reaction mixtures were transferred to filter vessels, and reagents were drained. This was followed by washing three times with DMF, three times with alternating acetonitrile/DMF, three times with alternating acetonitrile/methylene chloride, and twice with THF. Cleavage aliquots were taken (cleaved in 40/60 TFA/CH<sub>2</sub>Cl<sub>2</sub>) to check for completion of reaction by monitoring the protodestannylation products.

#### Step 4: Preparation of 11

To each of the 8 mL vials holding resins 10, 2 mL of a THF (degassed with carbon monoxide) stock solution containing: 0.0046 g of tris (dibenzylidene acetone)

5 dipalladium (0) (0.05 eq., 0.005 mmol, per 2 mL THF); 0.0052 g of triphenylphosphine (0.2 eq, 0.02 mmol, per 2 mL THF); and 0.139 mL diisopropyl ethylamine (8 eq., 0.80 mmol, 0.103 g, 0.139 mL per 2 mL) were added.

Commercially available acid chlorides (8 eq., 0.8 mmol) were then added. The reaction vessels were purged with carbon monoxide, capped and shaken at 60 °C for 18 h. When cool, the reaction mixtures were filtered through fritted syringe barrels, then the resins rinsed six times with alternating acetonitrile/methylene chloride washes and dried under vacuum at room temperature.

#### Step 5: Preparation of 1

To each of the fritted vessels containing resins 11, 2 mL of the cleavage cocktail (40/60 TFA/CH<sub>2</sub>Cl<sub>2</sub>) were added and the mixtures swirled for 45 minutes. Cleavage filtrates were then collected in tared vials followed by stripping of solvents *in vacuo*. The residues were analyzed by HPLC and ESMS separately. The library was then purified by preparative HPLC. Results for the library both pre- and post-purification are compiled in Table 5.

#### **Preparation of Oximes 13**

Ketone precursors to the oxime derivatives were produced as shown above. To 0.1 gram (~0.12 mmol) of the ketone resins 11 in a 48 well Robbins Block, 2 mL of pyridine were added followed by 10 equivalents (1.2 mmol) of each alkoxyamine (hydroxylamine hydrochloride; methoxyamine hydrochloride; obenzylohydroxyamine hydrochloride; and o-allylhydroxylamine hydrochloride). The reaction block was sealed and mixed overnight at room temperature in the rotating oven. After 20 hours, the resins resins were drained and washed with 3 X (MeOH, CH<sub>2</sub>Cl<sub>2</sub>) and 3 X (MeCN, CH<sub>2</sub>Cl<sub>2</sub>). Methanol was used early in the wash cycle because MeCN and CH<sub>2</sub>Cl<sub>2</sub> left a precipitate in the supernatant at that point. Treatment of the resins with 40 % TFA/CH<sub>2</sub>Cl<sub>2</sub> for 45 minutes afforded crude

products. Four of the library compounds (shown in Table 3) were then successfully purified (>90 % pure) via LC/MS.

#### **Amine Derivatives**

10

15

20

25

30

### 5 Preparation of Amines 12

Into four 8 mL vials containing 0.1 grams (~0.12 mmols) of ketone resin 11, 1.5 mL of toluene along with 0.12 grams (0.42 mmol) of titanium isopropoxide and 2.5 equivalents (0.30 mmol) of each respective amine were added. The vials were purged with nitrogen, sealed with teflon-lined caps, and mixed at room temperature for 16 hours on an orbital shaker. At that time, 0.5 mL of THF, 0.1 mL of acetic acid, and 0.24 grams (1.14 mmol) of sodium triacetoxyborohydride were added and the slurry was mixed at room temperature. After 4 hours, the reagents were drained and the resin washed: 3 X (MeOH, DMF), 4X (MeOH, CH<sub>2</sub>Cl<sub>2</sub>). Treatment of the resin with 40 % TFA/CH<sub>2</sub>Cl<sub>2</sub> for 45 minutes afforded crude products in the purities included in Table 6. Crude product identities were confirmed by ES/MS.

#### Step 1: Preparation of 8

To 10.0 grams of Wang resin (Irori Unisphere, 1.36 mmol/g loading, 13.6 mmol) in a 250 mL serum bottle, 90 mL of DMF were added followed by 13.4 grams (5 eq., 68 mmol) of 5-chloroisotoic anhydride and 8.3 grams (5 eq., 68 mmol) of 4-dimethylaminopyridine. The serum bottle was purged with nitrogen, capped, and shaken on an orbital mixer at 60 °C. Initially, the reagent cocktail was not homogeneous, but after several hours, a concentrated solution had formed around the swelled resin. After 18 hours, the reaction slurry was cooled and transferred to a 60 mL syringe-barrel reaction vessel. The reagent cocktail was then drained and the resin washed as follows: 3 X (acetonitrile, DMF), then 3 X (acetonitrile, methylene chloride). The resin was treated a second time with 90 mL of DMF, 13.4 grams (5 eq., 68 mmol) of 5-chloroisotoic anhydride, and 13.4 grams (5 eq., 68 mmol) of 4-dimethylaminopyridine. Following mixing at 60°C for 6 hours, the reagent cocktail was again drained and the resin washed as above. In a vacuum oven at 25°C, the resin was dried for 72 hours to give a final weight of 10.46 grams (1.30 mmol/g loading).

#### Step 2: Preparation of 9 (R = H)

To 6.2 grams (25 mmol) of the meta- and para- iodo benzoic acids suspended in 70 mL of methylene chloride, 40 µL of DMF and 4.4 mL (6.35 grams, 50 mmol, 2 eq.) of oxalvl chloride were added. The serum bottles were sealed and stirred with occasional release of gas build-up. After stirring for 5 hours, the reaction mixtures had become almost completely homogeneous with no more gas build-up. Solvent and excess oxalyl chloride were then evaporated in vacuo to dryness. The acid chlorides were re-dissolved in 30 mL of methylene chloride and added to 4 gram of resin 8 (1.30 mmol/gram loading, 5.2 mmol) swollen with 30 mL of pyridine in 125 mL serum bottles. Some furning was observed initially. The mixtures were purged with nitrogen for 10 seconds then the vials capped, and the mixtures shaken at room temperature for 4 hours. By that time, the resins had taken on a light orange color and a tan precipitate had formed in the supernatant. The reagent solutions were then drained in syringe-barrel reaction vessels and the resins rinsed five times with alternating acetonitrile and methylene chloride washes. The resins were then dried in vacuo to afford 5.14 g of the meta-iodo product and 5.09 g of the para-iodo product. Cleavage aliquots were >95 % pure by HPLC, with their identities confirmed by ESMS.

#### Step 3: Preparation of 10 (R = H)

5

10

15

20

25

30

A stock solution of palladium acetate (0.012 M), triphenylphosphine (0.03 M), and diisopropylethylamine (0.06 M) in 80 mL DMF (degassed with N<sub>2</sub>) was prepared. To 4.0 grams (~5.0 mmol) of each resin (9) in 125 mL serum bottles, 40 mL of the stock catalyst solution were added, followed by 2.0 mL of hexamethyl ditin (2.0 eq., 9.6 mmol, 3.14 g). Each bottle was purged with nitrogen and then capped. The reaction mixtures were then heated to 60 °C and mixed in an orbital shaker for 17 h. By that time, the two resins had turned black in color. Following cooling, the reaction mixtures were transferred to filter vessels, and reagents were drained. This was followed by washing three times with DMF, three times with alternating acetonitrile/DMF, three times with alternating acetonitrile/methylene chloride, and twice with THF. Cleavage aliquots were taken (cleaved in 40/60 TFA/CH<sub>2</sub>Cl<sub>2</sub>) to check for completion of reaction by monitoring the protodestannylation products. Following cleavage, the meta-substituted resin gave 87 % of the expected destannylated product by HPLC, while the para-substituted isomer gave 70 %. Little

to no iodide starting material remained. The major impurity in both cases was an unidentified peak with  $[M+H]^+ = 369 \text{ m/z}$ .

#### **Step 4: Preparation of 11**

To each carboxylic acid weighed into a 20 mL vial (2.88 mmol), 6.5 mL of THF, 10 5 uL of DMF, and 0.293 ml of oxalyl chloride (0.95 eq., 2.7 mmol, 3.35 g) were added. The vials were sealed and reaction mixtures shaken at room temperature for 4 hours with occasional release of evolved gas. In the meantime, the two stannylated resins (10) were distributed into Irori minikans (60 mg per kan), and the 72 kans were then distributed into twelve 125 mL serum bottles (six kans per bottle). To each of the 10 bottles, 20 mL of a nitrogen degassed THF stock solution containing: tris (dibenzylidene acetone) dipalladium (0) (0.001 M); potassium carbonate (0.02 M); and diisopropyl ethylamine (0.10 M) were added. The THF solutions of acid chlorides (2.88 mmol, 6 eq.) were then added to their respective set of six bottles. The capped reaction vessels were purged with nitrogen, degassed, and shaken at 65 °C for 15 18 h. When cool, the resin containing kans were rinsed five times with alternating acetonitrile/ methylene chloride washes and dried under vacuum at room temperature. A cleavage aliquot revealed that ketone formation had gone to completion.

# 20 Step 5a: Preparation of Resin-Bound Amines 12

To a 125 mL serum bottle containing 24 Irori cans loaded with resin 11, 30 mL of toluene were added, followed by 1.23 grams (6.0 mmol, 3.5 eq.) of titanium isopropoxide and 0.25 grams (4.3 mmol, 2.5 eq.) of propyl amine. The bottle was degassed to remove air bubbles from the Irori kans, then purged with nitrogen, sealed and mixed for 17 hours at room temperature. At that time, 10 mL of toluene, 2 mL of acetic acid, and 3.5 grams (16.3 mmol, 9.5 eq.) of sodium triacetoxy borohydride were added, and bottle re-purged and sealed, and mixed for 14 hours. Reagents were then drained and the resins washed three times with methanol and five times with alternating methanol/methylene chloride.

30

25

# **Step 5b: Preparation of Resin-Bound Oximes 13**

To a 125 mL serum bottle containing 24 Irori kans loaded with resin 11, 40 mL of pyridine were added followed by 1.2 grams (17.2 mmols, 10 eq.) of hydroxylamine

hydro chloride. The bottle was degassed to remove air bubbles from the Irori kans, then purged with nitrogen, sealed and mixed for 17 hours at room temperature. At that time, reagents were drained and the resins were washed three times with methanol, and five times with alternating methanol/methylene chloride.

5

10

15

20

25

30

#### Step 6: Preparation of 15

The 72 kans containing resins 12,13 were distributed into tared 8 mL vials and treated with 3 mL of TFA/CH<sub>2</sub>Cl<sub>2</sub> (40/60). The vials were degassed, capped, and mixed at room temperature for 1.5 hours. The kans were then plucked out of the vials using a syringe needle and washed with another 1 mL of CH<sub>2</sub>Cl<sub>2</sub>. Solvent in the vials was evaporated *in vacuo* (Genevac), leaving product residue.

#### Preparation of 5-iodoisatoic anhydride

To a red-brown solution of 2-amino-5-iodobenzoic acid (25 grams, 95 mmol) in 300 mL of dioxane, 9.58 grams (32.3 mmol) of triphosgene were carefully added. The resulting slurry was refluxed for 4 hours. By that time, all starting material had disappeared by HPLC. The solid product was then filtered, washed once with ethyl ether, then dried overnight in a vacuum oven at 40 °C. The tan colored needles amounted to 22.9 grams (83 %). HPLC (MRH1 method):  $t_R = 2.15$  min. (100 %); <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.12 (s, 1 H), 8.00 (d, J = 8.6 Hz, 1 H), 6.95 (d, J = 8.5 Hz, 1 H); MS (ES) m/z (rel. intensity) 288 (M-, 100), 244 (5), 289 (5); 577 (10).

#### Preparation of 6-Chloroindoline

In a 250 mL round bottom flask, 12.4 grams of sodium cyanoborohydride (198 mmol, 2 eq.) were added potion-wise over 5 minutes to a solution of 15 grams (98.9 mmol) of 6-chloroindole. After stirring for 22 hours, the mixture had become a brown solution and analysis by HPLC (MRH 1 method) revealed no starting material remaining and a mixture of two product peaks. The mixture was diluted with 100 ml of water, then made basic with  $\sim$ 200 mL of 6N sodium hydroxide. The desired product was extracted into 3 X 400 mL of methylene chloride. The extracts were then dried over anhydrous magnesium sulfate and evaporated *in vacuo* leaving a cloudy oil. The crude product was chromatographed over a plug of silica in 100 % methylene chloride giving a mixed fraction (Rf = 0.9 and 0.7), a pure product fraction (Rf = 0.7),

and a baseline fraction (Rf = 0.0 - 0.2). The pure fraction was evaporated to dryness *in vacuo* to yield a clear, colorless oil weighing 10.90 grams (72 %). It was stored at 4°C and saved for future use. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  6.95 (d, J = 5 Hz, 1 H), 6.46 (d, J = 5 Hz, 2 H), 3.43 (t, J = 6, 2 H), 2.86 (t, J = 6, 2 H).

5

#### **Example 4: AMIDE DERIVATIVES**

Standard procedure for attaching 5-bromoanthranile acid to hydroxymethyl styrene resin,:

15

10

To a slurry of 24.8 g (36.7 mmol) hydroxymethyl styrene resin in 1 L DMF was added 24 g (197 mmol) 4-dimethylamino pyridine and 50 g (207 mmol) 5-

bromoisatoicanhydride. The mixture was stirred at 60 °C for 18 hours and room temperature for four hours. The mixture was then filtered and the resin washed repeatedly alternating with dichloromethane and DMF (3x) then repeatedly alternating with dichloromethane and methanol (3x) followed by methanol (3x). The resin was dried over night in a vacuum oven.

25

20

#### Resin 2 and 3:

Standard procedure for attaching 3 or 4- N-boc-amino benzoic acid to resin 1.

To 5.1 g (21.5 mmol) 3-N-boc-aminobenzoic acid in 200 mL of anhydrous THF was added 100 µL DMF and 2.3 mL (25.8 mmol) oxalyl chloride in five portions over 20 minutes. After 40 minutes the mixture was concentrated in vacuo and then dissolved in 50 mL dichloromethane. This was added to a slurry of 3.79 g (4.32 mmol) resin 1 in 150 mL dichloromethane and 3.7 mL diisopropylethyl amine. The mixture was heated to reflux over night. The resin was then collected by vacuum filtration and washed repeatedly alternating with dichloromethane and methanol (4x) followed by methanol (3x) and dried in a vacuum oven. The same procedure was followed to prepare resin 3 from 4-N-boc-aminobenzoic acid.

Standard procedure for the acylation of resins 2 and 3 with acid chlorides, isocyanates, and isothiocyanates.

15

20

5

10

On average 55 mg (Ca. 0.055 mmol) resin was treated with 33% TFA in DCM for two hours. The resin was collected by filtration and washed repeatedly alternating with dichloromethane and methanol (4x) followed by methanol (3x) and dried in a vacuum oven. The resin is then treated with 0.6 mmol of the acylating reagent and 0.86 mmol diisopropylethyl amine in DCM and shaken over night. The resin was then collected by vacuum filtration and washed repeatedly alternating with dichloromethane and methanol (4x) followed by methanol (3x) and dried in a vacuum oven

Standard procedure for the acylation of resins 2 and 3 with sulfonyl chlorides:

10

On average, to 60 mg (Ca. 0.06 mmol) resin in 2 mL DCM was added 10 equivalents of a sulfonyl chloride and 174 µL (0.6 mmol) 2-tert-butylimino-2-diethyl-amino-1,3-dimethylperhydro-1,3,2-diazaphosphorine (BEMP). After mixing overnight, the resin was collected by vacuum filtration and washed repeatedly alternating with dichloromethane and methanol (4x) followed by methanol (3x) and dried in a vacuum oven. The resin was then treated with 2 mL of 40 % TFA in DCM for one hour and then collected by vacuum filtration and washed repeatedly alternating with dichloromethane and methanol (4x) followed by methanol (3x) and dried in a vacuum oven.

Standard cleavage procedure to provide products.

The resin was treated with 1.5 mL THF and 0.5 mL 1 N sodium hydroxide over night.

The mixtures were filtered and the collected filtrate was treated with 250 mg of IR120 acidic resin for 2.5 hours. The mixtures were filtered and the filtrates
concentrated to provide the following products. If initial purity was less than 80 % by
HPLC those products were purified by chromatography.

Several compounds were produced by the above-described methodologies.

2-{[3-(benzoylamino)benzoyl]amino}-5-bromobenzoic acid

5-bromo-2-{[3-(2-furoylamino)benzoyl]amino}benzoic acid

5-bromo-2-({3-[(thien-2-ylacetyl)amino]benzoyl}amino)benzoic acid

```
5-bromo-2-({3-[(mesitylcarbonyl)amino]benzoyl}amino)benzoic acid
     5-bromo-2-({4-[(mesitylcarbonyl)amino]benzoyl}amino)benzoic acid
     2-({3-[(1,3-benzodioxol-5-ylcarbonyl)amino]benzoyl}amino)-5-bromobenzoic acid
     5-bromo-2-({3-[(2,4-dimethoxybenzoyl)amino]benzoyl}amino)benzoic acid
     5-bromo-2-[(3-{[(phenylthio)acetyl]amino}benzoyl)amino]benzoic acid
 5
     5-bromo-2-({3-[(methoxyacetyl)amino]benzoyl}amino)benzoic acid
     2-({3-[(anilinocarbonyl)amino]benzoyl}amino)-5-bromobenzoic acid
      5-bromo-2-{[3-({[(2,4-
     difluorophenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid
     5-bromo-2-{[3-({[(3-cyanophenyl)amino]carbonyl}amino)benzoyl]amino}benzoic
10
     acid
     5-bromo-2-{[3-({[(3-chlorophenyl)amino]carbonyl}amino)benzoyl]amino}benzoic
     acid
     5-bromo-2-({3-[({[3-
15
     (methylthio)phenyl]amino}carbonyl)amino]benzoyl}amino)benzoic acid
     2-{[3-({[(3-acetylphenyl)amino]carbonyl}amino)benzoyl]amino}-5-bromobenzoic
     acid
     5-bromo-2-({4-[(phenylsulfonyl)amino]benzoyl}amino)benzoic acid
     5-bromo-2-{[3-({[4-
     (trifluoromethoxy)phenyl]sulfonyl}amino)benzoyl]amino}benzoic acid
20
     5-bromo-2-{[4-({[4-
     (trifluoromethoxy)phenyl]sulfonyl}amino)benzoyl]amino}benzoic acid
     5-bromo-2-[(4-{[(3,4-dichlorophenyl)sulfonyl]amino}benzoyl)amino]benzoic acid
     5-bromo-2-({4-[(thien-2-ylacetyl)amino]benzoyl}amino)benzoic acid
     5-bromo-2-({3-[(5-nitro-2-furoyl)amino]benzoyl}amino)benzoic acid
25
     5-bromo-2-({4-[(5-nitro-2-furoyl)amino]benzoyl}amino)benzoic acid
     5-bromo-2-{[4-({[(2,4-
     difluorophenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid
     5-bromo-2-{[3-({[(3,5-
     dichlorophenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid
30
     5-bromo-2-{[3-({[(5-chloro-2-
     methoxyphenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid
```

5-bromo-2-{[3-({[(4-phenoxyphenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid

5-bromo-2-{[4-({[(4-phenoxyphenyl)amino]carbonyl}amino)benzoyl]amino}benzoic acid

5 2-{[3-({[(4-acetylphenyl)amino]carbonyl}amino)benzoyl]amino}-5-bromobenzoic acid

5-bromo-2-{[4-({[(4-

nitrophenyl)amino]carbonothioyl}amino)benzoyl]amino}benzoic acid 5-bromo-2-({3-[({[2-

10 (trifluoromethyl)phenyl]amino}carbonothioyl)amino]benzoyl}amino)benzoic acid 5-bromo-2-{[3-({[(3,4,5-

trimethoxyphenyl)amino]carbonothioyl}amino)benzoyl]amino}benzoic acid 5-bromo-2-({3-[({[3-

(methylthio)phenyl]amino}carbonothioyl)amino]benzoyl}amino)benzoic acid

2-{[3-({[(3-acetylphenyl)amino]carbonothioyl}amino)benzoyl]amino}-5-bromobenzoic acid

5-bromo-2-({3-[(phenylsulfonyl)amino]benzoyl}amino)benzoic acid

5-bromo-2-[(3-{[(3,4-dichlorophenyl)sulfonyl]amino}benzoyl)amino]benzoic acid

 $5\text{-}bromo-2\text{-}[(4\text{-}\{[(4\text{-}methylphenyl)sulfonyl]amino}\}benzoyl)amino]benzoic acid$ 

Analogs with an alternative linkage, such as ureas, in place of the sulfonamides described in Example 1 were also synthesized.

#### Scheme 4.1

25

20

### Methyl 4-[(2,3-dihydro-1H-indol-1-ylcarbonyl)amino]benzoate

Methyl-4-aminobenzoate (1.00 g, 7.29 mmol) in DCM (50 mL) was slowly added to a solution of phosgene (1.93 M /toluene, 7.5 mL, 14.5 mol, 2.0 equiv) in DCM (200 mL) at 0°C, follwed by the addition of diisopropylethyl amine (1.14 mL, 6.56 mmol, 0.9 equiv). The mixture was allowed to warm to rt, then stirred for 1 h, and then concentrated in vacuo to ca 5 mL. The suspension was redissolved in DCM followed by the addition of indoline (2.45 mL, 21.87 mmol, 3.0 equiv) and diisopropylethyl amine (1.14 mL, 6.56 mmol, 0.9 equiv). The resulting mixture was stirred for 2h, at rt, then washed with 1N HCl, brine, dried (MgSO<sub>4</sub>) filtered and concentrated in vacuo. The residue was recrystallized from EtOH to afford 1.67 g of **5.7** as a white solid.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.04-8.01 (m, 2 H), 7.90 (d, J = 7.9 Hz, 1 H), 7.58-7.55 (m, 2 H), 7.28-7.20 (m, 2 H), 7.01 (t, J = 8.2 Hz, 1 H), 6.70 (s, 1 H), 4.12 (t, J = 8.3

15

5

10

#### 4-[(2,3-Dihydro-1H-indol-1-ylcarbonyl)amino]benzoic acid

Hz, 2 H), 3.91 (s, 3 H), 3.26 (t, J = 8.2 Hz, 2 H).

Methyl 4-[(2,3-dihydro-1H-indol-1-ylcarbonyl)amino]benzoate (1.30 g, 4.37 mmol) was placed in dioxane (50 mL) with 5 N NaOH (10 mL) and the resulting solution was heated at 70 °C for 7h. The reaction was cooled to rt, acidified, diluted with EtOAc and washed with H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>) filtered and concentrated in vacuo. The residue was recrystallized from EtOH to afford 776 mg (63%) of a white solid.

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ) δ 8.82 (s, 1 H), 7.87 (d, J = 8.6 Hz, 3 H), 7.71 (d, J = 8.7 Hz, 2 H), 7.22-7.14 (m, 2 H), 6.92 (t, J = 7.3 Hz, 1 H), 4.16 (t, J = 8.4 Hz, 2 H), 3.18 (t, J = 8.5 Hz, 2 H).

10

15

20

5

# 5-Bromo-2-({4-[(2,3-dihydro-1H-indol-1-ylcarbonyl)amino]benzoyl}amino)benzoic acid, PNU-290877

4-[(2,3-dihydro-1H-indol-1-ylcarbonyl)amino]benzoic acid (627 mg, 2.22 mmol) was dissolved in DCM (30 mL) follwed by the addition of oxalyl chloride (490  $\mu$ L, 5.55 mmol, 2.5 equiv) and DMF (30  $\mu$ L). The mixture was stirred for 1h, then diluted with heptane (10 mL), concentrated in vacuo to dryness. The residue was redissolved in DCM (50 mL) followed by the addition of methyl-2-amino-5-bromo benzoate (510 mg, 2.2 mmol, 1.0 equiv.) and pyridine (360  $\mu$ L, 4.4 mmol, 2.0 equiv.) The reaction was stirred for 3 h at rt, then washed with 1 N HCl, 1 N NaOH, H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>) filtered and concentrated in vacuo. The residue was purified by silica gel chromatography (heptane/ EtOAc 19/1, 9/1, 4/1, 1/1, 0/1) to afford 198 mg (18%) of a white solid as the methyl ester. The ester (177 mg, 0.35 mmol) was dissolved in dioxane (10 mL) follwed by the addition of 5 N NaOH (5 mL). The reaction was

10

15

20

25

30

stirred for 3h at rt, diluted with EtOAc, washed with 1 N HCl, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was recrystallized from EtOH to afford 76 mg (44%) of a white solid.

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ) δ 8.88 (s, 1 H), 8.69 (d, J = 9.0 Hz, 1 H), 8.13 (d, J = 2.4 Hz, 1 H), 7.86-7.78 (m, 6 H), 7.21-7.14 (m, 2 H), 6.93 (t, J = 8.6 Hz, 1 H), 4.17 (t, J = 8.2 Hz, 2 H), 3.19 (t, J = 8.2 Hz, 1 H).

#### **Example 5: ALKYL DERIVATIVES**

#### Preparation of 3-(Phenylethynyl)benzoic acid

A flask containing ethyl 3-iodobenzoate (2.21g, 8.00 mmol, Lancaster), copper (I) iodide (550 mg, 2.88 mmol, Alfa), and tetrabutylammonium iodide (5.9 g, 16 mmol, Aldrich) was placed under argon. DMF (40 mL), diisopropylethylamine (4.5 mL, 26 mmol, Aldrich), and tri-t-butylphosphine (1.8 g of 10 wt% solution in hexane, 0.89 mmol, Strem) were added by syringe. Tris(dibenzylideneacetone)dipalladium(0)chloroform adduct (220 mg, 0.21 mmol, Aldrich) was added as a solid under a flow or argon. The mixture was stirred for 5 minutes, and phenylacetylene (0.88 mL, 8.0 mmol, Lancaster) was added by syringe. After 40 minutes, the mixture was added to a separatory funnel with 200 mL of saturated aqueous NaHCO<sub>3</sub>. Product was extracted into 3 X 100 mL of EtOAc. The combined EtOAc was washed with 4 X 200 mL of water and then dried over MgSO<sub>4</sub>. Product was adsorbed onto silica and purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 25% - 40% CH<sub>2</sub>Cl<sub>2</sub> in heptane. The ethyl 3-(phenylethynyl)benzoate was isolated as 1.82 g of brown oil that was contaminated with tri-t-butylphosphine. 990 mg of this oil was dissolved in dioxane (15 mL) and treated with 1 M aqueous sodium hydroxide (6 mL), and the mixture was stirred for 3.5 hours. It was then added to a separatory funnel with 100 mL of 1 M aqueous HCl and 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. A few milliliters of THF were added to help with solubility. The organics were washed with an additional 100 mL of HCl followed by 100 mL of water and then dried over MgSO<sub>4</sub>. Solvent was removed leaving 782 mg of tan solid that was still contaminated with phosphine.

15

20

25

Most of this material was carried on without further purification. For the purposes of characterization, the remainder was recrystallized from ethanol/heptane yielding a white solid.

#### 5 5-Bromo-2-{[3-(phenylethynyl)benzoyl]amino}benzoic acid

To 3-(phenylethynyl)benzoic acid (569 mg, 2.56 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added DMF (40  $\mu$ L) and oxalyl chloride (450  $\mu$ L, 5.16 mmol). The mixture was stirred for 2.5 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (15 mL), and methyl 2-amino-5bromobenzoate (504 mg, 2.19 mmol, Avocado) in pyridine (6 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 50% - 60% CH<sub>2</sub>Cl<sub>2</sub> in heptane as eluent. Yield was 694 mg of white solid as the methyl ester. To a mixture of the methyl ester (485 mg, 1.12 mmol) in dioxane (15 mL) was added 1 M aqueous sodium hydroxide (2.2 mL). The mixture was stirred for 2.75 hours. The reaction mixture was added to a separatory funnel with 100 mL of 1 M aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. It was then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol/THF. The solids were washed with ethanol followed by heptane and then dried at 100 °C under vacuum yielding 295 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.06 (s, 1 H), 8.60 (d, J = 9.2 Hz, 1 H), 8.12 (d, J = 2.0 Hz, 1 H), 8.10 (s, 1 H), 7.97 (d, J = 7.6 Hz, 1 H), 7.87 (dd, J = 9.2, 2.5 Hz, 1 H), 7.83 (d, J = 8.1 Hz, 1 H), 7.66 (t, J = 7.6 Hz, 1 H), 7.59-7.63 (m, 2 H), 7.45-7.48 (m, 3 H).

#### Preparation of 3-(2-Phenylethyl)benzoic acid

A mixture of 3-(phenylethynyl)benzoic acid (418 mg, 1.88 mmol) and palladium on carbon (315 mg, 10%, Aldrich) in 1:1 methanol/THF (20 mL) was stirred under 1 ATM of hydrogen overnight. The mixture was then filtered through a plug of celite and concentrated yielding 406 mg of white solid. This material was carried forward without further purification. For the purposes of characterization, a small amount of the product was recrystallized from toluene.

10

15

20

25

5

### 5-Bromo-2-{[3-(2-phenylethyl)benzoyl]amino}benzoic acid

To 3-(2-phenylethyl)benzoic acid (292 mg, 1.29 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added DMF (20 μL) and oxalyl chloride (225 μL, 2.58 mmol). The mixture was stirred for 2.5 hours, and the solvent and excess oxalyl chloride were removed by rotary evaporation. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (10 mL), and methyl 2-amino-5-bromobenzoate (248 mg, 1.08 mmol, Avocado) in pyridine (4 mL) was added. The mixture was stirred overnight and then added to a separatory funnel with 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. This solution was washed with 2 X 100 mL of 1 M aqueous HCl and 100 mL of brine. The CH<sub>2</sub>Cl<sub>2</sub> was evaporated in the presence of silica gel, and the product was purified by chromatography using a Biotage Flash 40 M silica cartridge with a gradient from 50% - 100% CH<sub>2</sub>Cl<sub>2</sub> in heptane as eluent. Yield was 361 mg of white solid as the methyl ester. To a mixture of the methyl ester (285 mg, 0.65 mmol) in dioxane (10 mL) was added 1 M aqueous sodium hydroxide (1.0 mL). The mixture was stirred at room temperature for 1 hour and then heated in a 50 °C oil bath for 15 minutes. The reaction mixture was added to a separatory funnel with 100 mL of 1 M

aqueous HCl, and the product was extracted into 100 mL of CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> was washed with an additional 100 mL of 1 M aqueous HCl followed by 100 mL of water. It was then dried over MgSO<sub>4</sub> and evaporated. The residue was recrystallized from hot ethanol. The solids were washed with heptane and then dried at 100 °C under vacuum yielding 88 mg of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.10 (s, 1 H), 8.68 (d, J= 9.1 Hz, 1 H), 8.12 (d, J= 2.5 Hz, 1 H), 7.83-7.87 (m, 2 H), 7.75-7.78 (m, 1 H), 7.46-7.51 (m, 2 H), 7.16-7.31 (m, 5 H), 2.91-3.02 (m, 4 H).

#### Example 6:

#### Thioamide linkers.

15

5

10

# 2-[({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}carbonothioyl)amino]-5-cyanobenzoic acid

General procedure A: Methyl 2-[({3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}carbonyl) amino]-5-cyanobenzoate (989 mg, 1.99 mmol) and Lawesson's reagent (4.5 g, 11.1 mmol) were combined in a flask equipped with a reflux condensor. The flask was evacuated and purged with N<sub>2</sub> several times. Tol (30 mL) was added and the reaction was refluxed overnight. The reaction was cooled to rt and filtered to remove excess Lawesson's reagent. The filtrate was absorbed in SiO<sub>2</sub>

10

15

20.

25

and the product was purified by silica gel chromatography using Hept/EtOac (19:1, 9:1, 3:17, 4:1). The product was triturated with MeOH to afford 670 mg (66%) of an orange solid as the methyl ester. <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  12.40 (s, 1 H), 8.35 (d, J = 2 Hz, 1 H), 8.29 (s, 1 H), 8.19 (dd, J = 8, 2 Hz, 1 H), 8.14 (d, J = 8 Hz, 1 H), 8.06 (d, J = 8 Hz, 1 H), 7.98 (d, J = 8 Hz, 1 H), 7.73 (t, J = 8 Hz, 1 H), 7.49 (d, J = 9 Hz, 1 H), 7.30-7.25 (m, 2 H), 4.02 (t, J = 8 Hz, 2 H), 3.79 (s, 3 H), 2.97 (t, J = 8 Hz, 2 H).

General procedure B: to a solution of the methyl ester (300 mg, 0.605 mmol) dissolved in THF (7 mL) and H<sub>2</sub>O (1.5 mL) was added LiOH-H<sub>2</sub>O (450 mg, 10.7 mmol) and the reaction was heated to 45°C for 6 hr. The solution was diluted with MTBE, washed with 2 N HCl and brine, dried (MgSO<sub>4</sub>), concentrated, and triturated with MeOH to afford 252 mg (84%) of an orange solid. <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  8.62 (d, J = 8 Hz, 1 H), 8.36 (dd, J = 12, 2 Hz, 1 H), 8.18 (d, J = 8 Hz, 1 H), 8.12 (dd, J = 8, 2 Hz, 1 H), 7.95 (d, J = 8 Hz, 1 H), 7.71 (t, J = 8 Hz, 1 H), 7.48 (d, J = 9 Hz, 1 H), 7.27-7.25 (m, 2 H), 4.02 (t, J = 8 Hz, 2 H), 2.96 (t, J = 8 Hz, 2 H).

#### Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate

To a suspension of 3-(chlorosulfonyl)benzoic acid (10.8 g, 49.0 mmol) in  $CH_2Cl_2$  (105 mL) and three drops of DMF was added oxalyl chloride (12.5 mL) and the reaction was stirred at rt overnight. The solution was concentrated *in vacuo*, diluted with  $CH_2Cl_2$  (100 mL), and the solution was divided into two reactions. A 50 mL (24.5 mmol) aliquot of the acid chloride was added to a solution of PHA-522499 (4.49 g, 25.5 mmol) dissolved in  $CH_2Cl_2$  (50 mL) and pyridine (3.0 mL) and stirred at rt overnight. The solution was diluted with MTBE, washed with 2 N HCl and brine, concentrated, triturated with MTBE to afford 7.91 g (85%) of methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate as a tan solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  11.73 (s, 1 H), 8.67 (d, J = 9 Hz, 1 H), 8.37 (d, J = 2 Hz, 1 H), 8.25 (s, 1

H), 8.12 (dd, J = 9, 2 Hz, 1 H), 7.92 (d, J = 8 Hz, 1 H), 7.88 (d, J = 8 Hz, 1 H), 7.60 (t, J = 8 Hz, 1 H), 3.93 (s, 3 H).

# 5 5-cyano-2-{[3-(pyrrolidin-1-ylsulfonyl)benzoyl]amino}benzoic acid

General procedure C: To a solution of methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (1.863 g, 4.92 mmol) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (40 mL) was added pyrrolidine (1.5 mL, 18.0 mmol) and stirred at rt for 3 hr. The reaction was diluted 10 with MTBE, washed with 2 N HCl and brine, concentrated, and triturated with MeOH to afford 1.70 g (84%) of methyl 5-cyano-2-{[3-(pyrrolidin-1vlsulfonvl)benzovl]amino}benzoate as a tan solid. <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 11.75 (s, 1 H), 8.61 (d, J = 9 Hz, 1 H), 8.38 (d, J = 2 Hz, 1 H), 8.33 (s, 1 H), 8.25 (d, J= 8 Hz, 1 H), 8.14 (dd, J = 9, 2 Hz, 1 H), 8.10 (d, J = 8 Hz, 1 H), 7.90 (t, J = 8 Hz, 1 H)15 H), 3.91 (s, 3 H), 3.24-3.19 (m, 4 H), 1.71-1.66 (m, 4 H). Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) was dissolved in 15 mL of CHCl<sub>3</sub>. Pyrrolidine (145 mg, 2.0 mmol) and Et<sub>3</sub>N (1 mL) were then added and the reaction stirred at room temperature for 12 hr. The mixture was poured 20 into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography, providing 297 mg (72%) of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (249 mg, 87%) was 25 obtained as a white solid after recrystalization from MeOH. H NMR (300 MHz, DMSO) 1.67 (m, 4H), 3.20 (m, 4H), 7.88 (t, 1H), 8.09-8.14 (m, 2H), 8.26 (d, 1H), 8.33 (s, 1H), 8.42 (d, 1H), 8.83 (d, 1H), 12.56 (s, 1H)

# 5-Cyano-2-({[3-(pyrrolidin-1-ylsulfonyl)phenyl]carbonothioyl}amino)benzoic acid

Prepared according to general procedure A: Methyl 5-cyano-2-{[3-(pyrrolidin-1ylsulfonyl)benzoyl]amino}benzoate (1.12 g, 2.70 mmol) and Lawesson's reagent (5.5 5 g, 13.6 mmol) afforded 450 mg of a mixture of the methyl ester and Lawesson's reagent after purifiying by silica gel chromatography twice. The crude material was hydrolyzed according to general method B to afford 253 mg (29%) over two steps of an orange solid. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  9.80 (d, J = 9 Hz, 1 H), 8.42 (d, J= 2 Hz, 1 H), 8.33 (s, 1 H), 8.23 (d, J = 8 Hz, 1 H), 7.97-7.91 (m, 2 H), 7.75 (t, J = 7Hz, 1 H), 3.23-3.19 (m, 4 H), 1.71-1.65 (m, 4 H).

#### 5-cyano-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino}benzoic acid

Methyl 2-{[3-(chlorosulfonyl)benzoyl]amino}-5-cyanobenzoate (378 mg, 1.0 mmol) 15 was dissolved in 15 mL of CHCl<sub>3</sub>. Morpholine (156 mg, 2.0 mmol) and Et<sub>3</sub>N (1 mL) were then added and the reaction stirred at room temperature for 12 hr. The mixture was poured into 1 M HCl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The resulting residue was purified by silica gel chromatography, providing 373 mg (87%) 20 of the desired methyl ester. The ester was treated with LiOH in 1:1:1 THF/MeOH/H<sub>2</sub>O for 12 hrs followed by acidification and extraction with EtOAc. The organic solution was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated in vacuo. The title compound (298 mg, 82%) was obtained as a white solid after recrystalization from

10

MeOH. H NMR (400 MHz, DMSO) 2.94 (m, 4H), 3.65 (m, 4H), 7.96 (t, 1H), 8.03 (d, 1H), 8.13 (dd, 1H), 8.27-8.31 (m, 2H), 8.42 (d, 1H), 8.82 (d, 1H), 12.55 (s, 1H)

## 5-Cyano-2-({[3-(morpholin-4-ylsulfonyl)phenyl]carbonothioyl}amino)benzoic acid

Prepared according to general method A and B: Methyl 5-cyano-2-{[3-(morpholin-4-ylsulfonyl)benzoyl]amino} benzoate (1.02 g, 2.38 mmol) and Lawesson's reagent (4.78 g, 11.8 mmol) afforded 532 g (50%) of the ester, 35527-bdw-118 as an orange solid. The ester (495 mg, 1.09 mmol) was hydrolyzed by general procedure B to afford 87 mg (20%) of an orange solid. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  9.72 (d, J = 8 Hz, 1 H), 8.41 (d, J = 2 Hz, 1 H), 8.27-8.25 (m, 2 H), 7.95 (dd, J = 9, 6 Hz, 1 H), 7.90 (d, J = 9 Hz, 1 H), 7.79 (t, J = 6 Hz, 1 H).

## 15 Example 7: X-Y Derivatives

## Scheme 7.1

## Scheme 7.2

HO pyridine-chlorochromate 
$$O_2N$$
  $O_2N$   $O$ 

## Scheme 7.3

5

## Scheme 7.4

## Methyl 2-(bromomethyl)-5-cyanobenzoate

Methyl 5-cyanobenzoate (4.50 g, 25.6 mmol), NBS (5.03 g, 28.25 mmol) and AIBN (150 mg) were dissolved in dichloroethane (160 mL). The mixture was irratiated with a photolamp for 2h. The mixture was cooled to rt and concentrated in vacuo. The residue was purified by silica gel chromatography (DCM/heptane 1/9,  $\frac{1}{4}$ , 1/1, 1/0) to afford 4.79 g (73%) of methyl 2-(bromomethyl)-5-cyanobenzoate.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.29 (d, J = 1.7 Hz, 1 H), 7.79 (dd, J = 8.0, 1.7 Hz, 1 H), 7.63 (d, J = 8.0 Hz, 1 H), 4.97 (s, 2 H), 4.00 (s, 3 H).

## Methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-cyanobenzoate

10

15

5

Methyl 2-(bromomethyl)-5-cyanobenzoate (2.80 g, 10.9 mol) was added to a solution of triphenylphosphine (2.87 g, 10.9 mmol) in toluene (50 mL). The resulting mixture was heated at reflux for 3h, cooled to rt, the precipiate was isolated by filtration, washed with pentane to afford 4.64 g (82%) of methyl 2-{[bromo(triphenyl) phosphoranyl]methyl}-5-cyanobenzoate as a white solid.  $^{1}$ H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.22 (s, 1 H), 8.08 (d, J = 7.9 Hz, 1 H), 8.79-7.51 (m, 16 H), 5.63 (d, J = 16.2 Hz, 2 H), 3.48 (s, 3 H).

#### Methyl 2-methyl-5-nitrobenzoate

nitrobenzoate as a white solid.

20

25

2-Methyl-5-nitrobenzoate (5.0 g, 27.6 mmol) was dissolved in MeOH (0.4 L) followed by the addition of  $\rm H_2SO_4$  (7 mL). The mixture was heated at reflux for 36 h, then cooled to rt and concentrated to ca 100 mL. The solution was diluted with MTBE neutralized with 6N NaOH, washed with 1N NaOH, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo to afford 4.72 g (87%) of methyl 2-methyl-5-

## Methyl 5-amino-2-methylbenzoate

Methyl 2-methyl-5-nitrobenzoate (5.0 g, 25.6 mmol) was dissolved in EtOH with

Raney nickel under a 35 psi atmosphere of H<sub>2</sub>. The reaction was stirred for 20 h, then
filtered through Celite washed with MeOH and concentrated in vacuo to afford 4.2 g

(100%) of methyl 5-amino-2-methylbenzoate.

## Methyl 2-(bromomethyl)-5-nitrobenzoate

10

15

Methyl 2-methyl-5-nitrobenzoate (2.0 g, 10.2 mmol) NBS (2.73 g, 15.3 mmol) and AIBN (50 mg) were dissolved in dichloroethane (100 mL). The mixture was irratiated with a photolamp for 3h. The mixture was cooled to rt and concentrated in vacuo. The residue was purified by silica gel chromatography (heptane/EtOAc 1/0, 19/1, 9/1) to afford 2.40 g (85%) of methyl 2-(bromomethyl)-5-nitrobenzoate.

## Methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-nitrobenzoate

Methyl 2-(bromomethyl)-5-nitrobenzoate (666 mg, 2.43 mmol) was added to a solution of triphenylphosphine (640 mg, 2.4 mmol) in toluene (20 mL). The resulting mixture was heated at reflux for 3h, cooled to rt, the precipitate was isolated by

filtration, washed with pentane to afford 1.2 g (92%) of methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-nitrobenzoate as a white solid.

## Methyl 5-cyano-2-methylbenzoate

5

10

15

Methyl 5-amino-2-methylbenzoate (4.2 g, 25.4 mmol) was dissolved in MeOH/H<sub>2</sub>O (20 mL/46 mL) was cooled with icebath followed by the addition of HCl (54 mL), NaNO<sub>2</sub> (2.63 g, 38.1 mmol, in H<sub>2</sub>O 60 mL). The mixture was stirred for  $\frac{1}{2}$  h, then neutralized with solid NaHCO<sub>3</sub>, extensive gasevolution. Then a cold mixture of KCN (2.48 g, 38 mmol) and CuCN (2.9 g, 33 mmol)in a H<sub>2</sub>O (40 ml)/ EtOAc (80 mL) was added. The reaction was stirred for  $\frac{1}{2}$  h, then filtered through Celite, extracted with EtOAc then washed with H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered anc concentrated in vacuo. The residue was purified by silica gel chromatography (heptane/DCM 19/1, 9/1, 1/1, 1/0) to afford 1.89 g (42%) of a white solid.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.23 (d, J = 1.7 Hz, 1 H), 7.68 (dd, J = 1.8, 7.9 Hz, 1 H), 7.38 (d, J = 7.9 Hz, 1 H), 3.94 (s, 3 H).

## Methyl 5-chloro-2-methylbenzoate

Methyl 5-chloro-2-methylbenzoate (25 g, 147 mmol) was dissolved in MeOH (0.6 L) followed by the addition of H<sub>2</sub>SO<sub>4</sub> (50 mL). The mixture was heated at reflux for 12 h, then cooled to rt and concentrated to ca 200 mL. The solution was diluted with MTBE, washed with H2O, 1N NaOH, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo to afford 24.9 g (92%) of methyl 5-chloro-2-methyl-benzoate as a white
solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.91 (d, *J* = 2.3 Hz, 1 H), 7.38 (dd, *J* = 2.3, 8.1 Hz, 1 H), 7.19 (d, *J* = 8.2 Hz, 1 H), 3.91 (s, 3 H).

## Methyl 2-(bromomethyl)-5-chlorobenzoate

Methyl 5-chloro-2-methyl benzoate (10.0 g, 54 mmol) NBS (10.6 g, 59.5 mmol) and AIBN (200 mg) were dissolved in dichloroethane (300 mL). The mixture was irratiated with a photolamp for 2h. The mixture was cooled to rt and concentrated in vacuo. The residue was purified by silica gel chromatography (heptane/DCM 9/1, 4/1, 1/1) to afford 11.8 g (83%) of methyl 2-(bromomethyl)-5-chlorobenzoate.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.98 (d, J = 2.1 Hz, 1 H), 7.49 (dd, J = 2.2, 8.2 Hz, 1 H), 7.43 (d, J = 8.2 Hz, 1 H), 4.93 (s, 2 H), 3.97 (s, 3 H).

10

5

## Methyl 2-{[bromo(triphenyl)phosphoranyl|methyl}-5-chlorobenzoate

Methyl 2-(bromomethyl)-5-chlorobenzoate (11.8 g, 44.6 mmol) was added to a solution of triphenylphosphine (11.6 g, 44.6 mmol) in toluene (400 mL). The resulting mixture was heated at reflux for 3h, cooled to rt, the precipitate was isolated by filtration, washed with pentane to afford 18.7 g (80%) of methyl 2- {[bromo(triphenyl) phosphoranyl] methyl}-5-clorobenzoate as a white solid.  $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.85-7.68 (m, 5 H), 7.63-7.57 (m, 12 H), 7.38-7.28 (m, 1 H), 5.88 (d, J = 15.0 Hz, 2 H), 3.43 (s, 3 H).

20

15

# $2\hbox{-}((Z)\hbox{-}2\hbox{-}\{3\hbox{-}[(5\hbox{-}Chloro\hbox{-}2,3\hbox{-}dihydro\hbox{-}1H\hbox{-}indol\hbox{-}1\hbox{-}yl)} sulfonyl] phenyl}\ ethenyl)\hbox{-}4-nitrobenzoic acid$

10

15

20

$$CI$$
 $N-SO_2$ 
 $O_2N$ 
 $OH$ 

Methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-nitrobenzoate (1.20 g, 2.24 mmol) was added to DMSO (30 mL) followed by NaH (100 mg, 2.4 mmol), gas evolution was observed, and the resulting mixture was heated at 60 °C for 2h. Then 3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzaldehyde (800 mg, 2.5 mmol) in toluene (50 mL) was added the reaction was stired at rt for 2h, then at 60 °C for 2h. The mixture was diluted with MTBE, washed with H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was purified by silica gel chromatography (DCM/MeOH 1/0, 19/1) to afford 760 mg (68%) of a Z/E mixture (4/1). The solid was dissolved in THF/MeOH (2/1, 60 mL) and 6N NaOH (6 mL) was added. The mixture was stirred at rt for 1 h, then diluted with MTBE, washed with 1N HCl, H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was purified by silica gel chromatography (DCM/MeOH 1/0, 19/1) to afford 574 mg (77%). This was recrystallized from MeOH. The mother liquid was recrystallized several time to afford 182 mg. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.66 (d, J = 2.4 Hz, 1 H), 8.14-8.12 (m, 1 H), 7.63-7.57 (m, 1 H), 7.49-7.18 (m, 8 H), 6.84(d, J = 12.3 Hz, 1 H), 3.65 (t, J = 8.4 Hz, 2 H), 2.86 (t, J = 8.4 Hz, 2 H).

## 2-((E)-2-{3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}ethenyl)-4-nitrobenzoic acid,

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.56 (d, J = 2.4 Hz, 1 H), 8.33-8.31 (m, 1 H), 8.18 (d, J = 16.5 Hz, 1 H), 8.08-8.03 (m, 2 H), 7.92 (d, J = 7.8 Hz, 1 H), 7.75-7.73 (m, 1 H), 7.62 (t, J = 7.8 Hz, 1 H), 7.51-7.47 (m, 2 H), 7.27-7.23 (m, 2 H), 3.98 (t, J = 8.4 Hz, 2 H), 2.94 (t, J = 8.4 Hz, 2 H).

5

10

15

20

5-Chloro-2-((E)-2-{3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}ethenyl)benzoic acid

Methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-chlorobenzoate ( 392 mg, 0.74 mmol) was added to THF (10 mL) in icebath, followed by LiCl (260 mg, 6.2 mmol), and n-BuLi (300 μL, 0.74 mmol). The reaction was stirred at rt for 10 min, then 3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzaldehyde (200 mg, 0.6 mmol) was added and the reaction was stirred at rt for 2h. The mixture was diluted with MTBE, washed with H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was purified by silica gel plug (DCM) to afford 271 mg of a Z/E mixture. The solid was dissolved in toluene (10 mL) followed by the addition of thiophenol (32 μL, 0.28 mmol) and AIBN (14 mg, 0.08 mmol). The reaction was heated at reflux for 12 h, then concentrated in vacuo. The residue was dissolved in THF (60 mL) and 6N NaOH (5 mL) was added. The mixture was stirred at 100 °C for 4 h, then diluted with MTBE, washed with 1N HCl, H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was recrystallized from MeOH to afford 123 mg. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ) δ 7.97-7.85 (m, 5 H), 7.70-7.60 (m, 3 H), 7.48 (d, J = 8.2 Hz, 1 H), 7.33-7.24 (m, 3 H), 3.97 (t, J = 8.4 Hz, 2 H), 2.93 (t, J = 8.4 Hz, 2 H).

5-Cyano-2-((E)-2-{3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl] phenyl}ethenyl)benzoic acid

Methyl 2-{[bromo(triphenyl)phosphoranyl]methyl}-5-cyanobenzoate (1.36 g, 2.6 mmol) was added to DMSO (20 mL) followed by NaH (105 mg, 2.6 mmol), gas evolution was observed, and the resulting mixture was heated at 60 °C for 2h. Then 3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzaldehyde (564 mg, 1.7 mmol) in toluene (50 mL) was added the reaction was stirred at rt for 1h, then at 60 °C for 1h. The mixture was diluted with MTBE, washed with H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The residue was purified by silica gel chromatography (DCM/heptane 1/1, 1/0) to afford 616 mg (73%) of a Z/E mixture. The solid was dissolved in THF (60 mL) and 1N NaOH (10 mL) was added. The mixture was stirred at rt for 12 h, then diluted with MTBE, washed with 1N HCl, H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo to afford 567 mg (95%). This was purified by preparative reverse phase HPLC to afford 144 mg of pure (E) and 99 mg of (Z). <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 8.24 (s, 1 H), 8.05-7.89 (m, 5 H), 7.76-7.73 (m, 1 H), 7.63 (t, *J* = 7.7 Hz, 1 H), 7.49-7.44 (m, 2 H), 7.27-7.24 (m, 2 H), 3.98 (t, *J* = 8.5 Hz, 2 H), 2.93 (t, *J* = 8.5 Hz, 2 H).

# 5-Cyano-2-((Z)-2-{3-[(5-chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl] phenyl}ethenyl)benzoic acid

20

5

10

15

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.33 (d, J = 1.7 Hz, 1 H), 7.84-7.81 (m, 1 H), 7.59-7.57 (m, 1 H), 7.47-7.12 (m, 8 H), 6.82 (d, J = 12.2 Hz, 1 H), 3.66 (t, J = 8.5 Hz, 2 H), 2.88 (t, J = 8.3 Hz, 2 H).

# 2-(2-{3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}cyclopropyl)-5-cyanobenzoic acid

Diazomethane solution (400 ml, from 36 g Dizald, for procedure see Denmark, S. E.; Stavenger, R. A.; Faucher, A-M.; Edwards, J. P. J. Org. Chem. 1997, 62, 3375) was added to a solution of methyl 5-cyano-2-(2-{3-[(5-chloro-2,3-dihydro-1H-indol-1yl)sulfonyl] phenyl}ethenyl)benzoate (850 mg, 1.7 mmol) and Pdba (100 mg) in DCM (150 mL). Extensive gas evolution was observed, the resulting mixture was stirred for 12 h, then HOAc (5 mL) was added, filtered through Celite, washed with 1n NaOH, brine, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo to afford 982 mg of a solid. The residue in DCM (100 mL) was cooled with icebath and O3 was bubbled through for 30 min. Then NaBH4 (500 mg) was added and the mixture was stirred for 30 min at rt. The mixture was passed through a silica plug and concentrated in vacuo. The residue was purified by silica gel chromatography (heptane/DCM 9/1, 4/1, 1/1, 0/1) to afford 124 mg of the desired cyclopropane. The solid was dissolved in THF (25 mL) and 6N NaOH (5 mL) was added, the resulting mixture was stirred for 16h at rt, then diluted with MTBE, washed with 1N HCl, H<sub>2</sub>O, brine, dried (MgSO<sub>4</sub>) filtered and The residue was purified by silica gel chromatography concentrated in vacuo. (DCM/MeOH 1/0, 19/1, 9/1, 4/1) to afford 51 mg (6%). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.27 (d, J = 1.6 Hz, 1 H), 7.79 (d, J = 8.1 Hz, 1 H), 7.60-7.57 (m, 3 H), 7.40 (d, J =5.3 Hz, 2 H), 7.25 (d, J = 8.1 Hz, 1 H), 7.18-7.16 (m, 1 H), 7.04 (s, 1 H), 3.95 (t, J =8.3 Hz, 2 H), 3.18-3.13 (m, 1 H), 2.87 (t, J = 8.3 Hz, 2 H), 2.24-2.19 (m, 1 H), 1.65-1.60 (m, 1 H), 1.54-1.49 (m, 1 H).

25

5

10

15

20

#### Example 8:

In other embodiments, the invention includes benzoxazine derivatives of the formula

$$R_2$$
 $H$ 
 $O$ 
 $N$ 
 $R_4$ 

## 5 wherein

R<sub>2</sub> is an electron withdrawing group; and

R<sub>4</sub> is an optionally substituted aryl, provided that the aryl is not simultaneously substituted with a sulfonamide and a urea or thiourea, and further provided that the aryl is not solely substituted at the ortho-position relative to Y.

10

15

20

## 2-{3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]phenyl}-4-oxo-4H-3,1-benzoxazine-6-carbonitrile

2-({3-[(5-Chloro-2,3-dihydro-1H-indol-1-yl)sulfonyl]benzoyl}amino)-5-cyanobenzoic acid (PHA-524523, 884 m, 1.84 mmol) was dissolved in anhydrous THF (30 mL) and Et<sub>3</sub>N (0.563 mL, 4.04 mmol) under N<sub>2</sub>. Addition of ethyl chloroformate (0.193 mL, 2.02 mmol, Aldrich) to the yellow solution produced a white precipitate, which was stirred overnight at RT. The solvent was evaporated and the resultant residue suspended in CH<sub>2</sub>Cl<sub>2</sub> (100 mL). The organic layer was washed 2x with 1.0M HCl, 1x with water and 1x with brine (100 mL each). The crude product was purified on a Biotage Flash 40M (90 g) silica cartridge using a step gradient of 0-1% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub>. After evaporation the resultant solid was dried under vacuum at 100 °C to

afford 280 mg (33%) of white solid. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.65 (d, J = 1.9 Hz, 1 H), 8.52 (s, 1 H), 8.47 (d, J = 8.0 Hz, 1 H), 8.36 (dd, J = 8.4, 1.9 Hz, 1 H), 8.11 (d, J = 8.4 Hz, 1 H), 7.92 (d, J = 8.4 Hz, 1 H), 7.85 (t, J = 7.9 Hz, 1 H), 7.53 (d, J = 8.6 Hz, 1 H), 7.30 (d, J = 8.6 Hz, 1 H), 7.26 (s, 1 H), 3.99 (t, J = 8.4 Hz, 2 H).

## **Example 9: Library Synthesis**

## **General Experimental**

<sup>1</sup>H NMR spectra were measured using a Bruker AVANCE 300 spectrometer at rt in DMSO-d<sub>6</sub> at an operating frequency of 300.13 MHz and are referenced to residual DMSO-d<sub>6</sub> (2.54 ppm) unless otherwise noted. All coupling constants are reported in Hz. All non-combinatorial reactions were performed under a nitrogen atmosphere.

## 15 Synthetic Procedures Using Wang Resins

## Scheme 9.1

20

25

To a dry, 2-L polypropylene bottle equipped with a nitrogen inlet and an overhead stirrer was added Wang resin (21, 38.6 g, 49.7 mmol, 1.3 mmol/g, Novabiochem), DMF (600 mL), 5-bromoisatoic anhydride (22, 60.0 g, 248 mmol, dissolved in 100 mL of DMF), and DMAP (30.3 g, 248 mmol, dissolved in 100 mL of DMF). The reaction was heated under nitrogen to 65 °C and stirred for 12 h. The reaction was

then filtered and washed as follows: DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, and CH<sub>2</sub>Cl<sub>2</sub>. The washed resin was transferred back to the 2-L reaction flask and treated a second time with DMF (600 mL), 5-bromoisatoic anhydride (60.0 g, 248 mmol, dissolved in 100 mL of DMF), and DMAP (30.3 g, 248 mmol, dissolved in 100 mL of DMF). The reaction was stirred at 65 °C for 4 h and then filtered and washed with DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, and CH<sub>2</sub>Cl<sub>2</sub>) to afford (48.57 g) of 23 as an off-white resin. CNH analysis: Calcd (1.3 mmol): N, 1.82, Found: N, 1.67% (loading = 1.2 mmol/g).

10

5

## Scheme 9.2

$$\begin{array}{c} \text{COOH} \\ \text{R} & \text{SO}_2\text{Cl} \\ \\ \text{COCl} \\ \text{COCl} \\ \text{COCl} \\ \text{R} & \text{SO}_2\text{Cl} \\ \end{array}$$

15

20

25

To a suspension of a 3-chlorosulfonylbenzoic derivative (24, 31.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (100 mL) was added DMF (two drops), followed by oxalyl chloride (31.6 mL of a 2 M solution in CH<sub>2</sub>Cl<sub>2</sub>, 63.2 mmol) under a nitrogen atmosphere. Gas evolution and disappearance of the suspension was noted during the course of the reaction. After the reaction was stirred for 18 h, the acid chloride was concentrated to dryness, azeotroped with toluene (2 x 25 mL), and then placed on a high vacuum. Dry anthranilic acid-derivatized Wang resin (7.0 g, 8.4 mmol) was added to an 8-oz wide-mouth bottle, followed by CH<sub>2</sub>Cl<sub>2</sub> (35 mL) and pyridine (35 mL). The acid chloride was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) and added to resin, effecting HCl (g) evolution. The reaction jar was flushed with nitrogen, capped and shaken for 4 h.

The resin was then filtered and washed (CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, THF, MeCN, THF, CH<sub>2</sub>Cl<sub>2</sub>; 50 mL each wash) to afford **25** as a tan resin.

## **5** Scheme 9.3

10

15

20

25

Sulfonyl chloride resin (25, 50 mg, 60 μmol) was added down the columns of a 96-well microtiter filter plate using a CH<sub>2</sub>Cl<sub>2</sub> isopycnic slurry. After draining the wells, the plate was inserted into a solid phase reaction apparatus. Amines (300 μL of a 0.75 M solution, 225 μmol) were then added across the rows, followed by triethylamine (250 μL of a 1.8 M solution) and CH<sub>2</sub>Cl<sub>2</sub> (250 μL). The plate was capped and spun on an overhead rotisserie for 16 h. After removal of the plate from the solid phase reaction apparatus, the wells drained and each well was washed (DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, and CH<sub>2</sub>Cl<sub>2</sub>).

The plate was again inserted into the solid phase reaction apparatus and a 15% solution of TFA in  $CH_2Cl_2$  (625  $\mu$ L) was added. The plate was spun on an overhead rotisserie for 3 h and the crude sulfonamides were then drained into a 1-mL 96-well plate. The resin was washed with  $CH_2Cl_2$  (1.5 mL) and the washes collected in additional 1-mL plates. LC/MS samples were prepared by transferring 40  $\mu$ L of solution to a separate 96-well plate, concentrating the samples and then dissolving in DMSO (125  $\mu$ L) and diluting with acetonitrile (750  $\mu$ L).

## Scheme 9.4

10

15

20

To a dry, 2-L polypropylene bottle equipped with a nitrogen inlet and an overhead stirrer was added Wang resin (11, 15.1 g, 21.1 mmol, 1.4 mmol/g, Novabiochem), DMF (500 mL), 5-cyanoisatoic anhydride (10, 20.0 g, 106 mmol, dissolved in 100 mL DMF), and DMAP (13.0 g, 106 mmol, dissolved in 100 mL DMF). The mixture was heated under nitrogen to 53 °C and stirred for 16 h. The reaction was then filtered and washed with 500 μL of the following solvents: DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, DMF, DMF, and DMF. The resin was transferred back to the 2-L reaction flask and treated a second time with DMF (500 mL), 5-cyanoisatoic anhydride (10, 20.0 g, 106 mmol, dissolved in 100 mL DMF), and DMAP (13.0 g, 106 mmol, dissolved in 100 mL DMF). The reaction was stirred at 60 °C for 22 h and then filtered and washed with 500 μL of CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, to afford 15.3 g of 12 as a pale yellow resin. Elemental analysis: N, 3.20 % (loading = 1.14 mmol/g).<sup>5</sup>

## Scheme 9.5

- 123 -

Dry 5-cyano anthranilic acid-derivatized Wang resin (5, 5.0 g, 1.0 mmol/g loading, 5.0 mmol) was added to an 8-oz wide mouth bottle, followed by CH<sub>2</sub>Cl<sub>2</sub> (30 mL) and pyridine (30 mL). The acid chloride (4) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (30 mL) and added to the resin, effecting HCl (gas) evolution. The jar was flushed with nitrogen, capped, and shaken for 64 h. The resin was then filtered and washed (DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, THF, THF, THF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>; 400 mL each wash) to afford 6.

## Scheme 9.6

5

10

15

20

25

Sulfonyl chloride resin (**6**, 50 mg, 50 μmol) was added to the wells of a 96-well filter plate using a CH<sub>2</sub>Cl<sub>2</sub> isopycnic slurry. After draining the wells, the plate was inserted into a solid phase reaction apparatus. Amines (250 μL of a 2 M solution, 500 μmol) were then added, followed by triethylamine (250 μL of a 2 M solution) and CH<sub>2</sub>Cl<sub>2</sub> (250 μL). The plate was then capped and spun on an overhead rotisserie for 20 h. After removal of the plate from the solid phase reaction block, the wells were drained and washed (DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>; 375 μL each wash).

The plate was again inserted into the solid phase reaction apparatus and a 50% solution of TFA in  $CH_2Cl_2$  (500  $\mu$ L) was added. The plate was spun on an overhead rotisserie for 3 h and the crude sulfonamides (7) were then drained into a standard 96-well plate. The resin was washed with 250  $\mu$ L of additional 50% TFA solution. Products were concentrated under nitrogen and then analyzed by LC/MS (see general LC/MS procedure).

The crude samples were dissolved in THF, and eluted through a plug of Celite<sup>®</sup>. LC/MS showed a reduced amount of impurity in all of the samples. The samples that were less than 70% pure were then eluted through a plug of silica gel using THF as the mobile phase and the samples were analyzed by LC/MS.

5

#### Scheme 9.7

10

15

20

25

To a standard 96-well filter plate was added 50 mg (60 μmol) of 5-bromoanthanilic acid derivatized Wang resin as an isopycnic solution in CH<sub>2</sub>Cl<sub>2</sub> (3). After the wells were drained, the plate was inserted into a plate clamp assembly. The acid chloride diversity set (2) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (300 μL) and added to the plate, followed by TEA (250 μL, 1 M CH<sub>2</sub>Cl<sub>2</sub>, 250 μmol) and CH<sub>2</sub>Cl<sub>2</sub> (300 μL). The plate was capped and spun on an overhead rotisserie for 16 h. After removal of the plate from the plate clamp assembly, the wells were drained and the resin washed with 500 μL of the following solvents: CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, MeCN, CH<sub>2</sub>Cl<sub>2</sub>, CHCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, THF, MeCN, THF, CH<sub>2</sub>Cl<sub>2</sub>. The plate was reinserted into the plate clamp assembly and the washed resin was treated with 750 μL of 25% TFA/CH<sub>2</sub>Cl<sub>2</sub> solution for 3 h. The solution was then filtered from the Wang resin and collected in a separate plate to afford the crude amides (4). The plates were concentrated and analyzed by LC/MS (see general LC/MS procedure).

#### Scheme 9.8

00833 US1

5

10

15

Br 
$$CO_2Me$$
 +  $CI$   $COOH$   $OOH$   $OOH$ 

After concentration of the acid chloride solutions (2), methyl-2-amino-5-

bromobenzoate (5, 125 μL, 1 M THF, 125 μmol/well) was added to the plate followed by potassium carbonate (1 mL, 0.38 M THF, 380 μmol/well). The reactions were capped, heated to 50 °C and shaken for 12 h. Triethylenetetramine resin (160 mg, 464 μmol) was added to the wells to scavenge the excess acid chloride and the plate spun for 2.5 h. The crude methyl esters were purified (if necessary) using a column consisting of basic alumina (ca 200 mg), SAX (ca 200 mg), and SCX (ca 400 mg, activated with 1% HOAc/MeOH) in descending order. The products were eluted with THF and the fractions analyzed by LC/MS.

LiOH [375  $\mu$ L, 1 M H<sub>2</sub>O/THF (50:50), 3 equiv)] was added to the esters and the plate was capped and spun for 1 h. The THF was then removed in vacuo. The crude solids were suspended in methyl ethyl ketone (MEK, 500  $\mu$ L) and extracted with 2 N HCl (250  $\mu$ L). The MEK layer was removed and the aqueous layer extracted again with MEK (500  $\mu$ L). The combined organic layers were washed with 50% brine solution, passed through a plug of sodium sulfate, collected in a 1-mL plate, and dried under nitrogen to afford the amide products (6). The solids were then analyzed using LC/MS (see general LC/MS procedure).

#### Scheme 9.9

25

20

$$\begin{array}{c|c} & & & & & \\ & & & & \\ \hline & & & \\ & & & \\ \hline & & \\ & & \\$$

To each vial of an array of 1-mL vials arranged in a 96-well format was added 44 mg (50 μmol) of 5-cyanoanthranilic acid-derivatized Wang resin (**5**) as an isopycnic solution in CH<sub>2</sub>Cl<sub>2</sub>. The acid chloride diversity set<sup>2</sup> (**8**, 500 μmol) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (300 μL) and added to the vials, followed by TEA (250 μL, 2 M CH<sub>2</sub>Cl<sub>2</sub>, 500 μmol), and CH<sub>2</sub>Cl<sub>2</sub> (300 μL). The vials were capped, heated to 60 °C, and shaken for 21 h. After completion of the reaction, the resin was transferred to a 96-well filter plate and washed with of the following solvents: DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub> (500 μL/wash). The plate was placed into a clamp assembly and each well was treated with 500 μL of 50% TFA/CH<sub>2</sub>Cl<sub>2</sub> solution for 2 h.<sup>3</sup> The resultant solution was then filtered from the Wang resin, collected in a separate plate, and dried under nitrogen to afford the crude amides (**9**).

Scheme 9.10

5

10

15

$$\begin{array}{c|c} & & & \\ &$$

Resin-bound 4-Acetoxybenzoyl Anthranilic Acid. To a 500-mL round bottom flask under nitrogen was added 4-acetoxybenzoic acid (20.7 g, 115.5 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (200 mL). After cooling the flask to 0 °C, oxalyl chloride (57.8 mL of a 2 M solution, 116 mmol) and a few drops of DMF were added. The reactions were allowed to warm to room temperature and stirred for 3 h. These solutions were directly transferred to a

2-L serum flask containing 5-bromoanthranillic acid resin (21, 7.0 g, 7.7 mmol), pyridine (100 mL) and CH<sub>2</sub>Cl<sub>2</sub> (100 mL). The resulting mixtures were stirred under nitrogen overnight and then filtered into a glass fritted funnel. The resin was then washed with DMF (3 x 100 mL), CH<sub>2</sub>Cl<sub>2</sub> (5 x 100 mL), and MeOH (5 x 100 mL).

The resin was then dried in a vacuum oven at 60 °C for 72 h to afford **22** (8.0 g). A sample was cleaved from the resin by stirring in 25% TFA in CH<sub>2</sub>Cl<sub>2</sub> for 3 h: <sup>1</sup>H NMR (acetone- $d_6$ )  $\delta$  2.31 (s, 3H), 7.35 (d, J = 2.1, 1H), 7.37 (d, J = 2.0, 1H), 7.82 (d, J = 2.5, 1H), 7.86 (d, J = 2.5, 1H), 8.07 (dd, J = 2.1, 8.7, 1H), 8.27 (d, J = 2.5, 1H), 8.90 (d, J = 9.0, 1H).

10

Resin-bound 4-Hydroxybenzoyl Anthranilic Acid. To a 250-mL serum bottle was added acetoxy resin 23 (7.0 g, 7.7 mmol), CH<sub>2</sub>Cl<sub>2</sub> (70 mL), and piperidine (150 mL, 2 M CH<sub>2</sub>Cl<sub>2</sub>). The slurry was stirred for 2 h at room temperature. The resins were then filtered and washed with DMF (3 x 100 mL), Et<sub>3</sub>N (1 M CH<sub>2</sub>Cl<sub>2</sub>, 2 x 100 mL), and MeOH (2 x 100 mL), CH<sub>2</sub>Cl<sub>2</sub> (40 mL), MeOH (40 mL), CH<sub>2</sub>Cl<sub>2</sub> (40 mL), MeOH (40 mL), CH<sub>2</sub>Cl<sub>2</sub> (40 mL), and MeOH (40 mL). The resin was then dried for 72 h in a vacuum oven at room temperature to afford 6.6 g of 24 as a yellow resin.

20

15

Synthesis of Resin-Bound Phenol 12a. To a 200-mL Wheaton bottle equipped with an overhead stirrer was added resin-bound acetate (11a, 5.0 g) followed by piperidine (150 mL of a 2 M solution in CH<sub>2</sub>Cl<sub>2</sub>, 300 mmol). The reaction was stirred for 2 h at room temperature. The resin was then filtered from the reaction mixture, washed with DMF, DMF, DMF, Et<sub>3</sub>N (1 M in CH<sub>2</sub>Cl<sub>2</sub>), MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>

10

15

20

25

5

Mitsunobu Reaction (Procedure A). To each well of a fritted 96-well plate was added phenol resin (12a-c, 20.0 mg, 20.0 µmol) as an isopycnic solution (20% THF in CH<sub>2</sub>Cl<sub>2</sub>) and the plate was placed in a solid phase reaction assembly. The alcohol diversity element (200 µL of a 1 M solution in THF, 200 µmol) was then added, followed by triphenylphosphine (200 µL of a 1 M solution in THF, 200 µmol). The wells were flushed with nitrogen, capped, and placed in the -20 °C freezer for 1 h. While in the freezer, DIAD [200 µL of a cooled (-20 °C), freshly made 1 M solution in THF) was added to each well. The plate was removed from the freezer after 1 h and then spun on the rotisserie for 16 h. The reaction mixture was drained from the plate and the resin then washed with THF, THF, THF (the plate was capped and spun on an overhead rotisserie for 30 min), THF, MeOH, THF, MeOH, THF, MeOH, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH (the plate was capped and spun on an overhead rotisserie for 30 min), CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>; 500 μL each solvent. The crude aryl ethers were then cleaved from the resin using 500 µL of 50% TFA in CH<sub>2</sub>Cl<sub>2</sub>. The resulting products (13a-c) were concentrated under a nitrogen stream and analyzed by HPLC/MS.

10

15

Mitsunobu Reaction (Procedure B). To 72 wells of a fritted 96-well plate was added phenol resin as an isopycnic solution (12a-c, 20.0 mg, 20.0 µmol) and the plate was placed in a solid phase reaction assembly. The alcohol diversity element (200 µL of a 1 M solution in THF, 200 µmol) was then added, followed by triphenylphosphine (200 µL of a 1 M solution in THF, 200 µmol) and Et<sub>3</sub>N (200 µL of a 1 M solution in THF, 200 µmol). The wells were flushed with N<sub>2</sub>, capped, and placed in the -20 °C freezer for 1 h. While in the freezer, DIAD [200 µL of a cooled (-20 °C), freshly made 1 M solution in THF] was added to each well. The plate was removed from the freezer after an hour and then spun on the rotisserie for 16 h. The reaction mixture was drained from the plate and the resin then washed with THF, THF, THF (the plate was capped and spun on an overhead rotisserie for 30 min), THF, MeOH, THF, MeOH, THF, MeOH, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, MeOH (the plate was capped and spun on an overhead rotisserie for 30 min), CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>; 500 μL each solvent. The crude aryl ethers were then cleaved from the resin using 500 µL of 50% TFA in CH<sub>2</sub>Cl<sub>2</sub>. The resulting products (13a-c) were concentrated under a nitrogen stream and analyzed by HPLC/MS.

20

25

Synthesis of Acetate 15. To a 250-mL round bottom flask was added a solution of methyl-2-amino-5-bromobenzoate [14, 5.0 g, 21.7 mmol dissolved in pyridine (10 mL) and  $CH_2Cl_2$  (10 mL)], followed by o-acetoxybenzoyl chloride<sup>5</sup> (4.7 g, 33.8 mmol dissolved in 60 mL of  $CH_2Cl_2$ ). The mixture was stirred overnight under a nitrogen atmosphere. Polyamine resin (4.0 g) was then added to the reaction mixture and the reaction was stirred for 4 h. After filtration and concentration of the reaction mixture, a white residue was obtained. The residue was recrystallized from  $CH_2Cl_2$  to afford 8.0 g (94%) of 15 as a white solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  2.24 (s, 3H), 3.86 (s, 3H),

7.30 (d, J = 8.1, 1H), 7.46 (dt, J = 1.1, 7.6, 1H), 7.66 (dt, J = 1.7, 8.0, 1H), 7.82 (dd, J = 1.7, 7.7, 1H), 7.86 (dd, J = 2.5, 8.9, 1H), 8.06 (d, J = 2.5, 1H), 8.38 (d, J = 8.9, 1H).

5

10

15

**Synthesis of Phenol 16**. To a 50-mL round bottom flask was added o-acetoxy methyl ester **15** (1.0 g, 2.9 mmol), CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and piperidine (2.0 mL of a 2 M solution in CH<sub>2</sub>Cl<sub>2</sub>, 4.0 mmol). After the reaction mixture was stirred for 3 h, the solvent was removed and the crude residue dried under high vacuum overnight. The residue was then dissolved in CH<sub>2</sub>Cl<sub>2</sub> and acid chloride resin (2.0 g, 2.1 mmol) was added to scavenge excess piperidine. The mixture was stirred for 4 h, filtered, and concentrated to afford 0.54 g (60%) of phenol **16** as a white solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  3.90 (s, 3H), 6.98 (t, J = 7.6, 1H), 7.02 (d, J = 7.6, 1H), 7.44 (dt, J = 1.8, 8.2, 1H), 7.82 (dd, J = 2.5, 9.0, 1H), 7.93 (dd, J = 1.8, 7.9, 1H), 8.08 (d, J = 2.5, 1H), 8.60 (d, J = 9.0, 1H).

#### Scheme 9.11

$$\begin{array}{c} COCl \\ H_2N \\ O \\ Cl \end{array}$$

$$\begin{array}{c} 22 \\ \text{pyridine, } CH_2Cl_2 \end{array}$$

$$\begin{array}{c} COCl \\ HN \\ O \\ Cl \end{array}$$

20

Resin-Bound *m*-Iodo Benzamide 23. Acid chloride 22 was redissolved in CH<sub>2</sub>Cl<sub>2</sub> (30 mL) and added to resin-bound 5-chloroanthranilic acid (20, 3 g, 1.06 mmol/g loading, 3.18 mmol) swollen with pyridine (30 mL) in a 500-mL serum flask equipped with an overhead stirrer. The flask was purged with nitrogen and the resin stirred for

16 h. The resin was filtered from the reaction mixture and washed with alternating CH<sub>3</sub>CN and CH<sub>2</sub>Cl<sub>2</sub> washes (8 x 300 mL) to afford 23.

5

10

Resin-Bound Stannylate 24. To a CH<sub>2</sub>Cl<sub>2</sub> slurry of m-iodo resin (23, 1 g, 1.06) mmol/g) in a 250-mL serum flask was added 1 mL of the following solutions; palladium acetate (0.0022 g/1 mL, 0.01 mmol, 0.1 equiv,), triphenyl phosphine (0.0065 g/mL, 0.025 mmol, 0.25 equiv), DIPEA (0.0065 g/mL, 0.05 mmol, 0.5 equiv) in DMF. Hexamethyl ditin (0.065 g, 0.2 mmol, 2.0 equiv) was added to the flask, which was then purged with nitrogen and heated to 60 °C for 18 h. The reaction mixture was drained and the resin washed with alternating DMF, CH<sub>3</sub>CN and CH<sub>2</sub>Cl<sub>2</sub> (10 x 150 mL) to yield 24 as a dark brown resin.

SnMe<sub>3</sub>

$$\begin{array}{c}
1. \text{ Pd}_2(\text{dba})_3 \text{ (1 equiv), } R_1\text{COCl} \\
K_2\text{CO}_3, \text{ DIPEA, THF} \\
\hline
2. 50 \% \text{ TFA}
\end{array}$$

15

20

Resin Bound Library of Aryl Ketones. Hexamethyl ditin derivatized Wang resin (24, 24 mg, 24 µmol) was added as an isopycnic solution (degassed THF) to an array of 1-dram vials arranged in a 96-well format. Tris(dibenzylidene acetone) dipalladium (0) (22 mg, 24 µmol, 1.0 equiv) was added to each vial (in a solution of degassed THF). DIPEA (20 µL) was added to each vial followed by K<sub>2</sub>CO<sub>3</sub> (10 mg)

10

15

20

and degassed THF (0.5 mL). The vials were capped and shaken. The vials were uncapped and the acid chloride diversity elements<sup>7</sup> (10 equiv) were then added, the vials purged with nitrogen for 5 sec, capped, shaken and heated 60 °C for 20 h. After the reactions cooled to room temperature, the resin was transferred to a 96-well polypropylene fritted plate. The resin was washed (CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, CH<sub>3</sub>CN, DMF, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, H<sub>2</sub>O, THF, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 250 μL each wash) and the plate inserted into a solid phase reaction block. A solution of 50% TFA in CH<sub>2</sub>Cl<sub>2</sub> (600 μL) was added to the plate. The plate was capped and spun on an overhead rotisserie for 3 h. The crude aryl ketones (25) were then drained into a 96-well collection plate, concentrated to dryness, and analyzed by HPLC/MS.

#### **Purification Procedures**

**Liquid-liquid extraction (basic).** To a 96-well plate of crude samples was added methyl ethyl ketone (MEK,  $500 \, \mu L$ ) and 2 N NaOH ( $500 \, \mu L$ ). The plates were capped and shaken. After the plates were uncapped, the organic layer was separated from the aqueous layer.

**Liquid-liquid extraction (acidic).** The aqueous layer of the above extraction was treated with 6 N HCl (500  $\mu$ L) and extracted with MEK (1 mL). The plates were capped, shaken, and the organic layer was separated from the aqueous layer.

Hydromatrix® extraction (AMRI SEC-C-44). A set of 2-mL square-well plates were filled with Hydromatrix® and washed with MEK and CH<sub>2</sub>Cl<sub>2</sub> (500 μL/well).
The plates were then placed in a vacuum oven (T = 35 °C) overnight. After cooling, the Hydromatrix® was treated with 2 N HCl (600 μL)<sup>7</sup> and the plates were stacked. The crude library samples were dissolved in MEK and pipetted onto the columns. MEK was used to elute the compounds, and several 2-mL fractions were collected.

30

## Crystallization

After treatment with Hydromatrix  $^{\otimes}$ , several compounds crystallized out of the 50% MeOH/MEK solution. The liquid was removed from the well, and the solid dissolved in DMSO (250  $\mu$ L) and transferred to a Marsh tube.

5 **HPLC analysis method**. The purity of the library was determined from the relative peak area of the UV absorbance. The identity of the compound was determined by MS confirmation of the molecular weight. The samples from this library were best prepared from DMSO solutions of the crude compounds. To a 96-well LC/MS plate was added ca 30 μL of DMSO solution (solution concentration was typically ca 30 mM). DMSO (ca 50 μL) and MeCN (ca 750 μL) were then used to dilute the samples.

## **HPLC Conditions**

15 Column: Zorbax SB-C18 (4.6 x 75 mm, 3.5 microns)

Gradient: A solvent: 100% MeCN (0.075% HCO<sub>2</sub>H), B solvent: 100% H<sub>2</sub>0 (0.075%

HCO<sub>2</sub>H)

Flow: 2 mL/min

Detection wavelength: 220 nm (UV)

20 Autosampler: Gilson 215 Liquid Handler

Pump: Shimadzu LC-10AD VP

Detector: Shimadzu UV-VIS Detector SPD-10A VP

Injection volume: 40 μL

Mass Spectrometer: PESCIEX API 150EX

**Table 1. Gradient Profile** 

Time (min)	%В
0	75
4.5	10
7.0	10
9.0	75

## Preparation of Benzioic Acid Derivatives for Library Synthesis

COOH
$$\frac{\text{CISO}_3\text{H}}{\Delta}$$
SO<sub>2</sub>C

10

15

Chlorosulfonic acid (50 mL, 752 mmol) was added to a 250-mL round-bottom flask and cooled to 0 °C in the presence of nitrogen. p-Toluic acid (1, 10 g, 73.7 mmol) was added in small portions over 5 min to give a yellow solution. The solution was warmed to room temperature and heated to 100 °C overnight. The reaction mixture was then cooled to room temperature and poured over ice (ca 750 g). The resulting precipitate was filtered, washed with water and dried in a vacuum oven at 70 °C for 8 h to afford 14.38 g (83%) of 2 as an off-white solid:  $^{1}$ H NMR (DMSO- $d_{6}$ )  $\delta$  2.60 (s, 3H), 7.28 (d, J = 7.9 Hz, 1H), 7.79 (d, J = 7.8 Hz, 1H), 8.31 (s, 1H), 13.87 (br s, 1H).

10

15

20

$$COOH$$
 $CISO_3H$ 
 $A$ 
 $SO_2CI$ 

To a 250-mL round-bottom flask cooled to 0 °C under nitrogen was added chlorosulfonic acid (50 mL, 752 mmol), followed by o-bromobenzoic acid (3, 10.0 g, 49.7 mmol) in small portions over 2 min to give a brownish solution. The solution was warmed to room temperature and heated to 115 °C overnight. The reaction mixture was then cooled to room temperature and poured over ice (ca 750 g). The resulting precipitate was filtered, washed with water and dried in a vacuum oven at 80 °C for 7 h to afford 12.81 g (86%) of 4 as an off-white solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  7.75 (d, J = 10.1 Hz, 1H), 7.65 (d, J = 10.1 Hz, 1H), 8.46 (s, 1H), 13.96 (br s, 1H).

Chlorosulfonic acid (50 mL 752 mmol) was added to a 250-mL round-bottom flask and cooled to 0 °C in the presence of nitrogen. p-Bromobenzoic acid (5, 10.0 g, 49.7 mmol) was added in small portions over 2 min to give a brownish solution. The solution was warmed to room temperature and heated to 145 °C overnight. The reaction mixture was then cooled to room temperature and poured over ice (ca 750 g). The resulting precipitate was filtered, washed with water and dried in a vacuum oven at 80 °C for 7 h to afford 13.21 g (89%) of 6 as a tan solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  7.60 (dd, J = 2.1, 8.3 Hz, 1H), 7.69 (d, J = 8.2 Hz, 1H), 8.31 (d, J = 2.1 Hz, 1H), 14.05 (br s, 1H).

COOH
$$CISO_3H$$

$$SO_2CI$$

25

Chlorosulfonic acid (50 mL, 752 mmol) was added to a 250-mL round-bottom flask and cooled to 0 °C in the presence of nitrogen. o-Toluic acid (7, 10.0 g, 73.4 mmol) was added in small portions over 2 min to give a brownish solution. The solution was warmed to room temperature and heated to 145 °C overnight. The reaction mixture was then cooled to room temperature and poured over ice (ca 750 g). The resulting precipitate was filtered, washed with water and dried in a vacuum oven at 80 °C for 7 h to afford 15.53 g (90%) of 8 as an off-white solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  2.53 (s, 3H), 7.26 (d, J = 7.9 Hz, 1H), 7.63 (d, J = 7.9 Hz, 1H), 8.07 (s, 1H), 13.60 (br s, 1H).

10

15

20

25

5

COOH
$$CISO_3H$$

$$O$$

$$SO_2CI$$

Chlorosulfonic acid (50 mL, 752 mmol) was added to a 250-mL round-bottom flask and cooled to 0 °C in the presence of nitrogen. p-Anisic acid (9, 10.0 g, 73.4 mmol) was added in small portions over 2 min to give a yellow solution. The solution was warmed to room temperature and heated to 63 °C for 1 h. The reaction mixture was then cooled to room temperature and poured over ice (ca 750 g). The resulting precipitate was filtered, washed with water and dried in a vacuum oven at 70 °C for 12 h to afford 14.62 g (85%) of **10** as a white solid: H NMR (DMSO- $d_6$ )  $\delta$  3.84 (s, 3H), 7.06 (d, J = 8.7 Hz, 1H), 7.70 (dd, J = 2.3, 8.7 Hz, 1H), 8.31 (d, J = 2.3 Hz, 1H), 13.82 (br s, 1H).

Solid *p*-anisic acid (11, 10.0 g, 66 mmol) was added to an ice-cooled, 250-mL round-bottom flask containing chlorosulfonic acid (50 mL, 752 mmol) under nitrogen. The solution was heated at 65 °C for 1 h and turned bright yellow. The reaction mixture was cooled to room temperature and poured over ice (ca 750 g). The resulting

precipitate was then filtered, washed with water and dried in a vacuum oven at 70 °C for 8 h to yield 13.18 g (80%) of **12** as a pale yellow solid:  $^{1}$ H NMR (DMSO- $d_{6}$ )  $\delta$  3.88 (s, 3H), 7.06 (d, J = 8.7 Hz, 1H), 7.90 (dd, J = 2.4, 8.6 Hz, 1H), 8.31 (d, J = 2.3 Hz, 1H), 13.82 (br s, 1H).

5

# General Procedure for the Conversion of Acids to Acid Chlorides in a Plate Format

OH 
$$\frac{(COCI)_2}{CH_2CI_2}$$
  $R$ 

10

15

20

25

To a plate of 2-mL glass reaction tubes arranged in a standard 96-well format was added the diversity set of carboxylic acids (1, 250  $\mu$ L, 1 M THF, 250  $\mu$ mol). The samples were concentrated in a Genevac HT-4 (20% heat with no heat boost for 1 h). A solution of 1% DMF/CH<sub>2</sub>Cl<sub>2</sub> (50  $\mu$ L) was added to the wells, followed by CH<sub>2</sub>Cl<sub>2</sub> (250  $\mu$ L). The carboxylic acid plate was placed in a nitrogen-filled glove bag and oxalyl chloride (125  $\mu$ L, 2 M CH<sub>2</sub>Cl<sub>2</sub>, 250  $\mu$ mol) was added. After the addition of CH<sub>2</sub>Cl<sub>2</sub> (250  $\mu$ L), a capmat with 96 predrilled holes was fitted on the plate. The plate was shaken on an orbital shaker in a N<sub>2</sub> filled glove bag for 6-8 h.

## Preparation of Isatioc Anhydride Derivatives

1 2

To a dry, 4-L round bottom flask was added 175 g (810 mmol) of 2-amino-5-bromobenzoic acid (19), triphosgene (83 g, 278 mmol), and dioxane (3 L). The suspension was stirred under  $N_2$  and heated to reflux. The reaction was found to be complete by TLC and NMR after stirring at reflux for 3 h, but did not become homogenous at any time. After cooling to room temperature, the reaction was filtered

20

25

and the precipitate washed with ether. The solid was dried in the vacuum oven at 40 °C to afford 5-bromoisatoic anhydride (20, 151.1 g, 72%) as a white solid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  7.29 (d, J = 8.7, 1H), 7.91 (dd, J = 2.5, 8.7, 1H), 8.09 (d, J = 2.3, 1H).

Sodium cyanoborohydride (4.88 g, 77.8 mmol) was added to a solution of 6-chloroindoline (5.9 g, 38.9 mmol) in acetic acid (100 mL). Gas evolution was evident at the beginning of the reaction. After stirring for 10 h, the solution was diluted with water (100 mL) and 6 N NaOH was added until the pH of the reaction mixture was 12-13. The resulting mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 200 mL), and the combined organic layers dried over MgSO<sub>4</sub>. Flash column chromatography on silica gel (35% EtOAc/hexanes) yielded 2.3 g (39%) of a clear liquid: <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ 2.87 (t, J = 8.4 Hz, 2H), 3.44 (t, J = 8.4 Hz, 2H), 6.45 (d, J = 1.8 Hz, 1H), 6.47 (dd, J = 1.8, 7.6 Hz, 1H), 6.96 (d, J = 7.3 Hz, 1H).

To a 3-L, three-necked, round bottom flask equipped with a reflux condenser was added methyl-2-amino-5-bromobenzoate (7, 125 g, 543 mmol), copper cyanide (56.2 g, 624 mmol), and NMP (1 L). The reaction was heated to 200 °C and stirred for 4 h under nitrogen. The dark brown reaction mixture was allowed to cool and a brown precipitate was formed. The mixture was poured into a 16-L beaker containing sodium cyanide solution (1 kg NaCN in 6 L H<sub>2</sub>O) followed by the addition of EtOAc (4 L). The precipitate was dissolved by agitation and the layers were separated. The aqueous layer was extracted with EtOAc (2 x 1.5 L) and the combined organic layers were washed with 10% NaCN solution (2 L), H<sub>2</sub>O (2 L) and then dried over MgSO<sub>4</sub>. The light brown solution was concentrated and then dried in a vacuum oven overnight.

The ester was then dissolved in EtOH (2 L) and added to a 3-L round bottom flask followed by KOH solution (96.7 g of KOH in 500 mL H<sub>2</sub>O). The reaction mixture was heated to 50 °C and stirred for 2 h. The resultant dark-brown solution was poured into a chilled 2 N HCl solution (1.5 L), creating a yellowish precipitate. The solid was collected on a sintered glass filter frit, washed with cold water, and dried at 35 °C in a vacuum oven overnight to give 64.0 g (73%) of 5-cyanoanthranilic acid: <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  7.16 (d, J = 8.7, 1H), 7.79 (dd, J = 2.5, 8.7, 1H), 7.87 (d, J = 2.4, 1H).

$$H_2N$$
 $HO$ 
 $CN$ 
 $CN$ 
 $CN$ 
 $CN$ 
 $CN$ 
 $CN$ 
 $CN$ 

10

5

To a dry 4-L round bottom flask was added 5-cyanoanthranilic acid (9, 64 g, 395 mmol), triphosgene (39.4 g, 131 mmol) and dioxane (2 L). The suspension was stirred under  $N_2$  and heated to reflux. The reaction mixture became homogeneous after stirring at reflux for 2 h. As the carbonylation product was formed, white precipitate appeared in the solution. After stirring at reflux for an additional 3 h, the reaction was cooled to room temperature, filtered, and the precipitate washed with ether. The solid was dried in the vacuum oven to afford 5-cyanoisatoic anhydride (10, 51.5 g, 68%) as a pale yellow solid:  $^1$ H NMR (DMSO- $d_6$ )  $\delta$  6.86 (d, J = 9.3, 1H), 7.55 (dd, J = 2.6, 9.0, 1H), 8.04 (d, J = 2.4, 1H).

20

25

15

**3-Iodobenzoyl Chloride.** To a suspension of *m*-iodobenzoic acid (**21**, 5.0g, 20.1 mmol) suspended in CH<sub>2</sub>Cl<sub>2</sub> (60 mL) was added DMF (2 drops), followed by oxalyl chloride (20.1 mL of a 2 M solution in CH<sub>2</sub>Cl<sub>2</sub>, 40.2 mmol) under nitrogen atmosphere. After stirring for 18 h, the reaction mixture was nearly homogeneous.

Acid chloride 22 was then concentrated, azeotroped with toluene (2 x 25 mL), and placed on a high vacuum.

## Example 10: Additional Compounds Useful For Sterilization, Sanitation,

## 5 Antisepsis, and Disinfection

The following compounds may be synthesized using the methodology described above or via methods known in the art.

Compound No., Structure	Compound No., Structure
L-140191	L-140209
Br COOH  NH  SO <sub>2</sub>	Br COOH OCH <sub>3</sub>
L-140209A	L-140229
Br COOH OCH <sub>3</sub>	Br COOH  NH  SO <sub>2</sub> NH
L-140229A	L-140240
Br COOH  NH  O  SO <sub>2</sub> CF <sub>3</sub>	Br COOH NH SO <sub>2</sub> NH

Compound No., Structure	Compound No., Structure
L-159112	L-159113
CI OH OH O=S=O N	Br OH OSS=O N
L-159117	L-159120
OH ONH OSSO OSSO ON ON ON ON ON ON ON ON ON ON ON ON ON	
L-159121	L-159124
O NH OH  O=S=O  N	

Compound No., Structure	Compound No., Structure
L-159127	L-159128
CI ONH OH	O NH OH
0=\$=0 NH	O=S=O NH
L-159131	L-159133
O NH OH	O NH OH
O=S=O HN	O=S=O HN
F F	F

Compound No., Structure	Compound No., Structure
L-159139	L-159157
O NH OH  O=S=O NH  FF	CI ON OH ON ON ON OH ON ON OH ON ON OH ON ON OH ON ON OH ON ON OH ON OH ON OH ON OH ON OH ON ON OH ON ON OH ON ON OH ON OH ON
L-159157A	L-159158
CI ONH OH O=S=O	Br ONH OH O=S=O
L-159158A	L-159161
Br ONH OH O=S=O	O NH OH O=S=O CI

Compound No., Structure	Compound No., Structure
L-159161A	L-159164
O NH OH O=S=O N CI	O NH OH  O=S=O  N
L-159165	L-159168
O NH OH  O=S=O  N	O H OH OH

Compound No., Structure	Compound No., Structure
L-159171	L-159172
	Br O O O O O O O O O O O O O O O O O O O
L-159172A	L-159176
Br OHOH OHOH	

Compound No., Structure	Compound No., Structure
L-170148	L-170154
Br OH OH OH OH OH OH OH OH OH OH OH OH OH	Br ONH OH O=S=O NH
L-170160	L-170166
Br OH OH OH OH OH OH OH OH OH	Br ON DH OH OH OH OH OH
L-170166A	L-170178
O NH OH  O=S=O  N	Br ONH OH O=S=O HN

Compound No., Structure	Compound No., Structure
L-170178A	L-170184
Br ONH OH O=S=O HN	Br O O O O O O O O O O O O O O O O O O O
L-170190	L-170196
Br ONH OH O=S=O NOH OH	Br OH ONH OSS=ONH

Command No. Structure	Compound No., Structure
Compound No., Structure	
L-170210	L-170216
Br ONH OH	Br OH OH
L-181367	L-181368
Br ONH OH	
L-181370	L-181371
O NH OH  O=S=O NH  CI	O NH OH O S S O NH CI

Compound No., Structure	Compound No., Structure
L-181379	L-181382
Br OH OH OH OH OH OH CI	OH O
L-181383	L-181385
OHOONH O	Br ONH OH O=S=O CI NH
L-181388  O NH OH  O=S=O CI NH	L-181389  F O NH OH  O=S=O CI NH

Compound No., Structure	Compound No., Structure
L-181391	L-181392
Br ONH OH O=S=O NH	CI ONH OH O=S=O NH
L-181394	L-181395
O NH OH  O=S=O NH  Br	O NH OH  O=S=O NH  Br
L-181397	L-181398
Br ONH OH O=S=O NH	

Compound No., Structure	Compound No., Structure
L-181400	L-181401
O NH OH  O=S=O NH	O NH OH  O=S=O NH
L-181403	L-181404
Br ONHOH O=S=ONH	CI ONH OH O=S=O NH
L-181406	L-181407
	O H OH

Compound No., Structure	Compound No., Structure
L-181409	L-181410
Br ONH OH O=S=O NH Br	O NH OH  O S = O  NH  Br
L-181412	L-181413
O NH OH  O=S=O NH  Br	NH OH  O=S=O NH  Br
L-181419	L-181421
O NH OH  CIO=S=O NH	O NH OH  O=S=O NH  CI

Compound No., Structure	Compound No., Structure
L-181423	L-181424
O NH OH  O=S=O NH	O NH OH  O=S=O NH
CI	CI
L-181427	L-181429
NH OH OH	O NH OH  O=S=O NH  CI
L-181430	L-181432
O NH OH  O=S=O NH  CI	O NH OH  O=S=O NH  CI

Compound No., Structure	Compound No., Structure
L-181433	L-181435
O NH OH O=S=O NH CI	O NH OH  O=S=O CI NH
L-181436	L-181438
CI ONH OH O=S=O CI NH	O NH OH  O=\$=0 CI NH
L-181439  O NH OH  O=S=O CI NH	L-181442

Compound No., Structure	Compound No., Structure
L-181445	L-186544
O=S=O NH CI	O NH OH  O S O HN  O HN  O HN  O HN  O H
L-199155	L-199156
Br ONH OH	OHO NH  N-S=0
L-199157	L-199158
ONH OH  O=S=O  N	Br ONH OH

Compound No., Structure	Compound No., Structure
L-199159	L-199160
O NH OH  O=S=O  N	O NH OH  O=S=O  N O-N O
L-199161	L-199162
O NH OH  O=S=O  N OH	O_NH. OH  O=S=O N
L-199163	L-199164
O NH OH  O=S=O  N	O NH OH  O=S=O N

Compound No., Structure	Compound No., Structure
L-199165	L-199166
O NH OH  O=S=O  N	Br ONH OH
L-199167	L-199168
O NH OH  O=S=O  CI	Br OH OH OH
L-199169	L-199170
Br ONH OH	Br ONH OH O=S=O

Compound No., Structure	Compound No., Structure
L-199171	L-199172
O NH OH	O NH OH
O=S=O HO N	CI O=\$=0 N
L-199173	L-199174
Br ONH OH OH OH OH NH	Br ONH OH

Compound No., Structure	Compound No., Structure
L-199175	L-199176
O NH OH  O=S=O  CI	O NH OH  O=S=O N
L-199177	L-199178
O NH OH	O NH OH
	оно=s=0 N
L-199179	L-199180
O NH OH  HO O S=O	O NH OH  O=S=O

Compound No., Structure	Compound No., Structure
L-199181	L-199182
O NH OH  O=S=O	ONH OH  O=S=O
L-199183	L-199184
O NH OH	O NH OH
O=S=O	0=S=0

Compound No., Structure	Compound No., Structure
L-199185	L-199186
Br ONH OH HO O=S=O	O NH OH O=S=O
L-199187	L-199188
Br ONH OH	Br ONH OH
L-199189  Br  ONH OH  O=S=O N	L-199190  Br  ONH OH  O=S=O N

Compound No., Structure	Compound No., Structure
L-199191	L-199192
Br ONH OH	Br ONH OH O=S=O OH
L-199193	L-199194
Br ONH OH O=S=OO NOH	Br ONHOH

Compound No., Structure	Compound No., Structure
L-199195	L-199196
O NH OH  O=S=O  CI	Br ONH OH
L-199197	L-199198
Br OH OH OH OH OH OH OH OH OH OH OH OH OH	Br ONH OH

Compound No., Structure	Compound No., Structure
L-199199	L-217790
Br ONH OH O=S=O NH	O NH OH  O=S=O  CI
L-217791	L-218343
O NH OH  O=S=O  N	ONH OH OH

Compound No., Structure	Compound No., Structure
L-502902	L-502903
HO H	HO HN O
L-502904	PHA-500140
HO NAPARO	OH H OH OH OH OH OH OH OH OH O
PHA-500152	PHA-500200
О ОН Н О П О П О П О П О П О П О П О П О П О	OH OH OH SHOW N
PHA-500218	PHA-500219
OH OF N	CI OH OF SHOCI

Compound No., Structure	Compound No., Structure
PHA-500230	PHA-500236
OH ON S	OH O
PHA-500248	PHA-500284
OH NH O= O= N O= N	OHOHOHOHOO
PHA-502605	PHA-502606
OH OH	OH OH
PHA-520185	PHA-520200
CI ONH OH	O NH OH

Compound No., Structure	Compound No., Structure
PHA-520221	PHA-520245
O NH OH	O NH OH
PHA-520412	PHA-520413
CI————————————————————————————————————	CI ONH OH
PHA-520414	PHA-520416
CI ONH OH	O NH OH
PHA-523506	PHA-523507
OH OH	CI O OH CI

Command No. Structure	Compound No., Structure
Compound No., Structure	PHA-523509
PHA-523508	CI O OH CI
PHA-523510	PHA-523511
CI ONH OH	CI ONH OH
PHA-523512	PHA-523513
HO CI	N O OH CI
PHA-523514	PHA-523515
PHA-523516	PHA-523517
L .	CI CI OH

	Compound No., Structure
Compound No., Structure	
PHA-523518	PHA-523519
HO CI	· HO CI
PHA-523520	PHA-523521
HO HN CI	N O OH
	PHA-524553A
PHA-524545E	Br
Br OH N	$H_2N$ $O=S=O$ $HCI$
7774 525500	PHA-525501
PHA-525500  OH  Br  H  OH  OH  OH  OH  OH  OH  OH  OH  OH	O—————————————————————————————————————

Compound No., Structure	Compound No., Structure
PHA-525502	PHA-525503
O————O HN	
HO Br	HO Br
PHA-525504	PHA-525505
O HN HO O Br	HO
PHA-525506	PHA-526641
HO HN O	Br — H O NH <sub>2</sub>

Compound No., Structure	Compound No., Structure
	PHA-526648
PHA-526643  Br OH OH NH ONH NH NH NH NH <sub>2</sub>	PHA-320046  Br  ONH OH
PHA-526650  Br  ONH OH	PHA-526651  Br ONH OH Br SSO
PHA-526652  Br ONH OH	PHA-526653  Br ONH OH N=

Compound No., Structure	Compound No., Structure
PHA-526655	PHA-526660
Br O O O O O O O O O O O O O O O O O O O	Br OH H ON Br
F F	PHA-526679
PHA-526661  Br ONH OH F	O NH OH
	PHA-526683
PHA-526681  CI  N  N H HO	Br O OH OH NH O NH O NH O NH O NH O NH O

Compound No., Structure	Compound No., Structure
PHA-526705	PHA-526707
HN HO Br	O O O HN HO Br
PHA-526712	PHA-530914
HO Br HN O	HO HN O OH
PHA-530915	PHA-533232
HO HN O OH	O HN HO Br

Compound No., Structure	Compound No., Structure
PHA-533237	PHA-533243
-0 0	0—0 0—HN
Br Br	PHA-533247
PHA-533244	HO HO Br
PHA-533249  O	PHA-533252  N=OHN HO Br
PHA-533253  O  N  HN  HO  Br	PHA-533257  O O HN HO Br

Compound No., Structure	Compound No., Structure
PHA-533258	PHA-533259
HO HO Br	Вг
PHA-533261	PHA-533262
O O O HN HO Br	HO HN O
PHA-533264	PHA-533265
N—O HN HO Br	O O HN HO Br

Compound No., Structure	Compound No., Structure
PHA-533266	PHA-533268
HO HN O	HN O NH
PHA-533269	PHA-533272
HO HN O	HO HN O
PHA-533273	PHA-533274
O O O HN HO Br	O HN HO Br

Compound No., Structure	Compound No., Structure
PHA-533275	PHA-533276
	HO Br
HN HO Br	
PHA-533278	PHA-533281
HN HO Br	HO HO Br
PHA-533282	PHA-533285
N—O HN HO	HN HO

Compound No., Structure	Compound No., Structure
PHA-533286	PHA-533289
	H
HO HO DE	HN HO Br
PHA-533290	PHA-533401
HN O	ON HOO
	DIIA 527095
PHA-537084	PHA-537085
O N N OH	
least retained isomer by RP-HPL(	O NH OH

Compound No., Structure	Compound No., Structure
PHA-537089	PHA-537090
least retained isomer by RP-LC/MS	least retained isomer by RP-LC/M:
PHA-537091	PHA-537092
CI HN O NH OH CI	HN OH OH
PHA-537098	PHA-537099
CI OH HIN	CI OHN HN

Compound No., Structure	Compound No., Structure
PHA-537100	PHA-537101
OH HN O	OH HN O
PHA-537106	PHA-537110
O O O O O O O O O O O O O O O O O O O	CI ONH OH
PHA-537112	PHA-537114
CI ONH OH	O CI

Compound No., Structure	Compound No., Structure
	PHA-537122
PHA-537121	O NH OH
727100	PHA-537133
PHA-537128  CI O HN O OH	OH HNO
PHA-537138	PHA-537139  O  H  HO  O
о он РНА-537142	PHA-537143
N O OH	O O O O O O O O O O O O O O O O O O O

Compound No., Structure	Compound No., Structure
PHA-537144	PHA-537150
HO CI HN O	O=S=O OH OH
PHA-537152	PHA-537155
O OH CI	O H O O O O O O O O O O O O O O O O O O
PHA-537157	PHA-537158
N OH OH OH CI	HO CI
PHA-537162	PHA-537202
O=S=O OH CI	most highly retained isomer by RP-LC/MS

Compound No., Structure	Compound No., Structure
PHA-537203	PHA-537204
most highly retained isomer by RP-LC/MS  PHA-538016	most highly retained isomer by RP-LC/M  PHA-539146
NH O	Br NH <sub>2</sub>
PHA-539148	PHA-539149
Br ON HOO NH <sub>2</sub>	OH HN ON NH SSOO

Compound No., Structure	Compound No., Structure
PHA-539150	PHA-539152
Br ONH OH	ONH OH  ONH Br
PHA-539153	PHA-539154
Br ONH OH	Br ON NH ON NH NN NN NN NN NN NN NN NN NN NN NN NN
PHA-539155	PHA-539156
Br ONH OH O=S-N	Br NH OH F

Compound No., Structure	Compound No., Structure
PHA-539164	PHA-539168
Br O OH OH ONH ON NH	Br OH OH OH ONH ON NH OH
PHA-539169	PHA-539170
Br OHO O N Br	O Br
PHA-539171	PHA-539172
ON HOO	O=S=O CI

Compound No., Structure	Compound No., Structure
PHA-539174	PHA-539175
PHA-3391/4  Br  ONH OH  ONH OH  CI	Br ONH OH
PHA-539177	PHA-539179
Br O NH OH	F CI O S HN HO Br
PHA-539180  Br  ONH OH	PHA-539181  Br O NH OH O O S O O O O O O O O O O O O O O O

Compound No., Structure	Compound No., Structure
PHA-539183	PHA-539186
Br ONH OH	Br O O O O O O O O O O O O O O O O O O O
PHA-539187	PHA-539188
Br O O O O O O O O O O O O O O O O O O O	O NH OH  O=S=O  O O O O O O O O O O O O O O O O
PHA-539190	PHA-539193
O NH OH  O NH OH  CI	ON HO HO

Compound No., Structure	Compound No., Structure
PHA-539194	PHA-539195
Br O O O O O O O O O O O O O O O O O O O	Br O O O O O O O O O O O O O O O O O O O
PHA-539197	PHA-539198
Br O OH OH NH O S NH Br	O NH HOOO

Compound No., Structure	Compound No., Structure
PHA-539199	PHA-539203
PHA-339199  ON NH  ON N	Br ONH OH
PHA-539206	PHA-539207
Br ONH OH Br Sio	O NH OH  O STO CI
PHA-539208  F CI O N HN HO Br	PHA-539209  Br  ONH OH  CI NI OH

Compound No., Structure	Compound No., Structure
PHA-539234	PHA-539235
HO HN Br	HO HN F
PHA-539245	PHA-539246
HO HN N	HO HN Br
PHA-539247	PHA-539248
HO HN F	HO HN CI
PHA-539249	PHA-539250
HO HN CI	HO HN N
PHA-539251	PHA-539252
HO HN O	HO HN Br

Compound No., Structure	Compound No., Structure
PHA-539253	PHA-539254
HO HN Br	HO HN Br
PHA-539255	PHA-539256
HO HN O	HO HN Br
PHA-539257	PHA-539258
HO HN FFF	HO HN Br
PHA-539259	PHA-539260
HO HN N	O HN F
PHA-539262	PHA-539263
HO HN N	HO HN N

Compound No., Structure	Compound No., Structure
PHA-539264	PHA-539265
HO HN Br	HO HN Br
PHA-539266	PHA-539267
HO HN S	O HN S
PHA-539268	PHA-539269
HO HN O	HO HN F
PHA-539270	PHA-539271
HO HN Br	HO HN Br
PHA-539276	PHA-539277
HO HN Br	HO HN F

Compound No., Structure	Compound No., Structure
PHA-539278	PHA-539285
HO HN Br	HO HN O
PHA-539293	PHA-539294
HO HN Br	HO HN N=O
PHA-539295	PHA-539296
HO HN NO	HO HN S=O
PHA-539297	PHA-539298
HO HN CI	HO HN N=0
PHA-539302	PHA-539303
OH Br	HO HN F

Compound No., Structure	Compound No., Structure
PHA-539305	PHA-539307
HO HN Br	HO HN Br
PHA-539308	PHA-539310
HO HN Br	HO HN Br
PHA-539312	PHA-539313
HO HN Br	HO HN Br
PHA-539314	PHA-539317
HO HN F	HO HN F
PHA-539318	PHA-539322
HO HN CI	HO HN Br

Compound No., Structure	Compound No., Structure
PHA-539328	PHA-539329
F N N N N N N N N N N N N N N N N N N N	O NO HN HO Br
PHA-539332	PHA-539337
HO HN F	HOHOOH
PHA-539338	PHA-543684
HO HN O	DI O O NH OH CI O NH F

Compound No., Structure	Compound No., Structure
PHA-543685	PHA-543686
Br ONH OH	Br O O O O O O O O O O O O O O O O O O O
PHA-543689	PHA-543690
Br OH O N O S N O S N	CI HO Br
PHA-543692	PHA-543693
Br ONH OH F N N N N N N N	O NH OH  Br

Compound No., Structure	Compound No., Structure
PHA-543695	PHA-543698
ONH OH  CI  F	Br NH <sub>HO</sub> O NH <sub>2</sub>
PHA-543700	PHA-543701
ONH OH ONH OH ON NH Br	ONH OH ON Br

Compound No., Structure	Compound No., Structure
PHA-543706	PHA-543708
O NH OH	HO HN N'.N
O S N O F F F	
PHA-551625	PHA-551672
Br OH ONH ONH ON	Br OH ONH O
PHA-551675	PHA-551716
ON NH ON NH OH	ONH F OH

Compound No., Structure	Compound No., Structure
PHA-556420	PHA-563330
O NH O	HOHN
PHA-563331	PHA-563333
HO O HN O O O	HO HN N
PHA-563335	PHA-563340
HOHON	HO HN F
PHA-563341	PHA-563342
HO HN N	HOHON

Compound No., Structure	Compound No., Structure
PHA-563344	PHA-563345
HO HN N	HO HN N
PHA-563347	PHA-563350
HOHO	HO HN N
PHA-563351	PHA-563353
HO HN N	HO HN N
PHA-563354	PHA-563360
HO HN	HOHON

Compound No., Structure	Compound No., Structure
PHA-563363	PHA-563364
HO HN N	OH HN N
PHA-563365	PHA-563366
HOHO	HO HN N
PHA-563368	PHA-563370
F N N N	HO HN N
PHA-563371	PHA-563375
HO HN N	HO HN P

00055 051

Compound No., Structure	Compound No., Structure
PHA-563378	PHA-563386
HO HN O	HO HN F F
PHA-563388	PHA-563389
HO HIN O	F F O H HO Br
PHA-563390	PHA-563391
HO HN O	NH O OH Br
PHA-563392	PHA-563393
O=N+ HO O Br	O H HO O Br
PHA-563394	PHA-563396
F O HO O Br	HO HN O N

Compand No. Co.	
Compound No., Structure	Compound No., Structure
PHA-563397	PHA-563398
F OH OH	HO HN Br
PHA-563399	PHA-563401
O=N+OH OBr	HO HN N
PHA-563404	PHA-563406
O NH O OH	NH OH Br
PHA-563407	PHA-563408
HO HN O H	HO HN O
PHA-563409	PHA-563411
HO HN O N	HO HN O N

Compound No., Structure	Compound No., Structure
PHA-563413	PHA-563415
HO—Br	HO HN O
PHA-563417	PHA-563419
HO HN O N	HO O Br
PHA-563420	PHA-563426
NH OH Br	HO HN O
PHA-563427	PHA-563440
HO HN O	NH OH
PHA-563441	PHA-563442
HO Br	HO HN O

Compound No., Structure	Compound No., Structure
PHA-563449	PHA-569976
	OH OH
PHA-571150	PHA-571151
HO HN S	HO HN N
PHA-571152	PHA-571153
HO HN N	HO HN Br
PHA-571154	PHA-571155
HO HN F	HO HN CI

Compound No., Structure	Compound No., Structure
PHA-571156	PHA-571157
HO	HO HN N
PHA-571160	PHA-571161
OH OH OH OH OH OH OH OH OH OH OH OH OH O	HO N SOO
PHA-571162	PHA-571164
OH OH ONH ONH ONH ONH ONH ONH ONH ONH ON	N O O NH O S=O
PHA-571167	PHA-571169
OH ONH ONH OSSO I	N N N N N N N N N N N N N N N N N N N

Compound No., Structure	Compound No., Structure
PHA-571170	PHA-571172
N H O O O O O O O O O O O O O O O O O O	ON SOUTH ON THE PART OF THE PA
PHA-571174	PHA-571176
$N = - \begin{pmatrix} 0 \\ N \\ - N \\ $	
PHA-571182	PHA-571183
$N = \begin{array}{c} HO \\ O \\ N \\ O \\ \end{array}$	N=
PHA-571186	PHA-571188
OH OH OH	

Compound No., Structure	Compound No., Structure
PHA-571189	PHA-571194
N N N N N N N N N N N N N N N N N N N	HO HO
PHA-571196	PHA-571197
OH OH	ON S O OH OH
PHA-571198	PHA-571199
OH ONH OSIO	HO N N N N N N N N N N N N N N N N N N N

Compound No., Structure	Compound No., Structure
PHA-571202	PHA-571203
	N N S N O O O O O O O O O O O O O O O O
PHA-571205	PHA-571207
	NH OH NH ON
PHA-571208	PHA-571214
N H O O O O O O O O O O O O O O O O O O	N S O O O O O O O O O O O O O O O O O O

Compound No., Structure	Compound No., Structure
PHA-571215	PHA-571216
OH OH	H O O O O H
PHA-571219	PHA-571224
OH O	NH O OH
PHA-571226	PHA-571228
OH ONH OS=O OH	$N = - \begin{pmatrix} 0 \\ N \\ N \\ HO \end{pmatrix}$
PHA-571230	PHA-571231
F N O O O O O O O O O O O O O O O O O O	N=

- 211 -

Compound No., Structure	Compound No., Structure
PHA-571232	PHA-571234
N N N N N N N N N N N N N N N N N N N	$N = \begin{array}{c} HO \\ O \\ N \\ O \\ \end{array}$
PHA-571235	PHA-571237
HO————————————————————————————————————	N N N N N N N N N N N N N N N N N N N
PHA-571238	PHA-571239
N O O O O O O O O O O O O O O O O O O O	N S S S S S S S S S S S S S S S S S S S
PHA-571240	PHA-571241
N N N N N N N N N N N N N N N N N N N	O=S N OH OH

Compound No., Structure	Compound No., Structure
PHA-571242	PHA-571243
$N = \bigvee_{\substack{O \\ N \\ HO}} \bigvee_{\substack{N \\ O \\ O}} \bigvee_{\substack{N \\ O \\ O \\ O \\ O}} \bigvee_{N \\ O \\ $	OH ONH ONH ONH ONH ONH ONH ONH ONH ONH O
PHA-571246	PHA-571249
HO O SO SO SO	OH ONH OSSIO
PHA-571253	PHA-571255
$N = \bigvee_{\substack{N \\ N \\ N}} \bigvee_{\substack{N \\ N \\ N}} \bigvee_{\substack{N \\ N \\ N \\ N \\ N}} \bigvee_{N \\ N \\$	N N N N N N N N N N N N N N N N N N N
PHA-571257	PHA-571258
N S S S S S S S S S S S S S S S S S S S	OH OH

Compound No., Structure	Compound No., Structure
PHA-571260	PHA-571262
OH ONH OSSO OH	N N N N N N N N N N N N N N N N N N N
PHA-571263	PHA-571264
OH OH OH	NH OH OH
PHA-571265	PHA-571267
$\begin{array}{c c} & & & & \\ & & & & \\ & & & & \\ & & & & $	F F F N N S N S N S N S N S N S N S N S
PHA-571269	PHA-571270
OH OH	OH ONH ONH ONH ONH ONH ONH ONH ONH ONH O

Compound No., Structure	Compound No., Structure
PHA-571271	PHA-571272
ON SHOOH	OH ONH ONH ONH ONH ONH ONH ONH ONH ONH O
PHA-571273	PHA-571280
OH O	O H HO O Br
PHA-571281	PHA-571282
O H HO O Br	O H HO O Br
PHA-571283	PHA-571285
O H HO O Br	HO—Br

Compound No., Structure	Compound No., Structure
PHA-571287	PHA-571289
Br O N O N O	HO HN O
PHA-571292	PHA-610940
HO HN O	OF 3 OF 3
PHA-610941	PHA-610942
OCH <sub>3</sub>	OH O

Compound No., Structure	Compound No., Structure
PHA-630426	PHA-656807
N=OH NH O <sub>2</sub> S	HO HO
PHA-656808	PHA-656809
HO ON SO ON	OH N S O OH OH OH
PHA-656810	PHA-656811
ON SHOOM ON THE PROPERTY OF TH	NH OH OH

Compound No., Structure	Compound No., Structure
PHA-656820	PHA-656859
CI OH ON NOTICE OF THE PARTY OF	HO NS O NH O OH
PHA-656860	PHA-656861
NH HO HO HO NO HO	HO O O O O O O O O O O O O O O O O O O
PHA-656862	PHA-656863
OH ON N N N OH OH N	OH ONH OH OH OH OH OH OH

Compound No., Structure	Compound No., Structure
PHA-656866	PHA-656867
OH ONH OS NH	OH ON ON OH OH OH
PHA-656868	PHA-656870
HO N N N O NH O O NH	NH OH OH
PHA-656871	PHA-656872
NH OH OH	

Compound No., Structure	Compound No., Structure
PHA-656880	PHA-656882
CI OH ONH ON N	CI OH ONH O
PHA-656883	PHA-656884
CI OH ONH O	CI OH ONH O
PHA-656885	PHA-656886
CI ONH ONH O	ON OH

Compound No., Structure	Compound No., Structure
PHA-656887	PHA-656888
CI ONH O	CI OH
PHA-656889	PHA-656890
CI OH NHO	HO HN O
PHA-656891	PHA-656892
HO HN	F O O HN CI

Compound No., Structure	Compound No., Structure
PHA-656893	PHA-656894
F O NH OH	CI OH ONH O
PHA-662253	PHA-662254
N=OH OH CH2-SO2-N	$N \equiv \bigcirc$ OH $\bigcirc$
PHA-662412	PHA-679756
N OH NH	HO-N H ON S-NO
PHA-679759	PHA-687570
O-N OH ON S-NO	HO OH NH O <sub>2</sub> S N

Compound No., Structure	Compound No., Structure
PHA-708922	PHA-708977
OH OH SIN NO SIN	O HN O
	HO Br
PHA-708979	PHA-708987
HN O HN O HO Br	HN O HO Br
PHA-713389	PHA-713390
CI N HN O HO Br	CI NEW HOOM HOOM HOOM HOOM HOOM HOOM HOOM HOO

Compound No., Structure	Compound No., Structure
PHA-713391	PHA-713392
CI SHN HN HO HO Br	S CI HN O HN O HO Br
PHA-713393	PHA-713395
HN O HN O HO Br	HE O HO H

Compound No., Structure	Compound No., Structure
PHA-713397	PHA-738531
HN O	N=OH ON SON
O HN O HO Br	
PHA-738532	PHA-740499
N=OH OUNCE NO	N=OHONS-N
PHA-748361	PNU-276556
COOH ONH	Br OCH <sub>3</sub>

Compound No., Structure	Compound No., Structure		
PNU-276672	PNU-276873		
Br ONH OH O=S=O HN CI	Br COOH OCH <sub>3</sub>		
PNU-281164	PNU-282858		
F O OH	Br OH NH CI		
PNU-282859	PNU-282860		
Br OH NH N N NH	Br OH NH NH N		

Compound No., Structure	Compound No., Structure
PNU-290881A	PNU-291997
O OH H CI NH <sub>2</sub> HCI	Br OH OH
PNU-292577	
Br OH ONH O	

## **Example 11: ACTIVITY DATA**

## MIC Test Method

5

10

The *in vitro* MICs of test compounds were determined by a standard agar dilution method. A stock drug solution of each analog was prepared in the preferred solvent, usually DMSO:H<sub>2</sub>O (1:3). Serial 2-fold dilutions of each sample are made using 1.0 ml aliquots of sterile distilled water. To each 1.0 ml aliquot of drug was added 9 ml of molten Mueller Hinton agar medium. The drug-supplemented agar was mixed, poured into 15 x 100 mm petri dishes, and allowed to solidify and dry prior to inoculation.

Vials of each of the test organisms are maintained frozen in the vapor phase of a liquid nitrogen freezer. Test cultures are grown overnight at 35°C on the medium

5

10

appropriate for the organism. Colonies are harvested with a sterile swab, and cell suspensions are prepared in Trypticase Soy broth (TSB) to equal the turbidity of a 0.5 McFarland standard. A 1:20 dilution of each suspension was made in TSB. The plates containing the drug supplemented agar are inoculated with a 0.001 ml drop of the cell suspension using a Steers replicator, yielding approximately 10<sup>4</sup> to 10<sup>5</sup> cells per spot. The plates are incubated overnight at 35°C.

Following incubation the Minimum Inhibitory Concentration (MIC  $\mu$ g/ml), the lowest concentration of drug that inhibits visible growth of the organism, was read and recorded. The data is shown in Tables I and II.

Table 1: Activity Data

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
L-217792	8	PHA-500334	16
		O=S=O CF <sub>3</sub> CF <sub>3</sub>	
PHA-501684	1	PHA-502339	2
Br—S		Br—S—N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-501685	2	PHA-502339A	8
Вг		Br—N HCI	
PHA-501748	2	PHA-509059	0.5
Br—OH S—OH S—OH		Br O OH	
PHA-504639	4	PHA-513535	2
Br—OH OCH <sub>3</sub>		Вг	
PHA-515448	2	PHA-513541	64
Вг		Br—OH OOO	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	•	MIC
PHA-515585	1	PHA-515583	8
Br OH OH		$O_2N$ $O_2N$ $O_3$ $O_4$ $O_5$ $O_5$ $O_5$ $O_5$ $O_5$ $O_5$ $O_7$ $O_$	
PHA-516113	2	PHA-516112	8
Br OH OH SO <sub>2</sub>		Br OH SO <sub>2</sub>	
PHA-519402	0.5	PHA-516116	0.5
NO <sub>2</sub>		Br OH OH	
PHA-521534	1	PHA-518226	2
Br—OH HO		CI—OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
-	MIC		MIC
PHA-522145	32	PHA-520446	16
Br — H		Br—H	
PHA-524523	0.125	PHA-520447	1
		Вг	
PHA-524545	0.25	PHA-520938	1
Br—OH NAMOH		Br OH OH O <sub>2</sub> S N	
PHA-526580	1	PHA-521535	>128
Br—OH S—OH		Br OCH <sub>3</sub> NECH <sub>3</sub>	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 140., Structure	MIC	Compound 1 (c., 2010)	MIC
PHA-530687	8	PHA-522146	0.5
N= OH OH OH		Br OH NOCH3	
PHA-535548	0.25	PHA-524524	1
N MOH		N O O O O O O O O O O O O O O O O O O O	
PHA-535549	0.25	PHA-526578	2
N MOCH3		Br—OH	
PHA-535553	1	PHA-530685	32
N OH OH		N OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-543140	1	PHA-530989	4
N=OH OH O <sub>2</sub> S <sub>N</sub>		Br—Sh	
PHA-546926	0.5	PHA-543139	0.125
N OH OH		N OH OH OCH <sub>3</sub>	
PHA-547267	0.125	PHA-543141	0.125
N S OH		N=OH NH O <sub>2</sub> S N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	*	MIC
PHA-552831	1	PHA-543681	0.125
SO <sub>2</sub>		N OH OH OZS N CI	
PHA-556214	1	PHA-555027	1
O=S=O N OH		N=OH OSI -N	
PHA-556658	8	PHA-556657	2
HO HN CI		HO HN CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
•	MIC		MIC
PHA-556663	8	PHA-556661	8
HO HN CI		HO HE NO CI	
PHA-561055	1	PHA-557035	4
OH OH		HO HN CI	
PHA-562733	0.25	PHA-562731	1
N O O O O O O O O O O O O O O O O O O O		P O OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
•	MIC		MIC
PHA-562862	4	PHA-562745	0.25
NH OH O <sub>2</sub> S		O OH O S N	
PHA-562863	2	PHA-563275	2
N=OH NH O <sub>2</sub> S-N:		HO HN C CI	
PHA-563274	2	PHA-563277	2
HO HN CI		HO HN CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound No., Structure		Compound 1 to, 2 statement	MIC
	MIC		
PHA-563276	2	PHA-563279	0.5
HO HN CI		HO HN CI	
PHA-563278	2	PHA-563281	1
HO HN CI		HO HN CI	
PHA-563280	1	PHA-563283	16
F <sub>3</sub> C CF <sub>3</sub> HO HN O CI		HO HN CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
_	MIC		MIC
PHA-563282	1	PHA-563285	2
HO HN CI		HO HN CI	
PHA-563284	2	PHA-564215	0.5
HO HN CI		OCH <sub>3</sub>	
PHA-563324	>128	PHA-564750	0.25
HO HEN CI		N OH OH OLD	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-564218	1	PHA-566948	1
O OH		G-)-enantiomer	
O <sub>2</sub> N PHA-566947	0.5	PHA-568197	16
Br—OH HO (+)-enantiomer		CI N—SO <sub>2</sub> O <sub>2</sub> NOH	
		6.3/93.7 trans/cis	
PHA-568196	1	PHA-568205	2
98/2 mixture of trans/cis		HO HN CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-568206	2	PHA-568376	16
HO HN CI		Br OH SO <sub>2</sub>	
PHA-568378	2	PHA-568420	0.5
Br OH OH O <sub>2</sub> S N		N=OH NH O <sub>2</sub> S, N	
PHA-568461	0.125	PHA-568422	0.125
N=OH NH O <sub>2</sub> S N		N=OH NH OLH <sub>3</sub> O CH <sub>3</sub> O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-568907	8	PHA-568424	1
		N OH  OH  OLY  OLY  OLY  OLY  OLY  OLY  OL	
PHA-569044	0.25	PHA-568425	8
N=OH S		HO OHO OH	
PHA-569064	1	PHA-568906	8
N OH OH OZS N		OCH <sub>3</sub> N OCH <sub>3</sub>	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 110., Structure			MIC
	MIC		
PHA-569887	0.25	PHA-569044A	0.5
CI—N SO <sub>2</sub>		O OH S HCI	
Trans			
PHA-569977	16	PHA-569077	1
N OH OH CI		N=OH NH O <sub>2</sub> S	
PHA-570949	1	PHA-569885	16
CI————————————————————————————————————		This is 97.9/2.1 cis/trans	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
,	MIC		MIC
PHA-571396	4	PHA-569974	1
Br OH OH O2S NO		N= OH O	
PHA-571458	4	PHA-570008	0.125
N S OH		N= OH NH SO <sub>2</sub> NCI	
PHA-615551	1	PHA-570042	2
CI————————————————————————————————————		N= OH OH	
PHA-630427	4	PHA-571395	4
N OH OH		CI OH OH O <sub>2</sub> S N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	-	MIC
PHA-630852	4	PHA-571397	4
N=COOH N=SO <sub>2</sub> -N		O <sub>2</sub> N OH OH O <sub>2</sub> S N	
PHA-630966	0.25	PHA-610938	1
N=OHONSON		OH OH	
PHA-630989	4	PHA-630368	0.5
Br—N		N=OH S	
PHA-662430	1	PHA-630726	4
NH O, S, O		N= N tBu	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 1 to, Structure	MIC	Compound too, Sudden	MIC
DVI (60051		DIA (200/5	
PHA-662951	32	PHA-630965	0.25
N=OH OH ODE		N=OHO S N	
PHA-666124	32	PHA-631082	0.25
он о ў — N		Br OH ON Br	
PHA-681768	1	PHA-662250	1
OHC OH		CI—OH ON SINGLE	
PHA-686834	4	PHA-662431	1
HO-N OH		NH ON S'N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-707801	4	PHA-664658	4
CI SO <sub>2</sub>		CI—OH ON SIN	
PHA-708976	32	PHA-670083	0.5
O HN O HO Br		OH NH O=S-N NH <sub>2</sub>	
PHA-708980	16	PHA-682996	64
S HN O HN O Br		CH <sub>2</sub> O-N  CH <sub>2</sub> O-N  OH  OH  OH  OH  OH  OH  OH  OH  OH  O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound No., Structure		Compound No., Structure	
	MIC		MIC
PHA-708982	128	PHA-687511	4
NH O HO HO Br		O_S=O OCI is >99 trans	
PHA-708984	32	PHA-708923	32
O O O H N O HO Br		OH ON NOW NOW NOW NOW NOW NOW NOW NOW NOW	
PHA-708986	64	PHA-708978	32
O H H O HO Br		O H N H N H N H N H N H N H N H N H N H	·

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	-	MIC
PHA-708989	8	PHA-708981	16
		HN O HO Br	
PHA-708991	4	PHA-708983	32
		HZ H	
PHA-708993	4	PHA-708985	8

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-708995	0.125	PHA-708988	32
F F O O O O O O O O O O O O O O O O O O			
Br PHA-708997	8	PHA-708990	8
CC CI ON SON O HO H			
PHA-713387	128	PHA-708992	4
ON HO HO Br			

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-713394	128	PHA-708994	8
F O O O O O O O O O O O O O O O O O O O			
PHA-713398	4	PHA-708996	16
		F F O S S O O O O O O O O O O O O O O O	
PHA-713400	16	PHA-713386	128
HZ O HZ O HZ O HZ O HZ O HZ O HZ O HZ O		S N N N N N N N N N N N N N N N N N N N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-713403	64	PHA-713388	16
O-T-NO HN S HN O HO HO Br			
PHA-713406	64	PHA-713396	8
S HN HN HN HN HN HN HN HN HN HN HN HN HN		CI CI O D D D D D D D D D D D D D D D D D D	
PHA-713408	64	PHA-713399	16

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-713410	1	PHA-713401	32
CI		° <del></del>	
CI			
0=\$=0		HN O	
HN		HN O	
HO HN O		Q HŅ ↓O	
		но	
Br		Br	·
PHA-717196	4	PHA-713405	128
но		~	120
0		F	
Br—N		S NH F	
		HN	
N—CI			
		HO HN O	
		Br	
PHA-728844	0.25	PHA-713407	32
HO == 0		l s	
N=\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\			
		HN S	
O Z S N CI		HN	
· ·			
		HO HN O	
	!		
		Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PNU-263533		PHA-713409	1
Br OH OH		O, S, O HN HO HO Br	
PNU-271584		PHA-713411	32
Br OH		Br O H	
PNU-276296		PHA-719201	2
Br OH ON NOT NOT NOT NOT NOT NOT NOT NOT NOT		OH ON S-N	
PNU-276637		PHA-735753	16
Br OH OH		Br—N CH <sub>3</sub> O O S O CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PNU-276670		PNU-268205	
C21H17BrN2O5S Exact wt. 488.0042		Br OH OH OF THE STATE OF THE ST	
PNU-276817	4	PNU-275747	
Br OH SE COOH		N SO <sub>2</sub> OH	
PNU-276854		PNU-276301	
Br OCH, OCH, OCH,		Br OH OH ON NO	
PNU-276933		PNU-276638	
Br OCH <sub>3</sub>		Br OH OH	
		C18H17BrN2O6S Exact wt. 467.9991	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
•	MIC		MIC
PNU-276988	16	PNU-276728	2
Br ONH OH		C22H17BrN2O5S Exact wt. 500.0042	
PNU-277231	1	PNU-276770	
OH ONH ON NO <sub>2</sub>		Br OH	
PNU-280772		PNU-276818	
Br OCH <sub>3</sub>		Br OH ON	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
_	MIC		MIC
PNU-283076	1	PNU-276913	
O=S=O N CI		Br OH O N	
PNU-283599	1	PNU-276952	
O OH O H O NO <sub>2</sub> CI NO <sub>2</sub>		Br OH S OH S OH OH OH	
PNU-283603A	16	PNU-280727	
O OH O H O H O H O H O H O H O H O H O H		Br OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PNU-288969	0.25	PNU-282958	
O=S=O CI		O NH O NO <sub>2</sub> SO <sub>2</sub>	
PNU-290821	64	PNU-283318	0.125
O=S=O  CI  Br  OH  OH  CI		Br OH	
PNU-290877	>128	PNU-283371	4
O OH  Br		Br COOH	
See Comments			

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 110., Structure		Compound 140., Structure	
	MIC		MIC
PNU-290905	1	PNU-283601A	32
Br OH OH		O OH  CI  N OH  NOH  HCI	
PNU-290906	1	PNU-283604	4
Br OH NH CI		O H CI	
PNU-291061	16	PNU-289815	8
Br OH OH OH		O S O OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PNU-291410	4	PNU-290882	1
O H CI		OH OH OH	
PNU-291570	8	PNU-291010	1
CH <sub>3</sub> O O O O O O O O O O O O O O O O O O O		Br OH OH CI	
PNU-291571	0.5	PNU-291011	0.25
O S S O OH		Br OH NH S S S S S S	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	•	MIC
PNU-292070	2	PNU-291129	0.5
Br OH		Br OH NH	0.5
PNU-293032	16	PNU-291130	4
NO <sub>2</sub> OH ON		Br OH	
PNU-293905	8 .	PNU-291408	32
N=O=S=O		O O O H	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-630331	2	PNU-291517	2
NC OH OH ON		Br OH OH	
PNU-293795	32	PNU-291679	1
Des = 0	•	Br OH OH	
PNU-294595	16	PNU-292379	0.5
N SO <sub>2</sub> N CI		O=S=O  CI	
PHA-630330	0.5	PNU-293049	4
NC OH OH		Br S	

**Table 2: Activity Data** 

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
L-170210	16	L-170216	
Br NH OH		Br OH	
L-199199		L-217790	4
Br ONH OH		O NH OH O S S CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	Compound 140., Structure	1
L-217791	4	1.010040	MIC
SZIMATI OS OSSO OSSO OSSO OSSO OSSO OSSO OSSO		L-218343	16
L-502902	128	L-502903	16
HO H	·	HO HN O	
L-502904	64	РНА-500140	32

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-500152	32	PHA-500200	4
О		о У-он . N	
PHA-500218	64	PHA-500219	32
OH OF SECOND		OHOH OF SEO CI	
PHA-500230	>128	PHA-500236	8
OH OS N		OH O	
PHA-500248	8	PHA-500284	32
OH. OH. OH.		OH OH OH	
PHA-502605	8	PHA-502606	16
OHOH		OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	1	MIC
PHA-520185	8	PHA-520200	2
CI ONH OH		CI ONH OH	
PHA-520221	2	PHA-520245	128
O NH OH		CI O DH	
PHA-520412	4	PHA-520413	4
CI OH	·	O H OH	
PHA-520414	8	PHA-520416	4
CI ONH OH		O NH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
,	MIC	ompound 110., Structure	MIC
PHA-523506	32	PHA-523507	4
O O O O O O O O O O O O O O O O O O O	52	CI CI CI	4
PHA-523510	8	PHA-523508	8
O NH OH		O OH	
PHA-523511	8	PHA-523509	8
O NH OH		CI OH CI	·
PHA-523513	4	PHA-523512	4
N O OH		O HO CI	
PHA-523516	2	PHA-523514	4
CI NO OH		OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	Sampound I to i, Birdotalo	MIC
PHA-523517	4	PHA-523515	4
CI NO OH		CI OH OH	4
PHA-523518	8	PHA-524553A	>128
HO CI		H <sub>2</sub> N OH HCI	
PHA-523519	4	PHA-525501	8
HO CI		O HN Br	
PHA-523520	2	PHA-525503	2
HO HO CI		HN HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-523521	16	PHA-525505	16
THA-525521		HO Br	
PHA-524545E	0.5	PHA-525506	64
Br OH N		HO HIN O	
PHA-525500	4	PHA-526643  Br OH OH ONH ONH ONH NH <sub>2</sub>	64

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-525502	8	PHA-526650	2
HN HO Br		Br OH OH	
PHA-525504	16	PHA-526652	1
HIN HO Br		Br ONH OH	·

Compound No., Structure	SA 9218	Compound No., Structure	SA 0210
	MIC	Compound 110., Structure	SA 9218
PHA-526641	8	DUA 526655	MIC
Br HA-320041	8	PHA-526655  Br OH H ON N N N N N N N N N N N N N N N N	16
PHA-526648	0.25	PHA-526661	16
Br OH OH		Br OHNH OHN F	
PHA-526651	0.25	PHA-526681  CI  N O N H HO O	8

Γ	Compound No., Structure	SA 9218	Compound No. Start	T Q 4 . 2.2.1.2
	<u> </u>	MIC	Compound No., Structure	SA 9218
-	PHA-526653		DIVA GO COO	MIC
	PHA-526653  Br O NH OH OH NNH OH	>128	PHA-526705	16
F	PHA-526660	0	DIVA 60 CTA	
	O O O O O O O O O O O O O O O O O O O	8	PHA-526712  HO  HN  O  O  O  O  O  O  O  O  O  O  O  O  O	64
PI	HA-526679	32	PHA-530915	32
	O NH OH N		HO HN O	

Compound No., Structure	SA 9218	Compound No., Structure	T. G. A. G.
, , , , , , , , , , , , , , , , , , , ,	MIC	Compound No., Structure	SA 9218
PHA-526683	>128	DIA 522227	MIC
Br.	7120	PHA-533237	16
OH		\ <u>\</u>	
NH			
		HN	
		HO	
N <sub>S</sub> S <sub>0</sub> O-		ó′	
		ы	
		•	
PHA-526707	12	Divi	
	2	PHA-533244	4
$\rangle$		<b>/</b>	
HN O		HN HO	٠.
HO,		HO	
		·	
Br		Br	
PHA-530914	32	PHA-533249	8
₿r			•
		,o-{\bigcirc}	
но			
		N S HN	
		HO	
		o' Br	
ОН		ы	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-533232	64	PHA-533253	32
O HN HO Br		O HN HO Br	,
PHA-533243	32	PHA-533258	32
O O HN HO Br		No HO HO Br	
PHA-533247	32	PHA-533261	8
HO HO Br		O O O HN HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-533252	32	PHA-533265	128
N O HN HO Br		HN HO Br	
PHA-533257	16	PHA-533268	64
HN HO Br		HN O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 10., Structure		Compound No., Structure	
	MIC		MIC
PHA-533259	32	PHA-533272  HO  HN  O  O  O  O  O  O  O  O  O  O  O  O  O	128
		,	
PHA-533262	64	PHA-533274	8
HO HN O		O HN HO Br	
PHA-533264	128	PHA-533276	64
N O HIN HO Br		HO HN O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-533266	128	PHA-533281	64
HO HN O		N N O HIN HO Br	
PHA-533269	16	PHA-533285	64
HO HIN O		HN HO Br	
PHA-533273	64	PHA-533289	64
HN HO Br		HO HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-533275	8	PHA-533401	0.5
		ON HOUSE	
HO HO Br		,	
PHA-533278	32	PHA-537084	2
N O HO Br		least retained isomer by RP-HPL	
PHA-533282	>128	PHA-537089	32
N O HO HO Br		least retained isomer by RP-LC/Ms	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-533286	128	PHA-537091	8
		HN	
HO HO Br		DH OH	
PHA-533290	64 ·	PHA-537098	16
HO HN O		CI HIN HIN OH	
PHA-537085	16	PHA-537100	16
O D D D D D D D D D D D D D D D D D D D		HN OH HN OCI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 0210
	MIC	Compound 1vo., Structure	SA 9218
PHA 527000			MIC
PHA-537090	32	PHA-537106	8
NAMO NAMO		OH CI	
NH OH			
least retained isomer by RP-LC/MS			
PHA-537092	16	PHA-537112	128
NH OH		O NH OH	
PHA-537099	8	рна-537121	4
CI HN HN OOH		N OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	1	MIC
PHA-537101	4	PHA-537128	8
HN		CI OH HN	
OH HN O			
PHA-537110	64	PHA-537138	32
CI O O O O O O O O O O O O O O O O O O O		CI ST. O ST. O	
PHA-537114	16	PHA-537142	4
O OH CI		OH OH	
PHA-537122	16	PHA-537144	8
OH OH OH		HO HO CI	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-537133	8	PHA-537152	8
OH HNO		N OH OH OH OH	
ĊI			
PHA-537139	4	PHA-537157	32
HOO		N OH OH	
PHA-537143	16	PHA-537162	16
CI NO CI		O=S=O OH	
PHA-537150	32	PHA-537203	32
O=S=O OH OH		most highly retained isomer by RP-LC/MS	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		
PHA-537155	64	DITA 520016	MIC
	04	PHA-538016	64
O OH		Br OH	
PHA-537158	32	PHA-539146	128
HO CI		Br OH ON	
PHA-537202	8	PHA-539149	64
most highly retained isomer by RP-LC/MS		OH HN ON NH	

Commound No Standard	T G A 0010		T
Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-537204	64	PHA-539152	64
most highly retained isomer by RP-LC/MS		Br ON NH OH	
PHA-539148	64	PHA-539154	32
Br HO NH <sub>2</sub>		Br O NH OH N	
PHA-539150	64	PHA-539156	8
Br O O O O O O O O O O O O O O O O O O O		Br NH OH F	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539153		DUA 520160	
	32	PHA-539168	64
Br		Br O	
		NH OH	
O NH OH		<u>○</u>	
0=\$-\\\0\\			
ö		N N	
		<b>'</b>	
PHA-539155	32	PHA-539170	64
Вr		/─\ "o HO YO	
	!		
		O Br	
O NH OH		S N Br	
		N	
		·	
$\bigcup_{0=\S-N} \bigvee_{i=1}^{N}$			
PHA-539164	128	PHA-539172	1
Br		O Br	
NH OH		H OH OH	-
∘≕(		´	
		N	
o			
O\S\NH		CI	
N			
1			

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	ompound 110., bitueture	MIC
PHA-539169	16	PHA-539175	1
Br OHO O Br		Br O O NH OH	
PHA-539171	128	PHA-539179	8
Br N HO		F CI O Br	
PHA-539174	8	PHA-539181	64
Br OH		Br O H	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539177	4	DUA 520196	
Br OH OH		PHA-539186  Br OHHOHOHOHOHOHOM	0.5
PHA-539180	1	PHA-539188	16
Br ONH OH			
PHA-539183	8	PHA-539193	32

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539187	4	PHA-539195	64
Br OH OO		Br OH OH	04
PHA-539190	16	PHA-539198	128
Br OH OH		O S NH HO O	
PHA-539194	64	PHA-539203	1
Br OH OH		Br ONH OH	

- 287 -

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	, , , , , , , , , , , , , , , , , , , ,	MIC
PHA-539197	64	PHA-539207	2
Br OH OH OH NH ON NH ON NH OH		Br ONH OH	
PHA-539199	32	PHA-539209	0.5
Br ON NH NH NH NH NH NH NH NH NH NH NH NH NH N		Br O H	
PHA-539206	2	PHA-539235	128
Br O O O O O O O O O O O O O O O O O O O		HO HN F	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539208	16	PHA-539246	8
CI CI O Br		HO HN Br	
PHA-539234	128	PHA-539248	8
HO HN Br		HO HN CI	
PHA-539245	>128	PHA-539250	32
HO HIN N		HO HN N	
PHA-539247	64	PHA-539252	32
HO HN F		HO HN HO	
PHA-539249	8	PHA-539254	128
HN CI		HO HO Br	

- 289 -

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539251	128	PHA-539256	32
HO HN Br		HO HIN Br	
PHA-539253	16	PHA-539258	64
HO HN Br		HO HN Br	
PHA-539255	>128	PHA-539260	64
HO HN Br		HO HN F	
PHA-539257	8	PHA-539263	32
HO HN FFF		HO HO N.	
PHA-539259	128	PHA-539265	32
HO HN N		HO HIN Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539262	32	PHA-539267	0.5
HO HN N		HO HN S	
PHA-539264	8	PHA-539269	32
HO HN Br		HO HN F	
PHA-539266	2	PHA-539271	>128
HO HIN S		HO HN Br	
PHA-539268	32	PHA-539277	32
HO HN O		HO HN F	
PHA-539270	>128	PHA-539285	16
HO HN Br		HO HN Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC	composite 170., Structure	MIC
PHA-539276	32	PHA-539294	128
HO HN Br		HO HN NEO	128
PHA-539278	2	PHA-539296	64
HO HN Br		HO HN S=0	
PHA-539293	>128	PHA-539298	64
HO HN O		HO H	
PHA-539295	32	PHA-539303	32
HN HO NO D		HO HN F	
PHA-539297	>128	PHA-539307	>128
HO HIN CI		HN HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-539302	>128	PHA-539310	>128
OH HN Br		HO HN Br	
PHA-539305	64	PHA-539313	>128
HO HN Br		HO HN Br	
PHA-539308	128	PHA-539317	16
HO HN Br		HO HN P	
PHA-539312	128	PHA-539322	16
HO HIN Br		HO HN S	
PHA-539314	64	PHA-539329	>128
HO HN F		HN HO Br	

Compound No., Structure	SA 9218	Companyed No. Structure	CA 0210
Compound No., Structure		Compound No., Structure	SA 9218
	MIC		MIC
PHA-539318	>128	PHA-539337	32
HO HIN CI		HO	
PHA-539328	>128	PHA-543684	128
F N O HN O Br		Br O H CI	
PHA-539332	64	PHA-543686	4
HN F F		Br OH OH	

0 121 04 -4	CA 0210	Campanya d No. Chapatyana	SA 9218
Compound No., Structure	SA 9218	Compound No., Structure	
	MIC		MIC
PHA-539338	64	PHA-543690	32
HO HE O		CI HO Br	
PHA-543685	>128	PHA-543693	2
Br ONH OH		Br ONH OH Br	
PHA-543689	64	PHA-543698	>128
Br—OH OSSO		Br NH <sub>2</sub>	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
,	MIC	*	MIC
PHA-543692	16	PHA-543701	128
Br ONH OH F		Br O H	
PHA-543695	>128	PHA-543708	16
Br OH OH CI		HO HN N N N N N N N N N N N N N N N N N	
PHA-543700	64	PHA-551716	128
O NH OH O NH Br		O NH F OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-543706  Br  ONH OH  FF	32	PHA-563331	>128
PHA-551625  Br OH O	2	PHA-563335	8
PHA-551672  Br OH ONH OO	8	PHA-563341	8

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 1.0., Su detail	MIC	,	MIC
PHA-551675	32	PHA-563344	64
PHA-331073	32	HO HO N	
PHA-556420	128	PHA-563347	64
Br OH ON NO	·	HO HN N	
PHA-563330	>128	PHA-563351	>128
HO		HO HO NO	
PHA-563333	>128	PHA-563354	2
HO HN N		HO HN N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-563340	64	PHA-563363	16
HO HN F		HN HO	
PHA-563342	2	PHA-563365	16
HO		HOHO	
PHA-563345	16	PHA-563368	>128
HN HN N		F N O HN O	
PHA-563350	64	PHA-563371	16
HO		HN HO	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-563353	128	PHA-563378	16
HO HIN N		HO	
PHA-563360	32	PHA-563388	>128
HO HIN N		HO HN O	
PHA-563364	>128	PHA-563390	32
OH OH		HO HN O	
PHA-563366	4	PHA-563392	16
HO HN N		0=N HO O HO O Br	
PHA-563370	32	PHA-563394	16
HO HN N		F HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-563375	8	PHA-563398	>128
HO HN P		HO HIN HO	
PHA-563386	32	PHA-563399	16
HO HN O F F		O-NHOH	
PHA-563389	64	PHA-563404	8
F F Br		NH OH	
PHA-563391	>128	PHA-563407	>128
NH OH Br		HO HN O	
PHA-563393	128	PHA-563409	64
Br HO		HO HN O N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-563396	>128	PHA-563413	128
HO HN O N		N N O HO Br	
PHA-563397	32	PHA-563417	>128
F OH OH		HO HN O N	
PHA-563401	>128	PHA-563420	16
HO HN N		NH OH	
PHA-563406	64	PHA-563427	>128
NH OH OH		HO HN O O	
PHA-563408	>128	PHA-563441	64
HO HN O		S HO Br	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
,	MIC	,	MIC
PHA-563411	128	PHA-563449	64
HO HN O N	120	0=N	
PHA-563415	128	PHA-571150	0.5
HO HN O		HO	
PHA-563419	64	PHA-571152	8
HO JO Br		HNHO	
PHA-563426	64	PHA-571154	128
HO HN O		HO HN N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound No., Structure		Compound No., Structure	
	MIC		MIC
PHA-563440	64	PHA-571156	16
NH OH		HO	
PHA-563442	>128	PHA-571160	64
HO HN O		OH OH OH OH OH OH OH OH OH OH OH OH OH O	
PHA-569976	32	PHA-571162	16
OH OH		OH OH ONH OSSEO	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571151	8	PHA-571167	32
HO			
PHA-571153	64	PHA-571170	64
HO HIN BIT		OH OH	
PHA-571155	32	PHA-571174	64
HN			
PHA-571157	32	PHA-571182	64
HO		N= NO	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571161	>128	PHA-571186	128
HO O N SOO		OH OH	
PHA-571164	8	PHA-571189	64
OH OH ONH ONH ONH ONH ONH ONH ONH ONH ON		N N N N N N N N N N N N N N N N N N N	
PHA-571169	32	PHA-571196	64
HO O O O O O O O O O O O O O O O O O O		OH OH	
PHA-571172	32	PHA-571198	>128
O D D D D D D D D D D D D D D D D D D D		OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571176	64	PHA-571202	128
PHA-571183	32	PHA-571205	32
N N N N N N N N N N N N N N N N N N N		N HO	
PHA-571188	8	PHA-571208	64
		O S H N O O H	
PHA-571194	4	PHA-571215	8

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 1.0., Structure	MIC	Compound 1 to 1, ou would	MIC
PHA-571197	16	PHA-571219	32
ON SHOOL ON SHOLD ON SHOOL ON		OH O	
PHA-571199	64	PHA-571226	64
		OH ONH OSSO OH	
PHA-571203	32	PHA-571230	16
N N N N N N N N N N N N N N N N N N N		F O O O O O O O O O O O O O O O O O O O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound No., Structure	MIC	Compound 100, Structure	MIC
PHA-571207	32	PHA-571232	>128
OH ONH OS		N N N N N N N N N N N N N N N N N N N	
PHA-571214	16	PHA-571235	8
N S O O O O O O O O O O O O O O O O O O		HO HO N	
PHA-571216	32	PHA-571238	128
O H		N O O O H	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571224	8	PHA-571240	16
OH OH		N N N N N N N N N N N N N N N N N N N	
PHA-571228	32	PHA-571242	32
N=-VH O OH		$N = \bigvee_{\substack{N \\ HO}} \bigvee_{N \\ H$	
PHA-571231	>128	PHA-571246	32
N= OH N I S = O		HO O N N N N N N N N N N N N N N N N N N	
PHA-571234	8	PHA-571253	16
N= HO N N N N N N N N N N N N N N N N N N		N= N S=O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571237	16	PHA-571257	64
N N N N N N N N N N N N N N N N N N N		N N N N N N N N N N N N N N N N N N N	
PHA-571239	128	PHA-571260	32
HO HO		OH ONH ONH ONH ONH ONH ONH ONH ONH ONH O	
PHA-571241	16	PHA-571263	16
O O OH		OH OH OH	
PHA-571243	4	PHA-571265	16
OH OH OS=O		N= HO N N N N N N N N N N N N N N N N N N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571249	16	PHA-571269	16
NH OH OH		D N S N O O H	
PHA-571255	>128	PHA-571271	64
N O O O O O O O O O O O O O O O O O O O		ON SHOOM ON THE PROPERTY OF TH	
PHA-571258	8	PHA-571273	8
OHH OH		OH ONH ONH ONH ONH ONH ONH ONH ONH ONH O	
PHA-571262	32	PHA-571281	128
N= HO O O S S N			

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound No., Structure		Compound No., Structure	
	MIC		MIC
PHA-571264	32	PHA-571283	16
NH OH OH		O HO O Br	
PHA-571267	32	PHA-571287	2
F F F N N S N S N S N S N S N S N S N S		Br O N N O	
PHA-571270	8	PHA-571292	32
OH ONH OS=O		HO HN O	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571272	32	PHA-610941	>128
OH ONH OH OH OH OH OH OH OH OH OH OH		OCH <sub>3</sub>	
PHA-571280	>128	PHA-630426	>128
		NE OH OH OD	
PHA-571282	16	PHA-656808	64
O HO O Br		HO ON STATE OF THE PART OF THE	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-571285	64	PHA-656810	2
HO—Br		O, S, O N N O OH	
PHA-571289	32	PHA-656820	>128
HO HN O		CI OH	
PHA-610940	>128	PHA-656860	8
CF <sub>3</sub>		HO NO	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
-	MIC		MIC
PHA-610942	>128	PHA-656862	32
OH OH ON		OH N S N O N O N O N O N O O N O O N O O N O O O N O O O O	
PHA-656807	64	PHA-656866	>128
HO HO		NH OH HZ	
PHA-656809	64	PHA-656868	>128
OH N S O O O O O O O O O O O O O O O O O O		HO N N N N	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-656811	32	PHA-656871	128
O NH O OH		O NH O OH	
PHA-656859	16	PHA-656880	16
HO N S O OH		CI OH OY NEW YORK ON THE COLUMN TO THE COLUM	
PHA-656861	32	PHA-656883	16
HO ON SUCO OH OH		CI OH OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
•	MIC		MIC
PHA-656863	8	PHA-656885	16
OH ONH O= S=O OH		CI OH ONH O	
PHA-656867	64	PHA-656887	8
OH ON		CI OH ONH O	
PHA-656870	8	PHA-656889	16
NH OH OH		CI OH HO	
PHA-656872	>128	PHA-656891	16
OH NH OH NH OH		HO HN	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-656882	16	PHA-656893	8
CI OH OH OH		F O O O O O O O O O O O O O O O O O O O	
PHA-656884	16	PHA-662253	128
CI OH ONH O		$N \equiv \bigcup_{i=1}^{N} OH$ $CH_2 - SO_2 - N$	
PHA-656886	16	PHA-662412	64
N CI OH		N OH NH OH	
PHA-656888	16	PHA-679759	>128
CI OH		O-N OH ON S-N O	

Compound No., Structure	SA 9218	Compound No., Structure	· SA 9218
	MIC	·	MIC
PHA-656890	16	PHA-708922	>128
HO HN O		О ОН ОП В	
PHA-656892	8	PHA-708977	>128
F F HO HO CI		HN O HN O HO Br	
PHA-656894	16	PHA-708987	>128
OH OH		HN H	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
,	MIC		MIC
PHA-662254	>128	PHA-713390	>128
$N = \bigcirc$ OH $CH_2 - SO_2 - N$ O			
PHA-679756	>128	PHA-713392	>128
HO-N OH ON S-NO		S CI O O O O O O O O O O O O O O O O O O O	
PHA-687570	128	PHA-713395	>128
HO OH NH O2S N			

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Compound 1400,	MIC		MIC
PHA-708979	>128	PHA-738531	64
HN 700375		N OH ON SEN N	
PHA-713389	>128	PHA-740499	128
CI N HN O HO Br		N OH	
PHA-713391	>128	PNU-276556	
CI SHN HN O HO Br		Br OCH <sub>3</sub>	

----

Company of the compan	104.0010		
Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
PHA-713393	>128	PNU-276873	
HO HO HO	,	Br COOH OCH <sub>3</sub>	
PHA-713397	>128	PNU-282858	
		Br OH NH CI	
PHA-738532	32	PNU-282860  Br OH	

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
Composite Every Composite	MIC	•	MIC
PHA-748361	8	PNU-291997	1
COOH ON NH		Br OH OH	-
PNU-276672		PNU-281164	>128
Br ONH OH O=S=O HN CI		F O OH	
PNU-292577  Br OH ONH O	128	PNU-282859  Br OH NH OF SO NO	32

00033 031

Compound No., Structure	SA 9218	Compound No., Structure	SA 9218
	MIC		MIC
		PNU-290881A	4
		ОУОН	
		CI	
		NH <sub>2</sub> HCI	